#### **Energy Storage**

- Energy is stored in a medium to use it at different time and/or place
- Energy storage is useful to compensate for lack of sunshine during nighttime or at times of sunshine interruptions

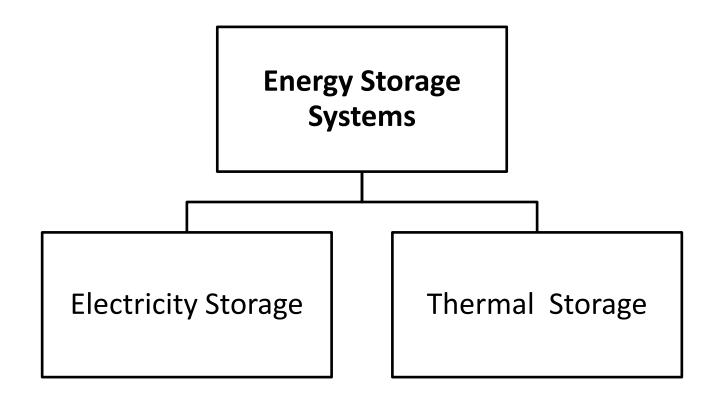
#### **Applications of Energy Storage**

- Power generating systems
- Solar heating
- Cooling
- Process heat applications

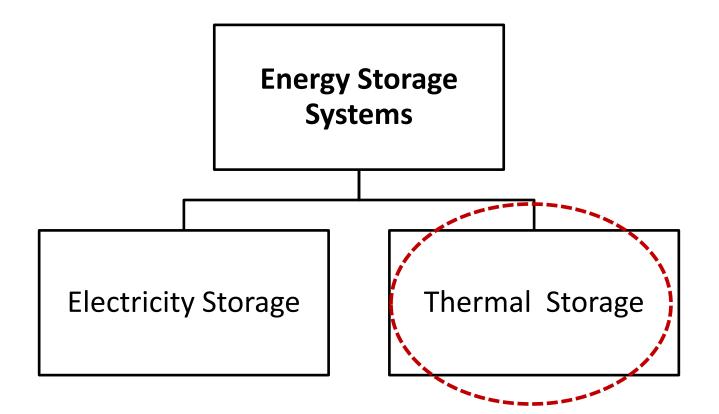
#### **Advantages of Energy Storage**

- Improved system reliability
- Extended system availability
- Cost reduction

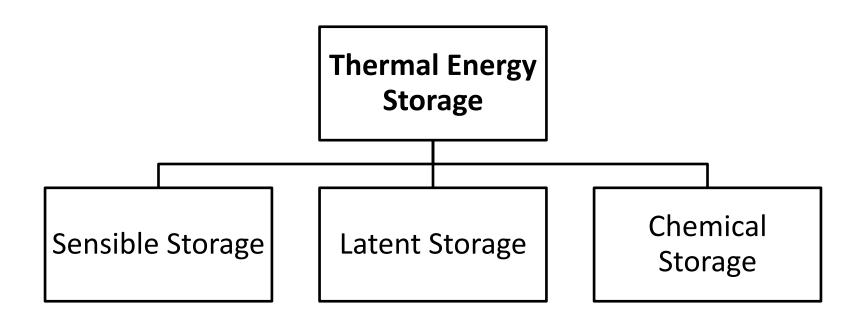
#### **Types of Energy Storage**



#### **Types of Energy Storage**



#### **Types of Thermal Energy Storage**



#### **Types of Thermal Energy Storage**

- Sensible storage: by change in temperature
- Latent storage: by change of phase
- *Chemical storage:* by exo- or endo-thermic chemical reactions

# Technical Requirements for Thermal Energy Storage Systems

- High energy density (per-unit mass or per-unit volume) in the storage material
- Good heat transfer between heat transfer fluid (HTF) and the storage medium
- Mechanical and chemical stability of storage material
- Chemical compatibility between HTF, heat exchanger and/or storage medium
- Complete reversibility for a large number of charging/discharging cycles

# Technical Requirements for Thermal Energy Storage Systems

- Good relationship between heat storage capacity and cost
- Very low vapor pressure at working temperatures
- Low thermal losses
- Ease of control
- Not flammable

# Thermal Energy Storage Sensible Storage

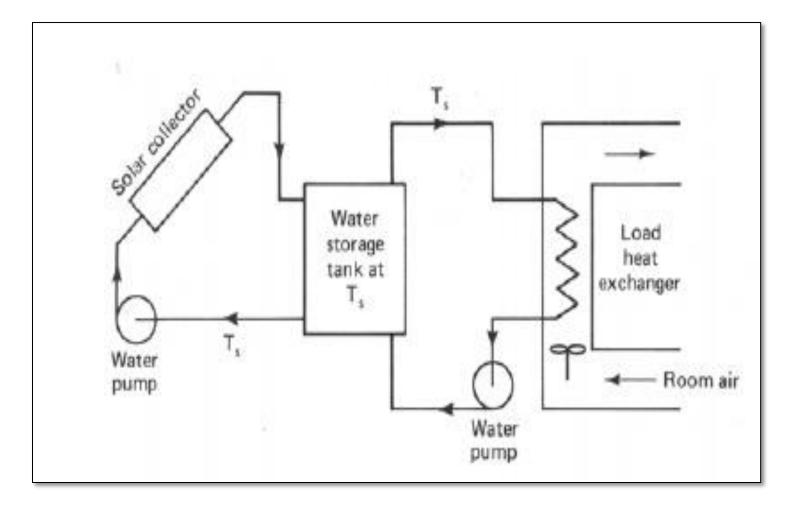
- **Definition:** A thermal energy storage system where the addition or removal of energy results in a change in temperature
- The energy storage medium could be liquid or solid  $\dot{Q} = \dot{m}C_p\Delta T$
- Examples of energy density of sensible storage materials:
  - Stone, concrete, etc: **1.5-3.5 MJ/m<sup>3o</sup>C**
  - Water: **4.15 MJ/m<sup>3o</sup>C**

- Water
- Mineral oil
- Nitrate salts
- Carbonate salts
- Liquid sodium

# **Thermal Energy Storage Sensible Storage in Water**

- Water is the most frequently used as storage medium for liquid systems.
- Water is standard storage medium for solar heating and cooling systems for buildings.
- Other liquids can be used for thermal storage above 100C

#### **Thermal Energy Storage Sensible Storage in Water**



# **Thermal Energy Storage Sensible Storage in Water**

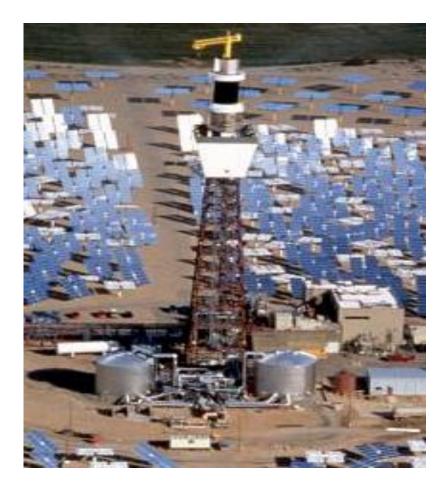
#### Advantages:

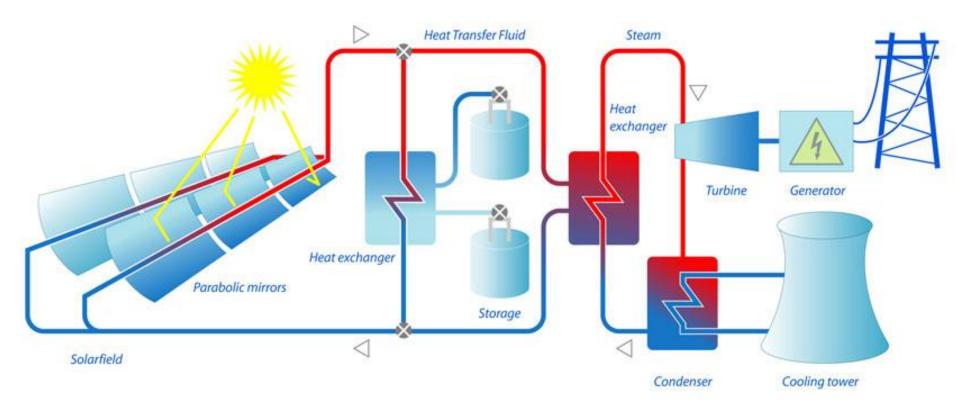
- 1. Abundant
- 2. Low cost
- 3. Non-toxic
- 4. High storage capacity
- 5. Easy to transport
- 6. Non combustible

#### Disadvantages:

- 1. High vapor pressure
- 2. Corrosive medium
- 3. Destructive expansion
- 4. Non isothermal energy delivery

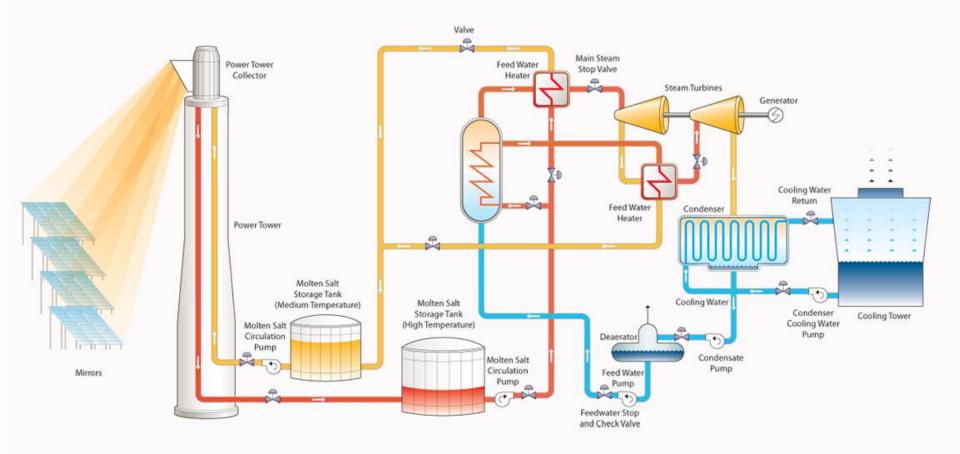
- Most common composition: Eutectic mixture of NaNO3 (60%) and KNO3 (40%)
- Fusion temperature: 221 °C
- Maximum Operating Temperature: 565 °C
- Usually two tanks (cold and hot) are used





- Storage Capacity: 7.5 hours @ 50 MW. 1,010 MWh.
- Thermal Storage Description: 28,500 tons of molten salt
- Tanks size: 14 m high, 36 m diameter.
- Cold tank temperature: 292 °C
- Hot tank temperature: 386 °C





$$Q_s = (mC_p)_s \ \Delta T_s$$

 $Q_s$  is the total heat capacity for a cycle operating through the temperature range  $\Delta T_s$  and *m* is the mass of liquid in the unit

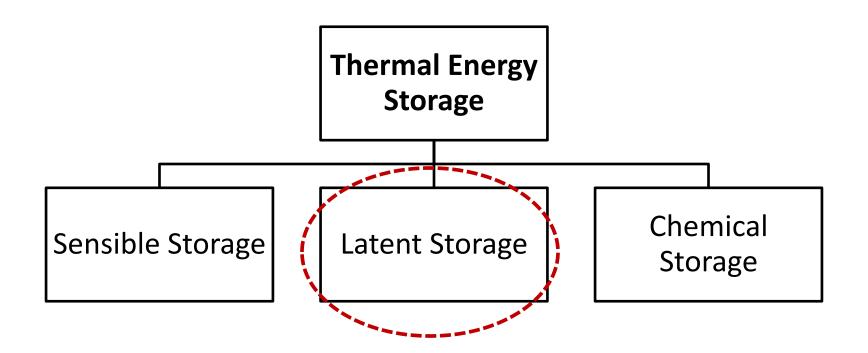
$$(mC_p)_s \frac{\mathrm{d}T_s}{\mathrm{d}t} = Q_u - \dot{L}_s - (UA)_s(T_s - T_a')$$

- $Q_u$  and  $\dot{L}_s$  are rates of addition or removal of energy from the collector and to the load
- $T'_a$  is the ambient temperature for the tank (which may not be the same as that for a collector supplying energy to the tank).

**Numerical Solution** 

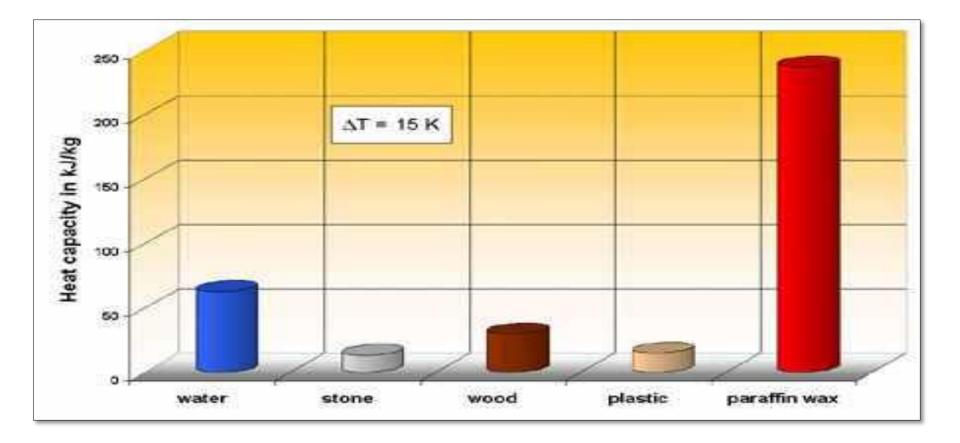
$$T_{s}^{+} = T_{s} + \frac{\Delta t}{(mC_{p})_{s}} \left[Q_{u} - L_{s} - (UA)_{s}(T_{s} - T_{a}')\right]$$

#### **Types of Thermal Energy Storage**



- Latent heat is the energy required to change the state of a unit mass of material from solid to liquid or liquid to gas without a change in temperature.
- Used when thermal energy storage systems have limited space.
- Used for specific applications in building systems, or specific processes in the agricultural and industrial sector.

Group	PCM type	Transition temperature (°C)	Latent heat (Wh/kg)	Thermal conductivity, liquid / solid (W/mK)
Inorganic salt	Calcium chloride	27-30	47-53	0.54 <sup>a</sup> / 1.09 <sup>b</sup>
hydrates	Sodium sulfate	32	70	-
	(Glauber's salt)			
	Zinc nitrate	36	41	0.46 <sup>c</sup> / -
	Magnesium nitrate	89	45	0.49d / 0.61e
Organics	Polyglycol E400	8	28	0.19 <sup>f</sup> / -
	Polyglycol E600	22	35	0.19 <sup>f</sup> / -
	Octadecane	28	68	-
	Ecosane	37	69	-
	Paraffin 116 (paraffin wax)	48	58	-
	Paraffin 6403 (paraffin wax)	62-64	48-53	0.17 <sup>g</sup> / 0.35 <sup>h</sup>
Fatty acids	Palmatic acid	63	52	0.16 <sup>i</sup> / -
-	Capric acid	32	42	0.15 <sup>j</sup> / -
Organic/inorganic	Mystiric acid (mainly	54	52	-
mixes (Eutectics)	inorg.)			
Aromatics	Naphtalene	80		$0.13^{k} / 0.34^{l}$



#### Thermal Energy Storage Latent (Phase Change) Storage Technical Requirements

- 1. High energy of phase change
- 2. Negligible corrosion
- 3. Negligible expansion
- 4. Good repeatability of freeze-thaw temperature
- 5. Must be resistant to chemical or physical change resulting from thermal cycling.

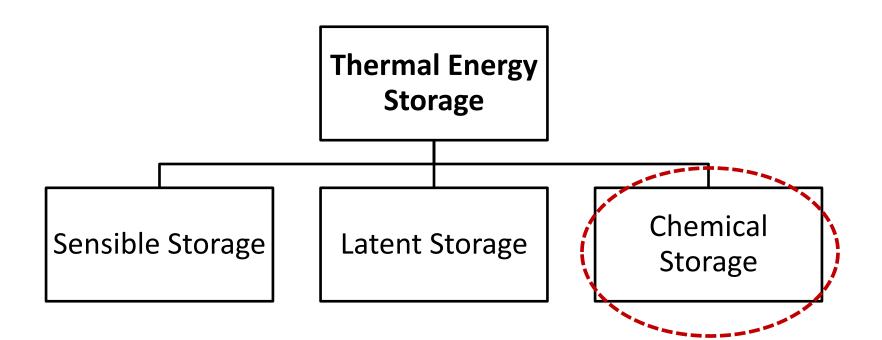
#### Advantages

- Store energy at the temperature of process application
- Allows higher thermal energy storage capacity per unit weight or material without change in temperature
- Smaller required storage size
- Relatively constant temperature during charging and discharging.

#### Disadvantages

- Higher investment costs
- Peak power during discharge is limited due to limited heat conduction in the solid state of the material.
- Limited experience with long-term operation of many thousands of charge-discharge cycles.
- Risks of loss of stability

#### **Types of Thermal Energy Storage**



#### Thermal Energy Storage Chemical Storage

