



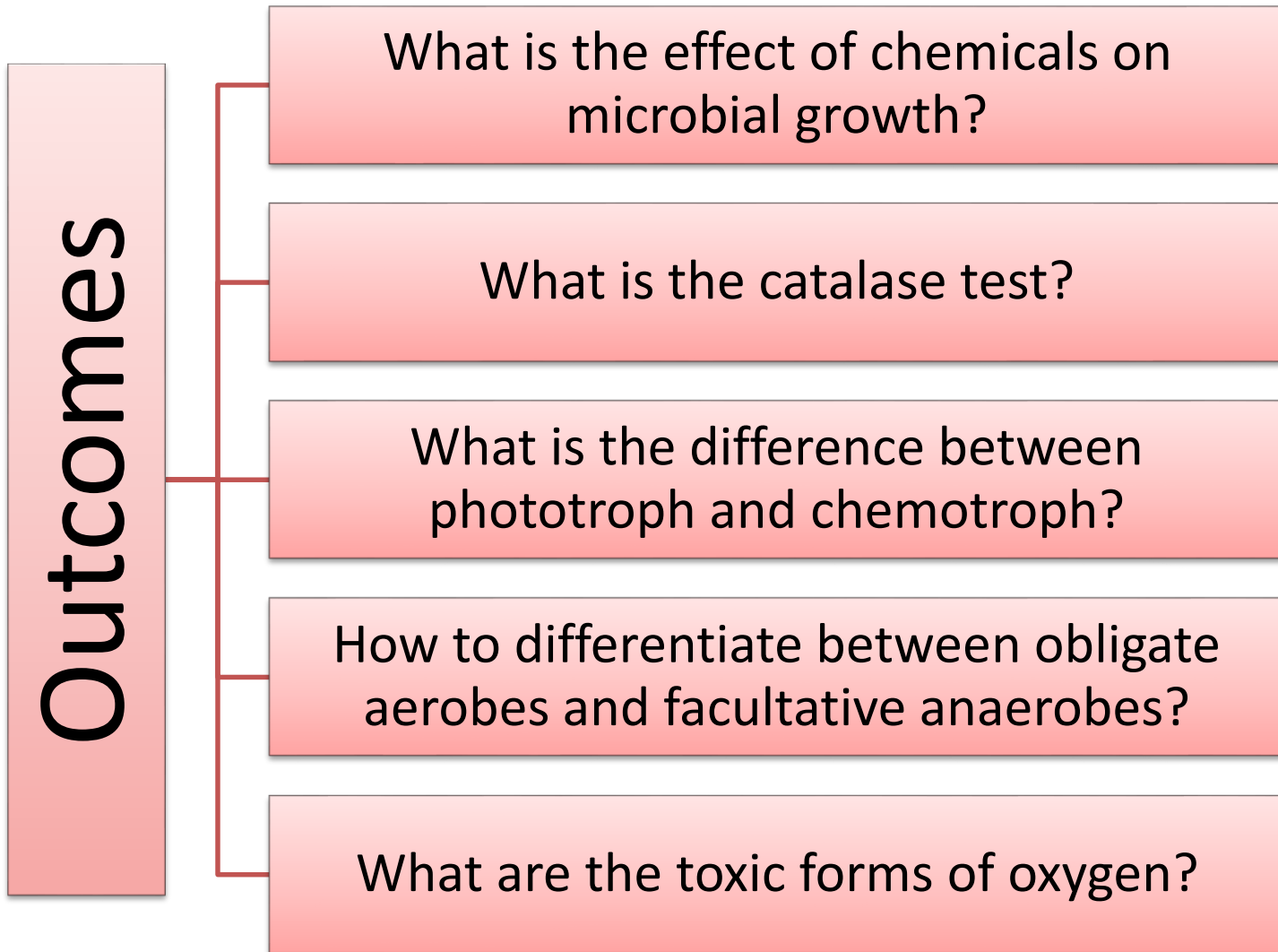
OPT0425
MICROBIOLOGY I

GAMAL EL-HITI

Microbial Growth Control through Chemical Requirements

Lecture Eleven

Learning Outcomes



Microbial Growth Control

• Chemical Requirements

1– Water

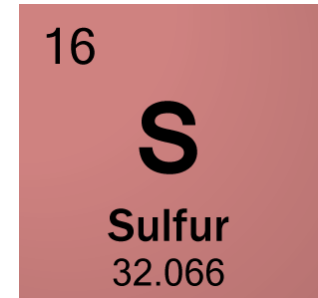
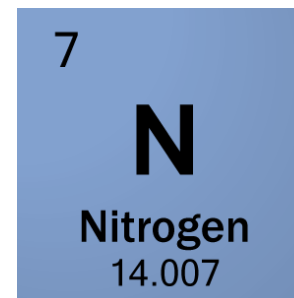
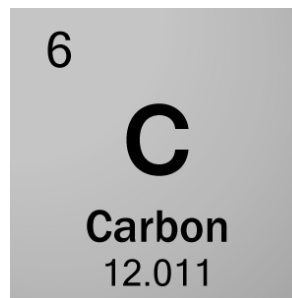
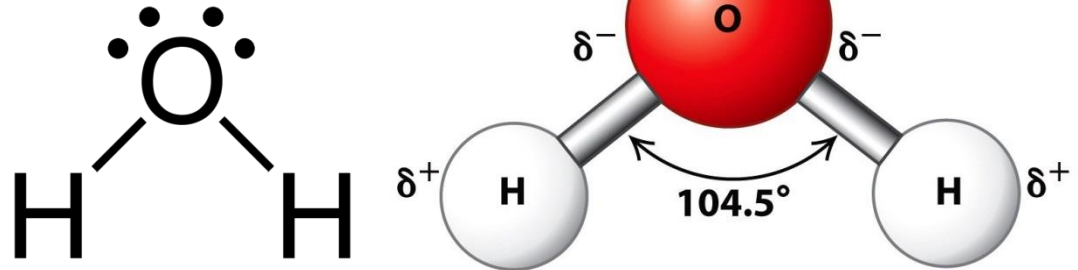
2– Elements

- Carbon

- Nitrogen

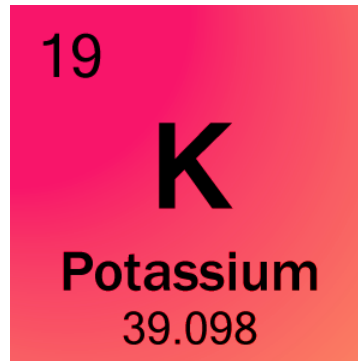
- Sulfur

- Traces of other elements (P, *etc.*)

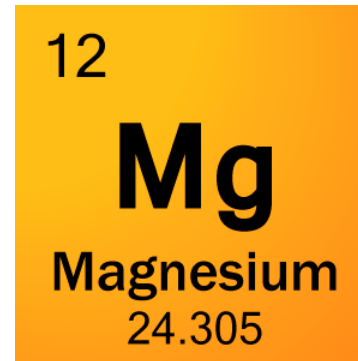


3– Other Elements

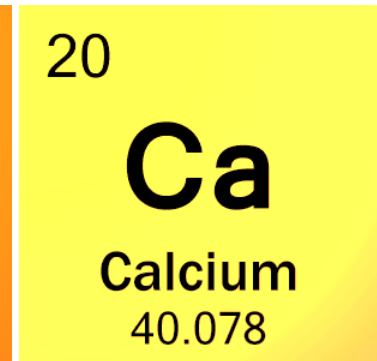
- Potassium



- Magnesium



- Calcium



- These elements are often required as enzyme cofactors.

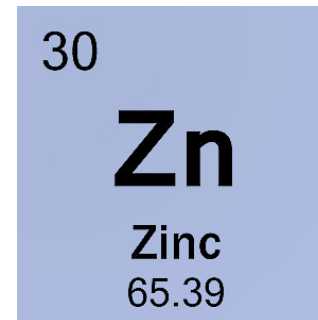
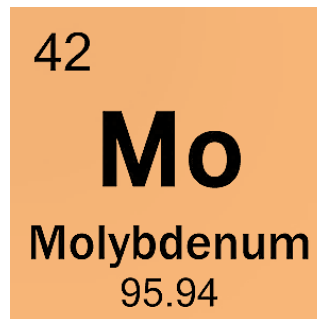
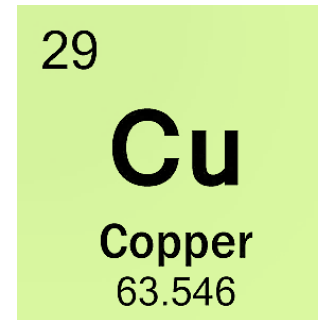
- For example, **calcium** is required for cell wall synthesis in Gram positive bacteria.

Microbial Growth Control

4– Trace Elements

- Many elements are used as enzyme cofactors.
- These elements commonly found in tap water.

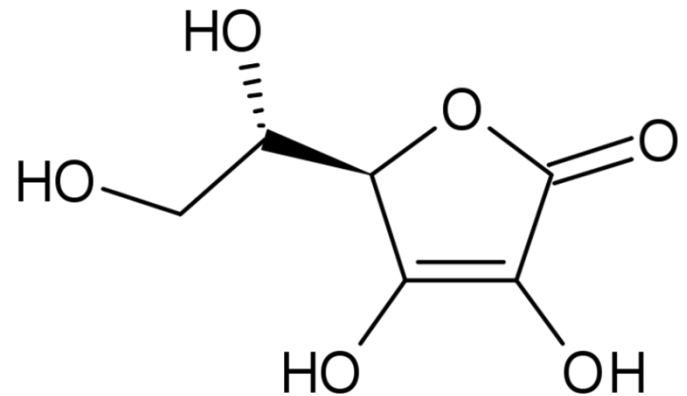
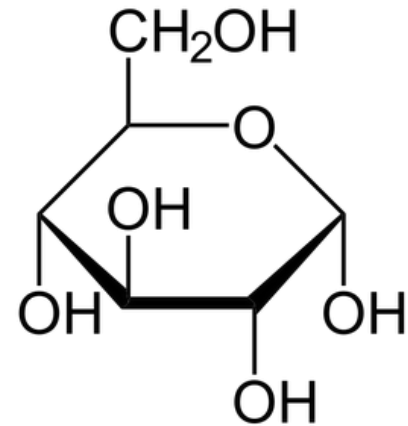
- Iron
- Copper
- Molybdenum
- Zinc



5– Organic Compounds

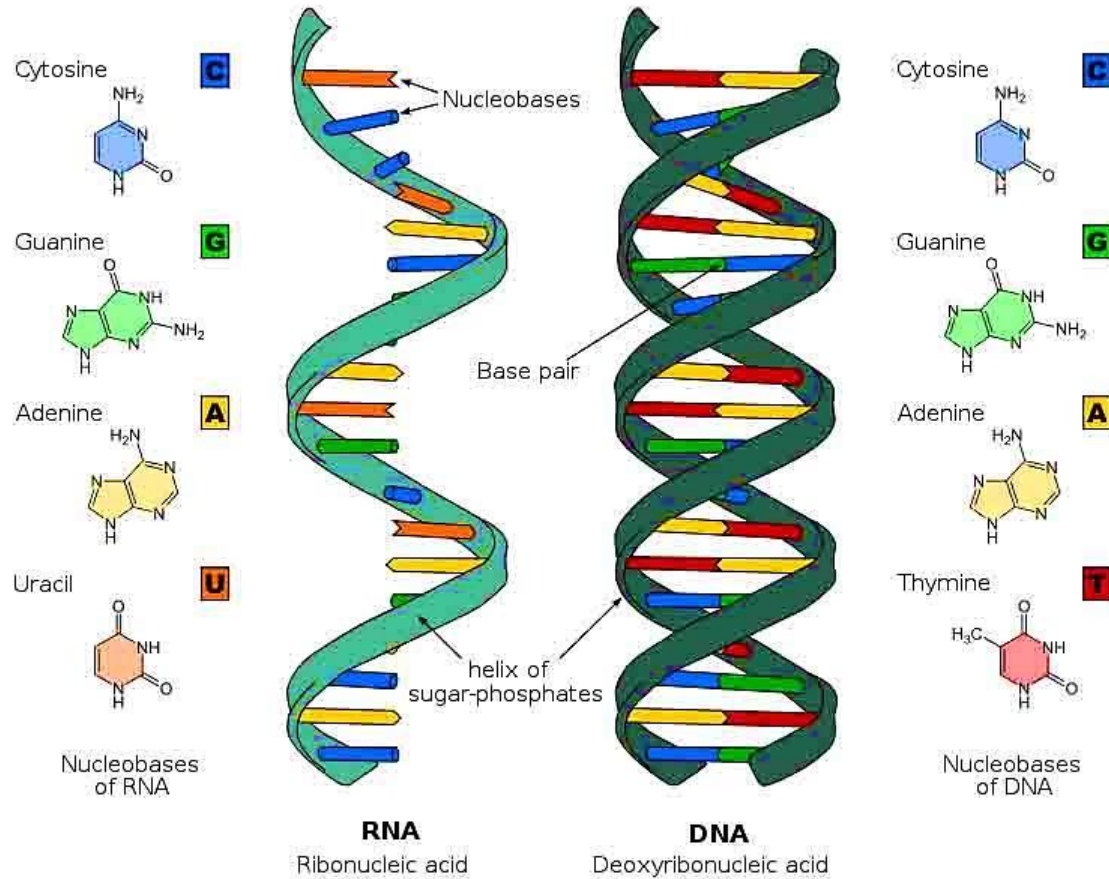
- Organic Compounds Containing Oxygen
- Glucose
- Source of energy

- Vitamins (coenzymes)
- *e.g.* Vitamin C or *L*-ascorbic acid



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- Purines and Pyrimidines



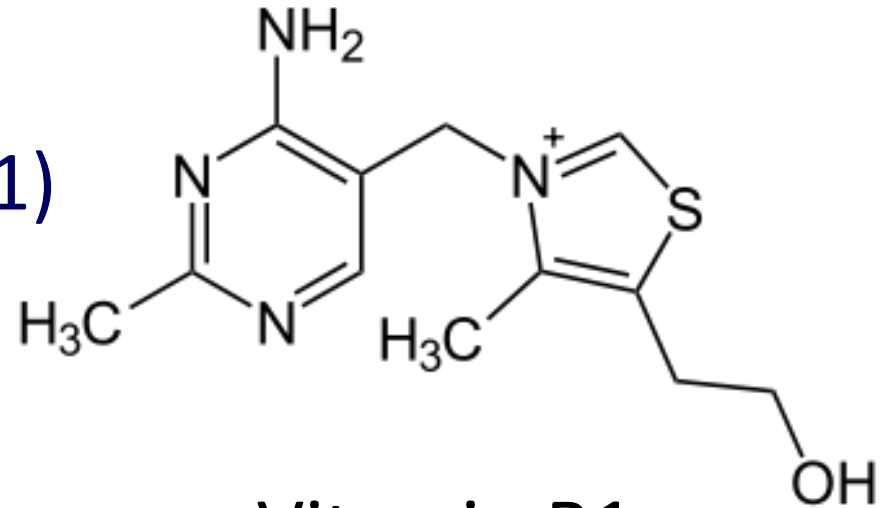
- Most bacteria decompose proteins.

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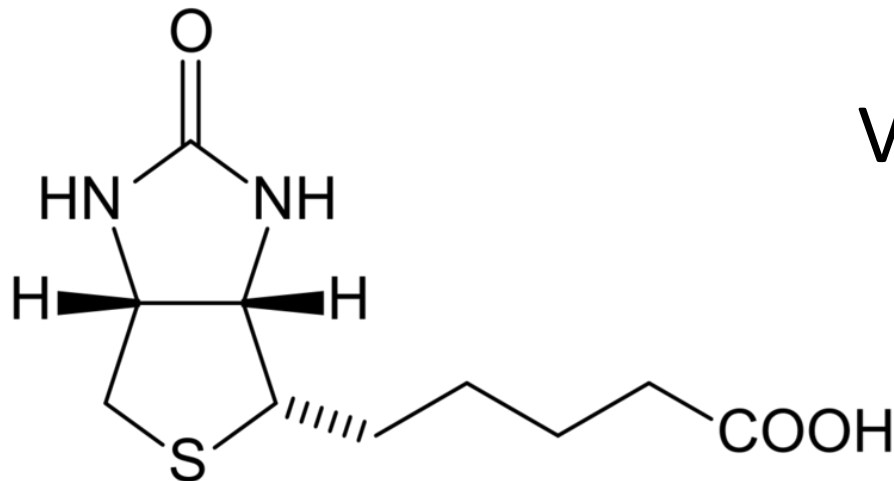
- Organic Compounds
Containing Sulfur

- Thiamine (vitamin B1)

- Biotin (vitamin H)



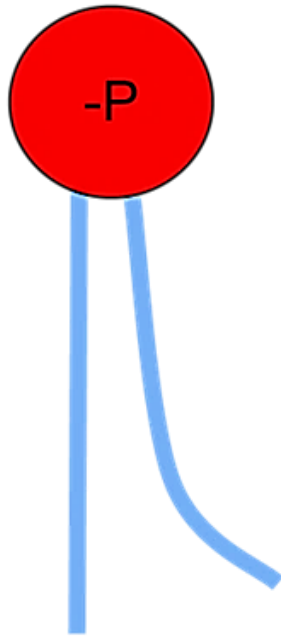
Vitamin B1



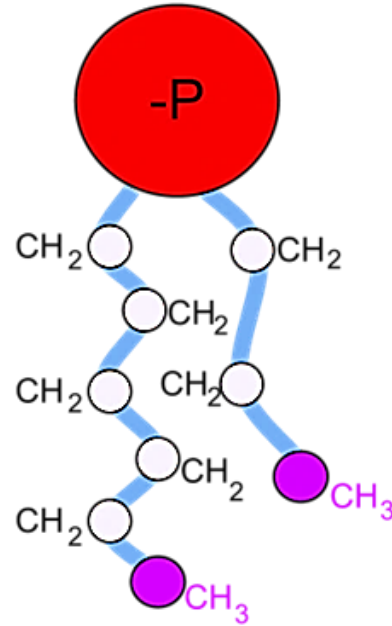
Vitamin H

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- Organic Compounds Containing phosphorous
- DNA, RNA, ATP and phospholipids.



A phospholipid with a hydrophilic head and a hydrophobic tail



Chemical makeup of a single phospholipid

6– Carbon

- All chemical substances in microbes contain carbon in some form, whether in the form of proteins, carbohydrates or lipids.
- Perhaps 50% of a bacterium's dry weight is carbon.
- Carbon can be obtained from organic materials in the environment, or it may be derived from carbon dioxide.

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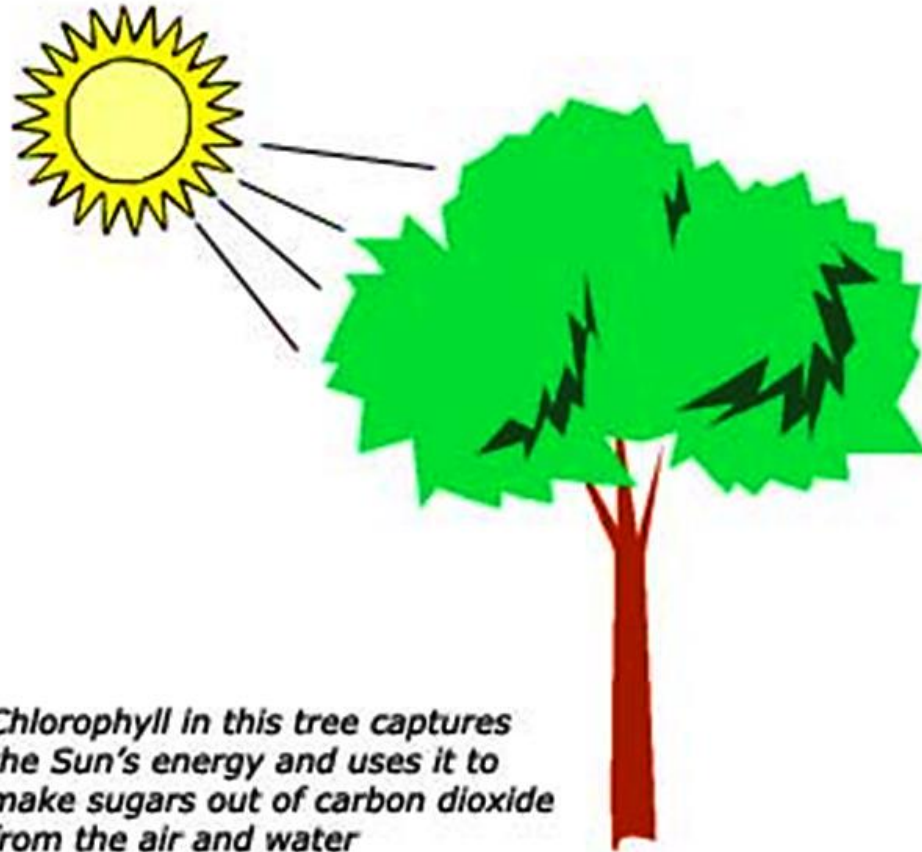
- What are chemoheterotrophs, chemoautotrophs and photoautotrophs?
- Chemoheterotrophs are microorganisms that obtain energy from the oxidation of organic compounds (derive both carbon and energy from organic compounds).
- Organic compounds could be proteins, carbohydrates and lipids (such as fats, waxes, sterols, *etc.*; natural molecules).
- Chemoautotrophs and photoautotrophs obtain carbon from carbon dioxide.

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- Carbon Sources
- CO₂: autotroph
- Organic: heterotroph
- Energy Sources
- Sunlight: phototroph
- Organic: chemotroph
- Saprobe: lives on organics of dead organisms.
- Parasite: lives on organics of living host.

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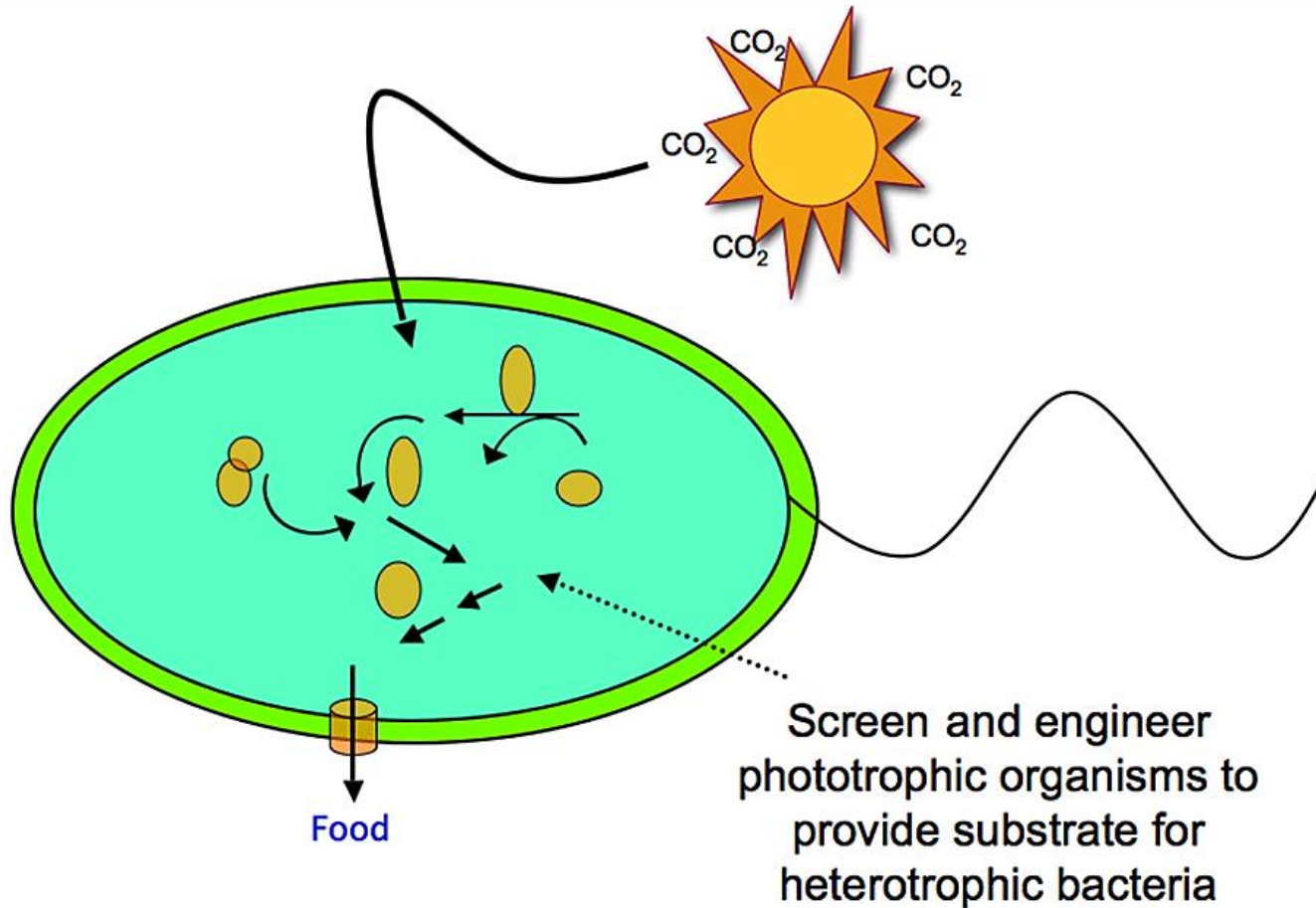
Autotroph



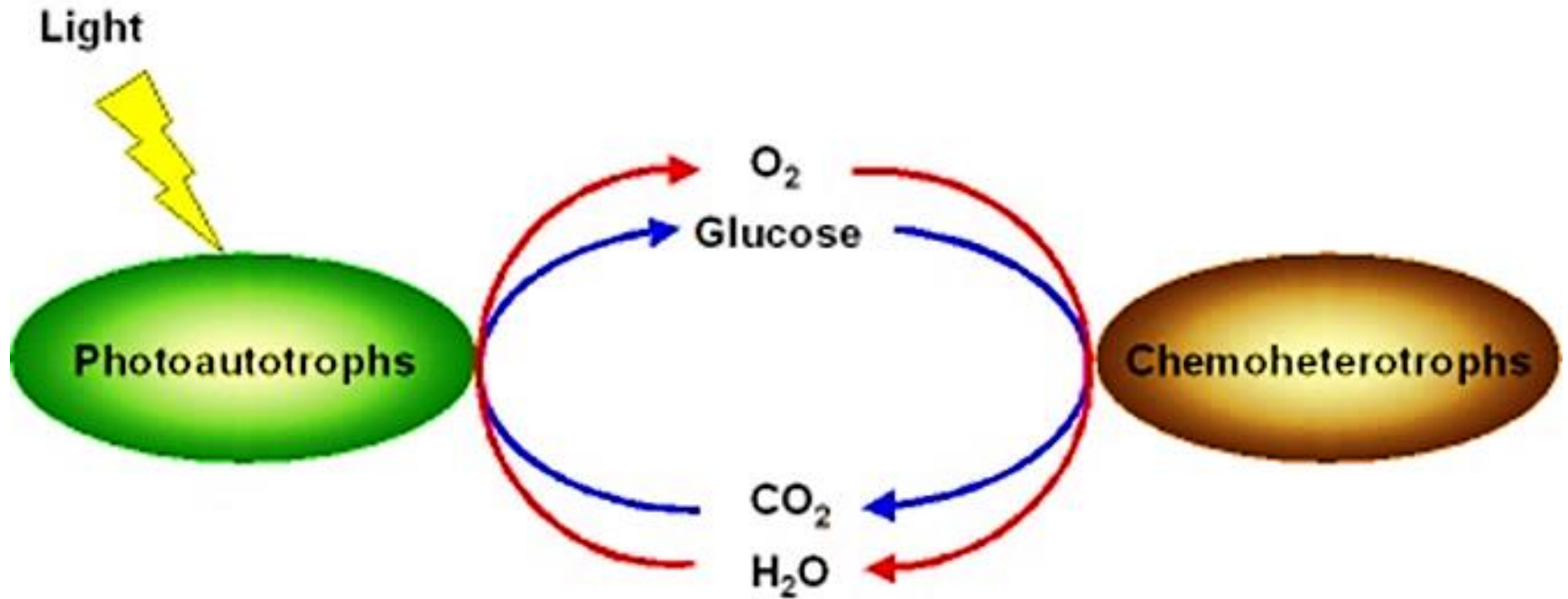
Chlorophyll in this tree captures the Sun's energy and uses it to make sugars out of carbon dioxide from the air and water

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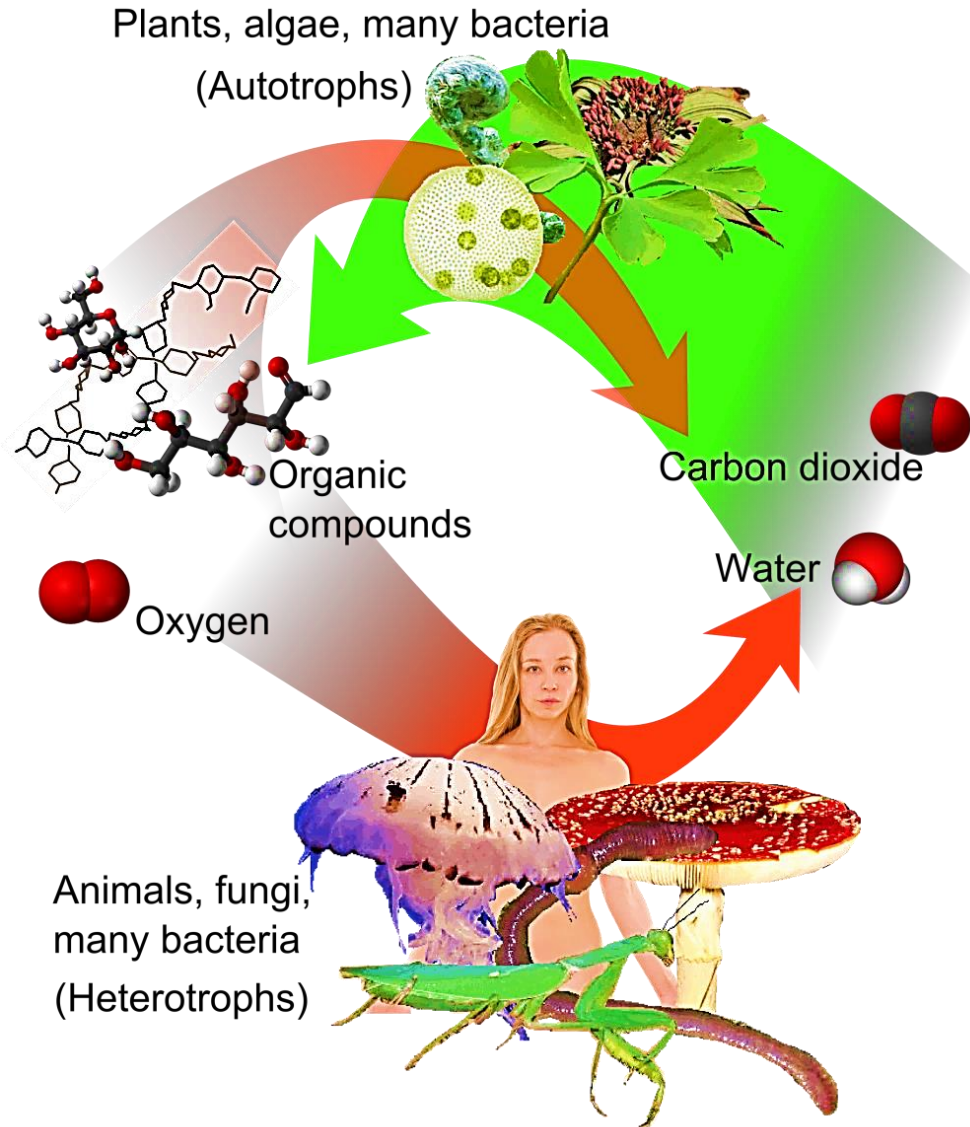
Phototroph



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Microbial Growth Control



7– Oxygen

- Microorganisms that use molecular oxygen (O_2), produce more energy from nutrients than anaerobes.
- Microorganism can be classified based on their oxygen requirements.
- **Obligate Aerobes:** require oxygen to live.
- **Disadvantage:** O_2 dissolves poorly in H_2O .
- **Example:** *Pseudomonas* which is a common nosocomial pathogen.

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- Facultative Anaerobes
 - They can use oxygen, but can grow in its absence.
 - Examples: *E. coli*, *Staphylococcus*, *yeasts*, and many intestinal bacteria.
- Obligate Anaerobes
 - They cannot use oxygen and are harmed by the presence of toxic forms of oxygen.
 - Examples: *Clostridium* bacteria that cause tetanus and botulism.

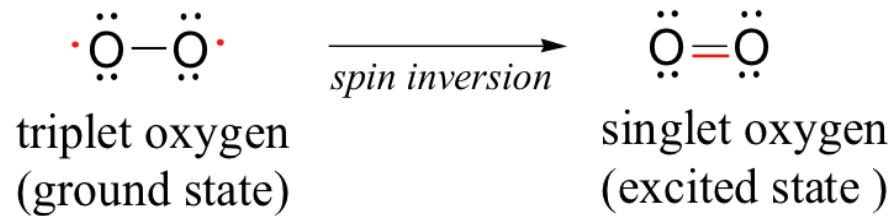
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- **Aerotolerant Anaerobes**
- They can't use oxygen, but tolerate its presence and can break down toxic forms of oxygen.
- **Example:** *Lactobacillus* that carries out fermentation regardless of oxygen presence.
- **Microaerophiles**
- They require oxygen, but at low concentrations and sensitive to the presence of toxic forms of oxygen.
- **Example:** *Campylobacter*.

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Toxic forms of oxygen

1– Singlet Oxygen: extremely reactive and present in phagocytic cells.

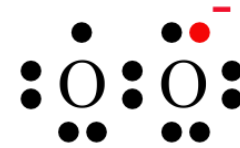


- Singlet oxygen is the common name used for electronically excited molecular oxygen (O_2 ; *i.e.* higher energy state).
- It is less stable than normal triplet oxygen.

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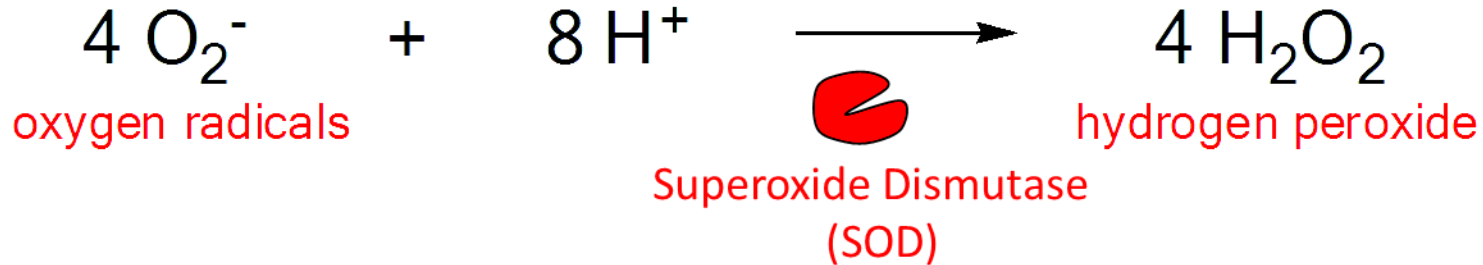
2– Superoxide Free Radicals ($O_2^{\cdot-}$):

extremely toxic and reactive.



- All organisms growing in atmospheric oxygen must produce superoxide dismutase (**SOD**) enzyme to get rid of toxic forms of oxygen.
- **SOD** is made by aerobes, facultative anaerobes and aerotolerant anaerobes, but not by anaerobes or microaerophiles.
- How to handle superoxide free radicals?

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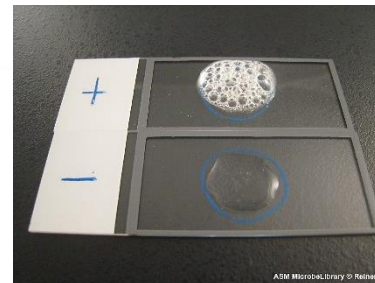
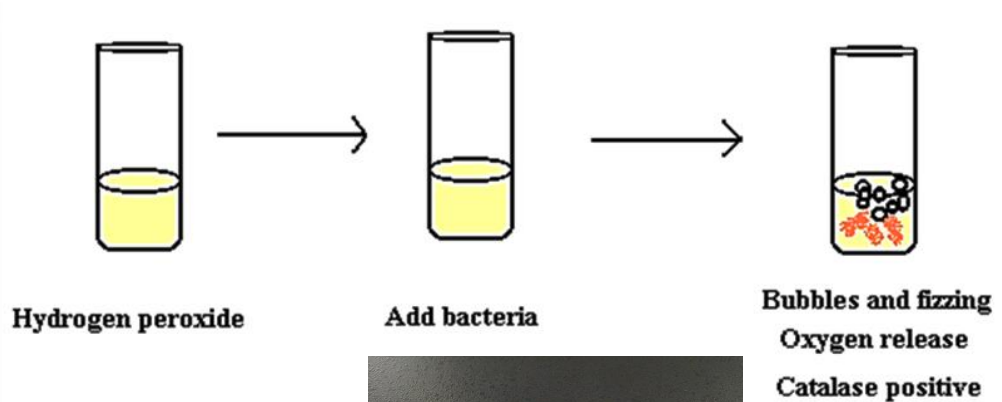
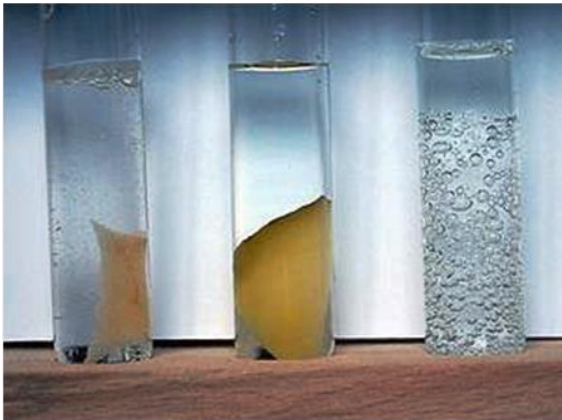
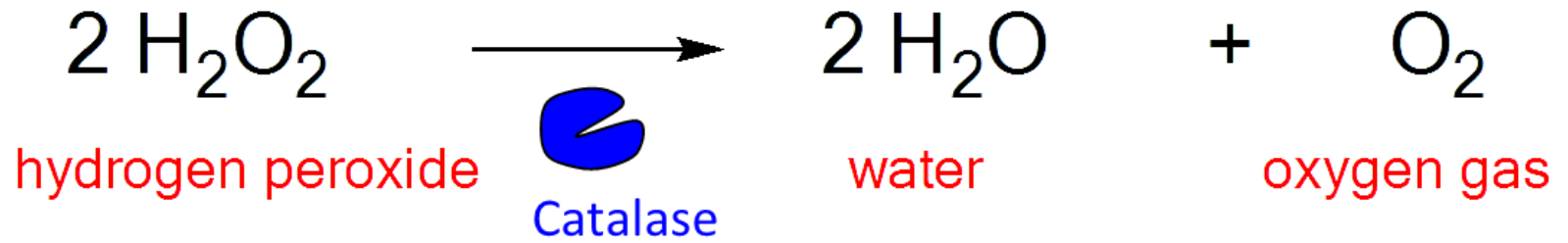


3– Hydrogen Peroxide (H_2O_2) is toxic and the active ingredient of several antimicrobials (*e.g.* benzoyl peroxide).

- There are two different enzymes that break down H_2O_2 (catalase and peroxidase).
- **Catalase:** breaks hydrogen peroxide into water (H_2O) and oxygen (O_2).
- Produced by humans and many bacteria.

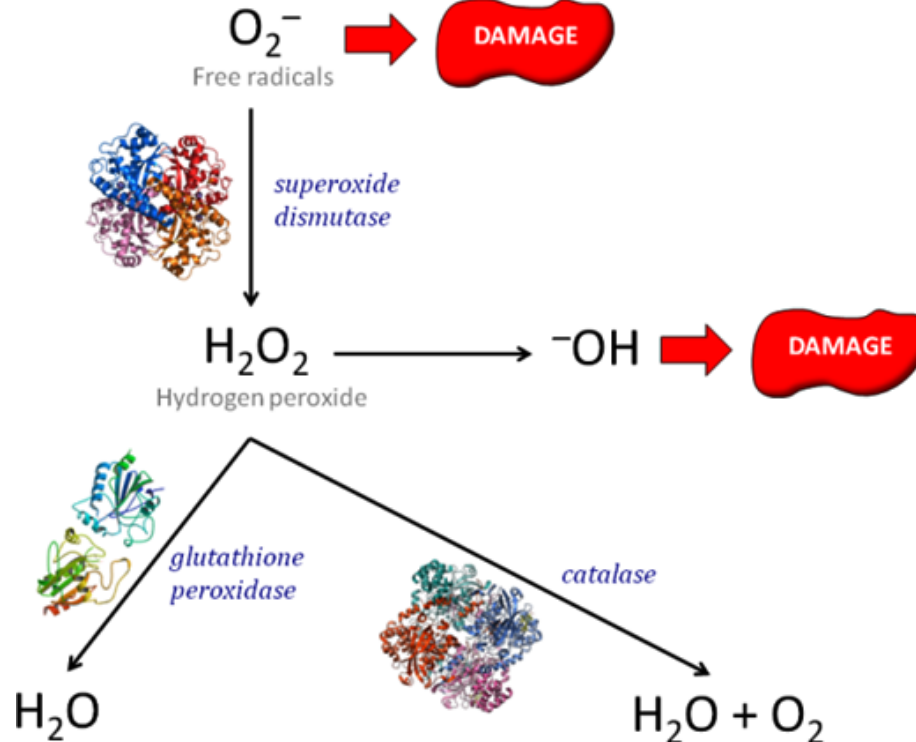
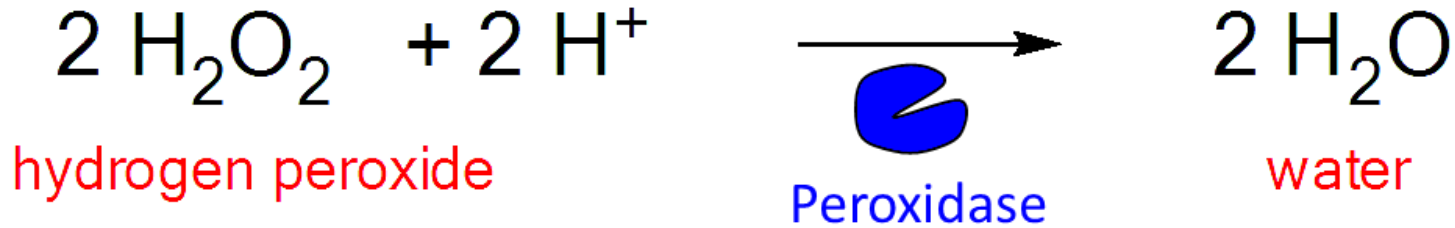
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Catalase Test: Bacteria in the presence of H_2O_2 gives as O_2 as bubbles.



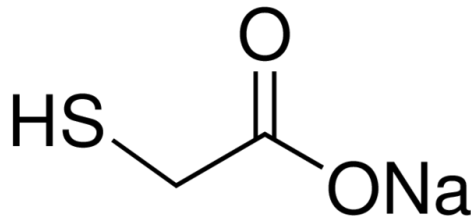
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- **Peroxidase:** converts H_2O_2 into water (H_2O).



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- Thioglycollate broth
- It is a multi-purpose, enriched, differential medium used to determine the oxygen requirements of microorganisms.
- Sodium thioglycolate in the medium consumes oxygen and permits the growth of obligate anaerobe:



20%



40%

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- Oxygen was diffused from the top of the broth produces various oxygen concentrations in the media along its depth.
- The oxygen concentration at a given level is indicated by a redox sensitive dye like resazurine.
- Dye turns pink in the presence of oxygen.

