Symmetry Operations and Space Groups


## Symmetry Elements

there are $\mathbf{5}$ types in point symmetry

1. center of symmetry (or inversion): point $\overline{\mathbf{1}}$
2. rotation (or proper) axis : line $n$
3. mirror : plane m
4. rotation-inversion axis : line $\bar{n}$
5. identity : no element

Center of Symmetry: $\overline{1}$
all points $(\boldsymbol{x}, \boldsymbol{y}, \boldsymbol{z}) \longrightarrow(-\boldsymbol{x},-\boldsymbol{y},-\boldsymbol{z})$ if $\overline{\mathbf{1}}$ is placed at the origin


## Crystal Symmetry

32 point groups of crystals compatible with 7 crystal systems
crystallographers use Hermann-Mauguin symmetry symbols


Carl Hermann
German
1898-1961
Charles-Victor Mauguin
French
1878-1958

## Center of Symmetry: $\overline{\mathbf{1}}$

a point in the molecule through which if another point on the molecule is taken, will meet an identical point on the molecule an equal distance away


## Rotation Axis: n

$\mathbf{n}$ is an integer which gives the degrees of rotation: $\frac{\mathbf{2 \pi}}{\mathbf{n}}$ or $\frac{\mathbf{3 6 0}}{\mathbf{n}}$
$\mathbf{n}$ is the number of times molecule is rotated, each time stopping at an identical appearance, before returning to the starting point

$\mathbf{n}$ is the foldness of the rotation axis
only 2, 3, 4, and 6-fold axes allowed in crystal symmetry


## Rotation-Inversion $\overline{\mathbf{n}}$

rotation followed by inversion
this is a different definition than Schoenflies system

Arthur Moritz Schönflies - German 1891
rotation followed by reflection

## Mirror: m

plane within the molecule that, when acting as a mirror, reflects the molecule into itself


## Representation of Symmetry

point symmetry often represented symbolically in the form of points on a circle (projection of a sphere)
a point above plane is a filled circle:
a point below plane is an open circle:
two points directly on top of each other: starting with one point, find other points generated by symmetry

32 point groups compatible with 7 crystal systems


## Monoclinic


$2\left(C_{2}\right)$
$m\left(C_{s}\right)$

monoclinic convention: symmetry located wrt $\mathbf{b}$ axis
2: 2-fold axis along b
$\boldsymbol{m}$ : mirror perpendicular to $\mathbf{b}$
2/m: 2-fold axis along $\mathbf{b}$, perpendicular to a mirror



## Lattices

14 Bravais lattices have Laue symmetry all have a center of symmetry
center of symmetry very important in crystallography:
centrosymmetric or noncentrosymmetric


## Translational Symmetry

in repeating lattices, two additional symmetry elements

## translational elements

1. screw axis rotation and translation: $\mathbf{n}_{\mathbf{r}}$
rotation by $\mathbf{3 6 0} / \mathbf{n}$;
followed by translation of $\mathbf{r} / \mathbf{n}$ along that axis (a,b or $\mathbf{c}$ )
2-fold screw axis most common: $\mathbf{2}_{1}$
2. glide plane reflection and translation: a,b, $\boldsymbol{c}, \boldsymbol{n}$ or $\boldsymbol{d}$ reflection across plane;
followed by translation of $\mathbf{1 / 2}$ (usually) unit cell parallel to plane along $\mathbf{a}, \mathbf{b}$, $\mathbf{c}$, face diagonal ( $\boldsymbol{n}$ ), or body diagonal ( $\boldsymbol{d}$ )


## Glide Plane - a


http://www.cut-the-knot.org/Curriculum/Geometry/GlideReflection.shtml
Space Groups
translational elements + point symmetry $\Rightarrow$ space groups
in 2-D, referred to as plane groups
there are 17 distinct ways of packing repeating object in 2-D
wallpaper patterns


Space Groups
translational elements +32 crystal point groups;
230 space groups
230 distinct ways of packing repeating object in 3-D


| tetragonal |  | Space Groups |  |  | I4 | I4 ${ }_{1}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 4 | P4 | P41 | $\mathbf{P 4} \mathbf{2}^{1}$ | P43 |  |  |
| $\overline{4}$ | $\mathbf{P} \overline{4}$ | I $\overline{4}$ |  |  |  |  |
| 4/m | P4/m | $\mathrm{P} 42 / \mathrm{m}$ | P4/n | $\mathrm{P} 42 / n$ | I4/m | I $4_{1} / a$ |
| 422 | P422 | P42,2 | P4122 | P4, $\mathbf{2 1}_{12}$ | P4222 | $\mathbf{P 4} \mathbf{2}_{1} \mathbf{2}$ |
|  | P43 22 | $\mathrm{P4}_{3} \mathbf{2 1}_{1}{ }^{2}$ | 1422 | I4,22 |  |  |
| 4 mm | P4mm | P4bm | $\mathrm{P4}_{2} \mathrm{~cm}$ | $\mathrm{P4}_{2}$ nm | P4cc | P4nc |
|  | $\mathrm{P4}_{2} \boldsymbol{m} \boldsymbol{c}$ | $\mathrm{P}_{2}{ }_{2} \boldsymbol{c} \boldsymbol{c}$ | I4mm | I4cm | I4 ${ }_{1} m$ m | $\mathrm{I}_{1} \mathrm{c}$ cd |
| $\overline{42} \mathrm{~m}$ | $\mathbf{P} \mathbf{4} 2 m$ | $\mathrm{P} \overline{42} c$ | $\mathbf{P} \overline{42}{ }_{1} m$ | $\mathbf{P} \overline{42}{ }_{1} c$ | $\mathbf{P} \overline{4} m 2$ | $\overline{\mathbf{P}} \mathbf{c} 2$ |
|  | $\mathbf{P} \overline{4} \mathbf{2} \boldsymbol{b}$ | $\mathbf{P} \overline{4} \boldsymbol{n} \mathbf{2}$ | I $\overline{4} \boldsymbol{m} 2$ | $\mathbf{I} \overline{4} \boldsymbol{c} 2$ | I42m | I42d |
| 4/mmm | $\mathrm{P} 4 / \mathrm{mmm}$ | P4/mec | $\mathrm{P} 4 / \mathrm{nbm}$ | P4/nnc | $\mathrm{P} 4 / \mathrm{mbm}$ | P4/mnc |
|  | $\mathrm{P} 4 / \mathrm{nmm}$ | P4/nnc | $\mathrm{P}_{2} / \mathrm{mmc}$ | $\mathbf{P} 4_{2} / \mathrm{mcm}$ | $\mathrm{P}_{2} / \mathrm{nb} \boldsymbol{c}$ | $\mathbf{P} 42 / \mathrm{nnm}$ |
|  | $\mathrm{P}_{2} / \boldsymbol{m b c}$ | $\mathbf{P} 4_{2} /$ mnm | $\mathrm{P} 4_{2} / \mathrm{nm} \boldsymbol{c}$ | $\mathbf{P} 4_{2} / \mathrm{ncm}$ | I4/mmm | 14/mcm |
|  | $\mathrm{I}_{1} /$ amd | I4 $1^{\text {/acd }}$ |  |  |  |  |


| cubic |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 23 | P23 | F23 | 123 | $\mathbf{P} 2,3$ | $\mathbf{I} 2,3$ |  |
| m3 | Pm3 | Pn3 | Fm3 | Fd3 | Im3 | Pa3 |
| 432 | P432 | $\mathbf{P} \mathbf{4}_{2} 32$ | F432 | F4 32 | I432 | P43 32 |
|  | $\mathbf{P} 4_{1} 32$ | $I 4_{1} 32$ |  |  |  |  |
| $\overline{43 m}$ | $\mathbf{P} \overline{4} 3 m$ | $F^{\frac{1}{4} 3 m}$ | I43m | P $\overline{43} \boldsymbol{n}$ | $\overline{\mathrm{F} 3} \boldsymbol{c}$ | I43d |
| $m 3 m$ | $\mathrm{Pm} 3 \mathrm{~m}$ | Pn3n | Pm3n | Pn3m | Fm3m | Fm 3 c |
|  | $\mathrm{F} d 3 m$ | $\mathrm{F} d 3 c$ | Im3m | $\mathbf{I} a 3 d$ |  |  |

## Space Groups

all compounds crystallize in one or more of these space groups usually possible to find $\mathbf{P 1}$, but always try to find the highest possible symmetry.
structures observed in all 230 space groups
$\sim 95 \%$ of all structures: monoclinic, triclinic, orthorhombic
$\sim 83 \%$ of all structures: $\mathbf{P 2} / \boldsymbol{c}, \mathbf{P 1}, \mathbf{P 2}_{1} \mathbf{2}_{1} \mathbf{2}_{\mathbf{1}}, \mathbf{C} 2 / \boldsymbol{c}, \mathbf{P} \mathbf{2}_{1}, \mathbf{P b c a}$

| trigonal/rhombohedral |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | P3 | P31 | P3 ${ }_{2}$ | R3 |  |  |  |
| $\overline{3}$ | P $\overline{3}$ | R $\overline{3}$ |  |  |  |  |  |
| 32 | P312 | P321 | P3 12 | P3 21 | P3 $\mathbf{2}^{12}$ | P3221 | R32 |
| 3m | P3m1 | P31m | P3c1 | P31c | R3m | R3 $\boldsymbol{c}$ |  |
| $\overline{3} \boldsymbol{m}$ | P $\overline{3} 1 m$ | $\mathbf{P} \mathbf{3} 1 c$ | $\mathrm{P} \mathbf{3} \boldsymbol{m} \mathbf{1}$ | $\mathbf{P} \overline{3} c \mathbf{1}$ | $\mathbf{R} \overline{3} m$ | $\mathbf{R} \overline{3} c$ |  |
| hexagonal |  |  |  |  |  |  |  |
| $\frac{6}{6}$ | $\begin{aligned} & \text { P6 } \\ & \text { P } \end{aligned}$ | P6 ${ }_{1}$ | P65 | $\mathbf{P 6}{ }_{2}$ | P6 ${ }_{4}$ | P63 |  |
| 6/m | P6/m | $\mathrm{P}_{3} / \mathrm{m}$ |  |  |  |  |  |
| 622 | P622 | P6122 | $\mathrm{P6}_{5} 22$ | $\mathbf{P 6}_{2} 22$ | $\mathrm{Pb}_{4} 22$ | $\mathrm{P6}_{3} 22$ |  |
| 6 mm | P6mm | P6cc | $\mathrm{P6}_{3} \mathrm{~cm}$ | $\mathrm{Pb}_{3} \boldsymbol{m c}$ | $6 m 2$ | P6m2 | P6c2 |
| $\overline{6}$ | P62m | P62c |  | P-62 |  |  |  |
| $\overline{6} m 2$ | $\mathbf{P} \overline{6} m 2$ | $\mathrm{P} \overline{6} c 2$ | $\mathbf{P} \overline{\mathbf{6}} \boldsymbol{m}$ | P62 $\boldsymbol{c}$ |  |  |  |
| 6/mmm | P6/mmm | P6/mec | $\mathrm{P}_{3} / \mathrm{mcm}$ | $\mathrm{P}_{3} / \mathrm{mmc}$ |  |  |  |


|  | Symmetry |
| :---: | :---: |
| $\mathbf{7}$ crystal systems: | point symmetry of external lattice |
| $\mathbf{1 4}$ Bravais lattices: | translational symmetry of lattice points |
| $\mathbf{3 2}$ point groups: | point symmetry of external crystal |
| $\mathbf{2 3 0}$ space groups: | translational symmetry inside crystal |
|  | molecules |




## Space Group Nomenclature

space group name comes from Bravais lattice symbol, modified for translational symmetry
easy to understand the components of many names, especially
monoclinic and orthorhombic:
$\mathbf{P} 2_{1} / c \quad$ (P 2-1 on c)
primitive unit cell (1 lattice point)
2-fold screw axis along $b$ (unique axis)
$\boldsymbol{c}$ glide (translation along $\mathbf{c}$ axis) in ac plane ( $\perp$ to $\mathbf{b}$ )
Pbca primitive unit cell (1 lattice point)
$\boldsymbol{b}$ glide (translation along $\mathbf{b}$ axis) in be plane ( $\perp$ to $\mathbf{a}$ )
$\boldsymbol{c}$ glide (translation along $\mathbf{c}$ axis) in ac plane ( $\perp$ to $\mathbf{b}$ )
$\boldsymbol{a}$ glide (translation along a axis) in ab plane ( $\perp$ to $\mathbf{c}$ )
Standard and Non-standard Settings
sometimes a space group that is not on the list of 230 is given
in a publication
some space groups can be derived which are identical with
another space group $\Rightarrow$ choice depends on convention
$\mathbf{P} \mathbf{2}_{\mathbf{1}} / \boldsymbol{a}$ identical with $\mathbf{P} \mathbf{2}_{1} / c$ switching a and $\mathbf{c}$ label in monoclinic
does not change the symmetry
$\mathbf{P} \mathbf{2}_{\mathbf{1}} / \boldsymbol{n}$ alternate setting of $\mathbf{P} \mathbf{2}_{1} / \boldsymbol{c}$ c $\mathbf{c}^{\prime}$
$\boldsymbol{\beta}$ closer to $90^{\circ}$ preferred
$\mathbf{P n a m}$ same as $\mathbf{P n m a}$

## Equivalent Positions, Asymmetric Unit and Z

equivalent positions are divided into:
general positions
special positions
asymmetric unit along with general and special positions allows an interpretation of $\mathbf{Z}$ (number of molecules in unit cell), and possible molecular symmetry

## Equivalent Positions

space groups used to locate symmetry related atoms in unit cell for example, if a benzene ring is located on a mirror:
locate $3 \mathbf{C}$ and $3 \mathbf{H}$,

asymmetric unit is the smallest part that generates the rest of the unit cell contents by all symmetry operations of space group

| Equivalent Positions from Centering |
| :---: |
| for centered groups, add the following to each $\mathbf{P}$ general position: $\begin{array}{ll} \text { A } & \boldsymbol{x}, \boldsymbol{y}+1 / 2, z+1 / 2 \\ \text { C } & x+1 / 2, y+1 / 2, z \\ \text { F } & x+1 / 2, y+1 / 2, z \\ & x+1 / 2, y, z+1 / 2 \\ & x, y+1 / 2, z+1 / 2 \\ \text { I } & x+1 / 2, y+1 / 2, z+1 / 2 \\ \text { R } & x+2 / 3, y+1 / 3, z+1 / 3 \\ & x+1 / 3, y+2 / 3, z+2 / 3 \end{array}$ |


| Transforming Coordinates |
| :---: |
| $\bar{x}-1 / 4=-(x+1 / 4)=-(x-1 / 4+1 / 2)=-(x+1 / 2)=\bar{x}-1 / 2=\bar{x}+1 / 2$ <br> (by adding 1 ) |
| $y+1 / 4=y-1 / 4+1 / 2=y+1 / 2$ |
| $\bar{x}+1 / 4=(\bar{x}-1 / 4)+1 / 2=\bar{x}+1 / 2+1 / 2=\bar{x}$ (by subtracting 1 ) |

## Transforming Coordinates

$\left.\begin{array}{rl}\text { related by a change in } \operatorname{sign} \\ \text { related by a change in } \operatorname{sign}\end{array} \begin{array}{l}x, y, z \\ \bar{x}, \bar{y}, \bar{z}\end{array}\right] \begin{aligned} & \bar{x}+1 / 2, y+1 / 2, \bar{z} \\ & x+1 / 2, \bar{y}+1 / 2, z\end{aligned}$
finally, change to preferred setting $\mathbf{P} 2_{1} / \boldsymbol{c}$; switch $\boldsymbol{x}$ and $\boldsymbol{z}$

$$
\begin{aligned}
& x, y, z \\
& \bar{x}, \bar{y}, \bar{z} \\
& \bar{x}, y+1 / 2, \bar{z}+1 / 2 \\
& x, \bar{y}+1 / 2, z+1 / 2
\end{aligned}
$$

| Transforming Coordinates $\begin{aligned} & \begin{array}{l} \bar{x}-1 / 4=-(x+1 / 4)=-(x-1 / 4+1 / 2)=-(x+1 / 2)=\bar{x}-1 / 2=\bar{x}+1 / 2 \\ \quad(\text { by adding } 1) \\ y+1 / 4=y-1 / 4+1 / 2=y+1 / 2 \end{array} \end{aligned}$ |
| :---: |



## Special Positions


if an object is located at $\boldsymbol{x}, \boldsymbol{y}, \boldsymbol{z}=\mathbf{0}, \mathbf{0}, \mathbf{0}$;
only unique point generated by symmetry is at $0,1 / 2,1 / 2$

$$
\begin{array}{ll}
\text { also true for: } \quad \begin{array}{l}
\mathbf{0}, \mathbf{0}, 1 / 2
\end{array} \longrightarrow 0,1 / 2,0 \\
& \longrightarrow 1 / 2,0,1 / 2 \longrightarrow 1 / 2,0 \\
1 / 2,0,0 & \longrightarrow 1 / 2,1 / 2,1 / 2
\end{array}
$$

## Special Positions

| $\mathbf{0}, \mathbf{0}, \mathbf{0}$ | $\mathbf{0}, 1 / 2,1 / 2$ |
| :---: | :---: |
| $\mathbf{0}, \mathbf{0}, 1 / 2$ | $\mathbf{0}, 1 / 2, \mathbf{0}$ |
| $1 / 2, \mathbf{0}, 1 / 2$ | $1 / 2,1 / 2,0$ |
| $1 / 2, \mathbf{0}, \mathbf{0}$ | $1 / 2,1 / 2,1 / 2$ |

note: an object (molecule) at a special position has to have the same symmetry as the special position
in $\mathbf{P} \mathbf{2}_{1} / c$, a center of symmetry
$\mathbf{Z}=\mathbf{4}$ for an object on a general position in $\mathbf{P} \mathbf{2}_{1} / \boldsymbol{c}$
$\mathbf{Z}=\mathbf{2}$ for an object on a special position in $\mathbf{P} \mathbf{2}_{1} / \boldsymbol{c}$ asymmetric unit is $1 / 2$ of the molecule

## Special Positions

an atom on a special position has at least one fixed coordinate; part of the atom generates the rest:
one fixed position (axis $\perp$ to plane) for an atom on a mirror two fixed positions (other axes) for an atom on a rotation axis three fixed positions for an atom on an inversion center


## International Tables for Crystallography

International Tables for Crystallography

| $P 2_{1} / c$ | $C_{2 k}^{b}$ | $2 / m$ | Monoclinic |
| :--- | :--- | :--- | ---: |
| No. 14 | $P 12_{1} / c 1$ |  |  |

esigee axis $b$, cete choce 1


0 one 1





