

HOMEWORK I (30 marks)

Dr. Vasileios Lempesis

Hand in: Thursday 5th October 2017 at 23:59

- The mass of a small glass beaker was measured with a balance and found to be 45.64g. When a liquid was added to the beaker the balance indicated 92.5g. Give the mass of the liquid in the beaker to the appropriate number of significant figures.

(5 marks)

Solution:

$$92.5 - 45.64 = 46.86$$

But we must keep one decimal so

$$46.86 \rightarrow 46.9 \text{ g}$$

- In a heat transfer experiment, an amount of heat, Q , is transferred to silver at its melting point, causing mass, m , of the silver to melt. The relationship between m and Q is $Q = mL$, where L represents the heat of fusion of the silver. Given that $L = 88 \times 10^3 \text{ J/kg}$ and $Q = 4550 \text{ J}$, calculate the mass of silver melted in kg. Express your answer in scientific notation to the appropriate number of significant figures.

(5 marks)

Solution:

$$m = \frac{Q}{L} = \frac{4550}{881000} = 5.164 \times 10^{-3} \text{ kg}$$

But we must keep two s.f. so

$$5.164 \times 10^{-3} \text{ kg} \rightarrow 5.2 \times 10^{-3} \text{ kg}$$

- Round the following recordings at the digit which is underlined:

Recorded Value	3.1 <u>0</u> 8	59 <u>2</u> 3	6. <u>9</u> 73	0.04 <u>5</u> 6	<u>9</u> .5
Rounded Value	3.11	5920	7. <u>0</u>	0.05	10 or 9

(5 marks)

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Comment [1]: No need to turn it to kg

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Comment [2]: Rounding has to be at the first decimal so 7.0 not 7

4. Fill in the following table by keeping significant figures and rounding properly the recorded values taken in an experiment for a physical quantity:

Recorded value	Before the selection of significant figures		After the selection of significant figures		Final Result $x \pm \delta x$
	x	δx	δx	x	
1	4.2838	0.01762	0.018	4.284	4.284 ± 0.018
2	91.2	0.0219	0.022	91.200	91.200 ± 0.022
3	27.2679	0.7538	0.8	27.3	27.3 ± 0.8
4	3982	213	210	3980	3980 ± 210
5	11348	84	80	11350	11350 ± 80

(5 Marks)

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Comment [3]: First s.f. of error is 1 so we must keep 2 s.f. in the error after rounding

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Comment [4]: First s.f. of error is 2 so we must keep 2 s.f. in the error after rounding