

# Detection and quantitative estimation of proteins by different methods



## Protein quantification:

- The accurate quantitation of protein content is a critical step in protein analysis.
- Depending on the **accuracy** required and the **amount and purity of the protein** available, different methods are appropriate for determining protein concentration.

## Methods:

1. **Direct assay:** measure the absorbance at 280 nm (UV range).
2. **Colorimetric and fluorescent, reagent-based protein assay:** Protein is added to the reagent, producing a color change or increased fluorescence in proportion to the amount added.

### **The most commonly used techniques**

biuret test, Bradford test, bicinchoninic acid assay (BCA assay) and Lowry test.

## Choosing the compatible method:

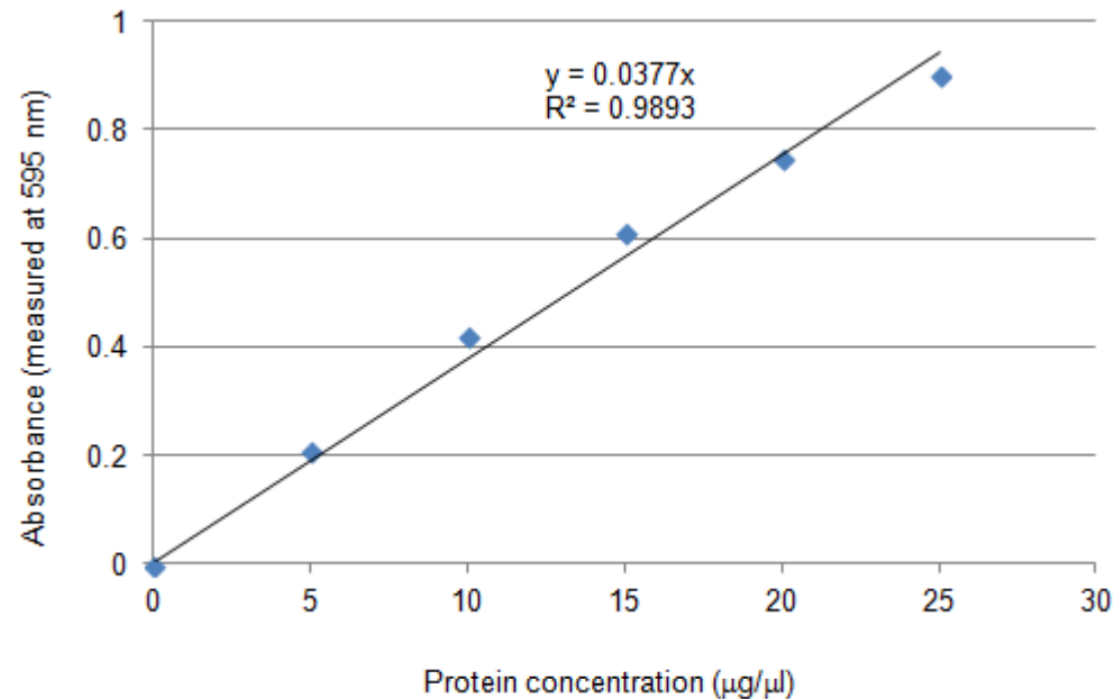
- Each method has its advantages and disadvantages.
- Choosing an appropriate assay:
  - Interfering substances
  - Accuracy
  - Incubation time
  - Availability

**Table 1. Comparison of various methods used for total protein concentration determination.**

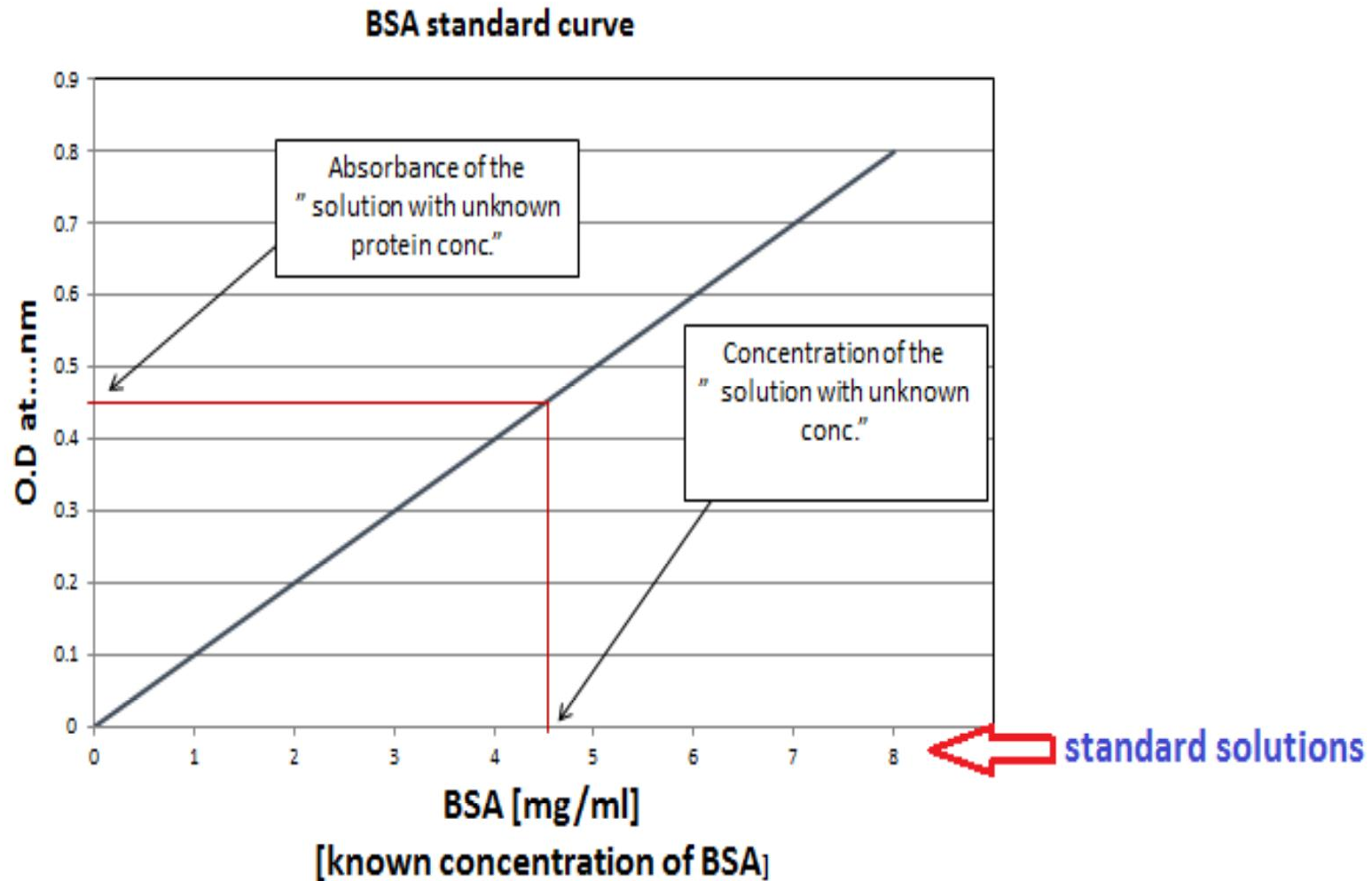
Method	Sensitivity	Time	Reagent	Interferences	Disadvantages and comments
<b>Biuret</b>	Low 1-20 mg	Moderate 20-30min	Alkaline copper sulphate	Zwitterionic buffers, Some amino acids	Similar color with all proteins. Destructive to protein samples.
<b>Lowry</b>	High ~ 5 µg	Slow 40-60min	Cu <sup>2+</sup> Folin– Ciocalteu	Ammonium sulphate, glycine, Zwitterionic, buffers, Mercaptans	Time-consuming. Color varies with proteins. Destructive to protein samples.
<b>Bradford</b>	High ~ 1 µg	Rapid 15 min	Coomassie Brilliant Blue G-250	Strongly basic Buffers, detergents Triton X-100, SDS	Stable color, which varies with proteins. Reagent commercially available. Destruction to protein samples. Discoloration of glassware.
<b>BCA</b>	High ~ 1 µg	Slow 60 min	Cu <sup>2+</sup> , bicinchoninic acid	EDTA, DTT, Ammonium sulphate	Compatible with detergents. Reagents commercially available. Destructive to Protein samples.
<b>Spectroph- otometric (A<sub>280</sub>)</b>	Moderate 50-100 µg	Rapid	-	Purines, pyrimidines, Nucleic acids	Useful for monitoring column eluent. Nucleic acid absorption can be corrected. None-destructive to protein samples. Varies with proteins.

## Determination of protein concentration:

- Protein concentration is determined by reference to a **standard curve** consisting of known concentrations of a purified reference protein.



## Standard curve:



# Practical Part



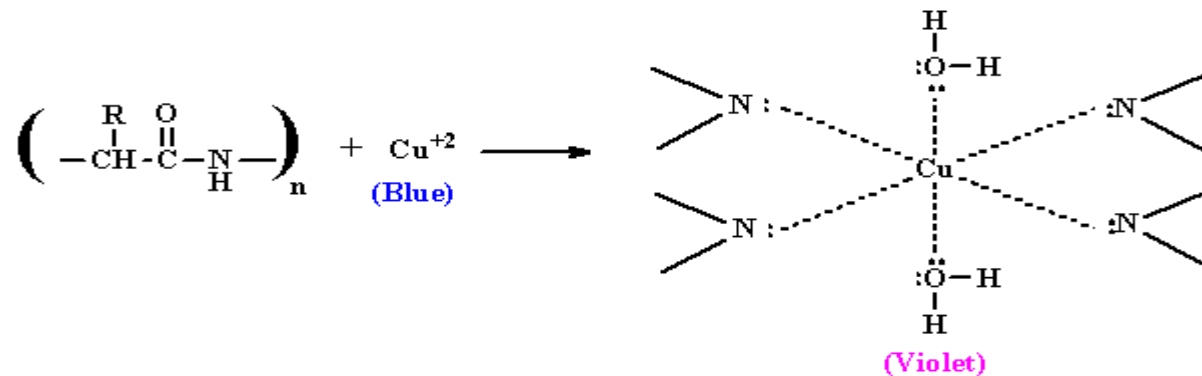
# Experiment (1). Qualitative detection of proteins by biuret test:

## Objective:

To detect the presence of a protein and peptides using Biuret test.

## Principle:

- peptide bonds in the proteins and peptides treated with an alkaline solution of dilute copper sulphate CuSO<sub>4</sub> (biuret reagent) forming a **purple colored complex**.
- The color density is **proportional** to the amount of proteins present.
- This test is specific for the peptide bond, positive result will be given if has two or more peptide bonds.



From lower to higher concentration



## Method:

1. Label three test tubes as A and B
2. In tube A: add 1 ml of animal crude extract.
3. In tube B: add 1 ml of water.
4. Add 1 ml of biuret reagent to all tubes and mix well.

## Result

Tube	Observation
Animal crude extract	
water	



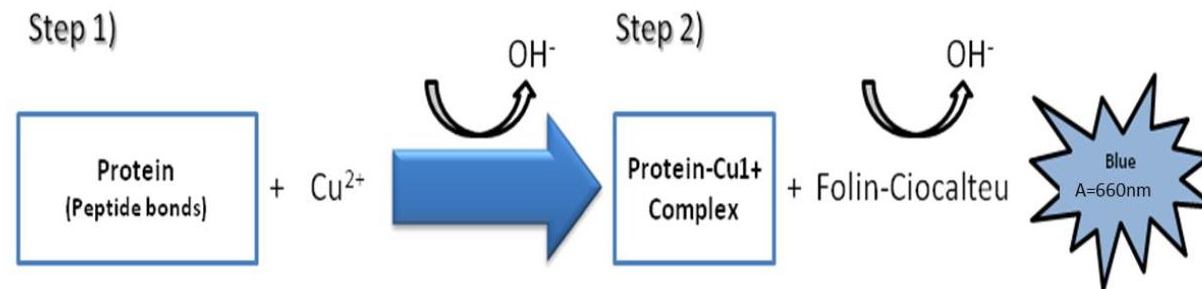
## Experiment (2). Quantitative estimation of proteins by Lowry assay:

### Objective:

To determine the concentration of extracted protein by Lowry assay.

### Principle:

- When the Folin reagent (a mixture of sodium tungstate, molybdate and phosphate), together with a copper sulphate solution, is mixed with a protein solution, a **blue-purple color** is produced.
- The method is based on two chemical reactions.



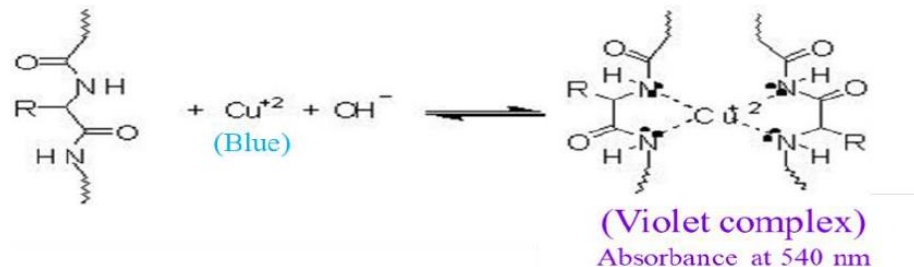
## Experiment (3). Quantitative estimation of proteins by biuret assay:

### Objective:

To determine the concentration of extracted protein by biuret assay.

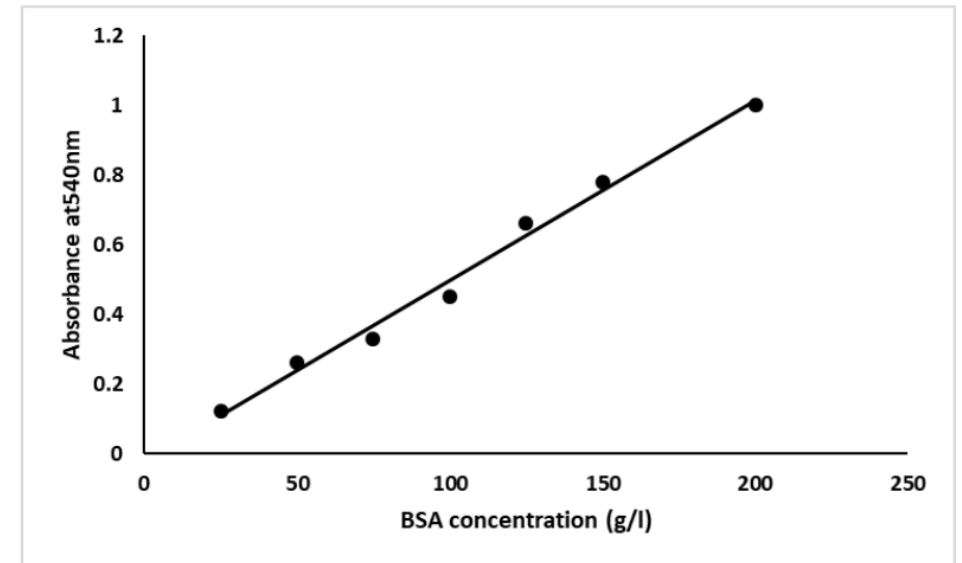
### Principle:

- Biuret method is based on copper ions  $\text{Cu}^{2+}$  binding to peptide bonds of protein under alkaline condition to give a violet color that have a maximum absorbance at 540 nm.
- The intensity of the color, and hence the absorption at 540 nm, is directly proportional to the protein concentration, according to the Beer–Lambert law



## Result

Test tube	Protein concentration (g/L) [X- axis]	Absorbance at 540 nm [Y- axis]
Blank		
A		
B		
C		
D		
E		
F		
G		
Animal crude extract (D1)	_____	
Animal crude extract (D2)	_____	
Plant crude extract (D1)	_____	
Plant crude extract (D2)	_____	
40% pellet	_____	
Dialyzed sample	_____	



**Figure 1. Standard curve of BSA using biuret method.**