

*BCH 447 : Practical Metabolism*  
*Course Outline*

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# BCH 447

## *Course Description:*

A selection of metabolic experiments that include metabolism of carbohydrates, lipids, and proteins .

# Safety in the Lab. :



Safe practices in the biochemistry laboratory are of great significance.

Students must adopt safe and skillful methods while working in the lab. To achieve this goal the student must obtain the **prerequisite knowledge** of properties of materials present in the lab and must be **acquainted** with different **hazards and harmful** effects associated with their improper handling.

# Regulations

- Immediately inform your instructor in case of any accident.
- Do not eat, drink, chew, or smoke in the laboratory.
- Do not depart from the lab leaving an experiment unattended. **If you need to leave the lab you must inform your instructor before leaving the lab.**
- Not sticking to these rules will result in instant removal from the lab.



## Precautions

- You must come to the lab with a serious **awareness** and consideration for others in the lab.
- You must acquaint yourself with **safety equipment location, acid-base neutralizing agents, eye washer, fire extinguisher, emergency shower, broom & dustpan and broken glass container.**
- If you are unsure of anything, always ask your instructor.



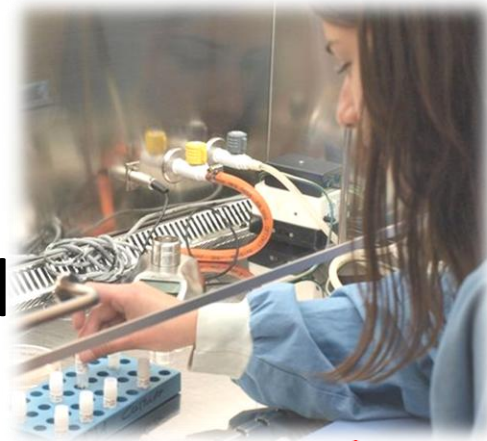
- While handling all electrical and heating equipments extra precautions must be taken to prevent shocks and burns.
- Do not handle broken glassware with your bare hands.
- You must wash your hands with soap after finishing the experiment



# Personal clothing



- Due to the harmful nature of some chemicals, the student should wear proper and suitable clothing.
- You must wear a lab coat to help keep clothes protected. Open toed shoes must not be worn because they cannot protect you against chemical spills.
- Long hair should be tied back to avoid interference with motion or observation.



- **Chemicals:**

In the experiments you will use different Chemicals. Hence care must be taken to **avoid skin contact**.

While handling these chemicals, **avoid eye and face contact**.

In case of acid or base contact with your skin, **wash** it with large amount of clean, **cold water** and **inform** the instructor immediately.

For your own protection, neutralize acid or base spills before cleaning them up.





- **Laboratory Notebooks**

- use the notebook to keep record of all primary **data** and **observations**. You must **organize** your notebook every week before coming to the lab by writing the title of the experiment on a new page, with important equations or formulae from the lab manual, and all necessary calculations involving solution preparations, molar masses, etc.

- *Try to understand theoretical concepts and particular instructions given by your instructor before the experiment.*

- Do not depend on your memory and write down all observations for example color and phase changes, etc

- Excellent note taking in the lab is an important skill that can be learned with little effort and practice.

# Course Grading :

- In lab short quizzes weekly -15 marks
- Evaluation of Lab reports weekly - 15 marks
- Midterms (Theoretical -10 marks and Practical -15 marks)
- Practical performance- 5 marks
- final exams (Theoretical -15 marks and Practical -25 marks)



# How To write a scientific report?

- The laboratory reports are major **written assignments** and should be written in the form of a **scientific paper**. The laboratory reports should contain the following sections:
- Title Page
- Brief Introduction
- Materials and Method
- Results
- Discussion
- References
- All of the laboratory reports are expected to be **well written, typed in English**.



## **Introduction section:**

This part should consist of **any theoretical background information**. This section should be around 15 -20 lines.

## **Materials and Methods section:**

In this section you will write the material and methods that you used, you must also mention exact volumes, concentrations, incubation times.

## Results section:

report all your results that you get from your experiment such as calculations, exact volumes, amounts, incubation times, etc.). You should present them in a **tabulated form** so it will be easy for quick reference.

You must number and label all the tables and figures (graphs, diagrams). This way it will be easy for you to refer to them in your discussion section.

You should also include your sample calculations (if any) in the result section.



## Discussion section:

In this section you are required to give a thorough description of what happened in the experiment.

The discussion section is also where you interpret or explain your results and make conclusions.

You should refer to your tables and diagrams while explaining your results. You should compare your results to expected values (calculated or from the literature).

- Even if you obtained unexpected results, the discussion section is the section to justify or explain the reasons why you have obtained such results.
- **Please remember how you interpret your results carries more weight than the results themselves.**

- **References section:**

- In this section you will provide an alphabetical listing (by first author's last name) of the references that you actually cited in the body of your report.

- **Questions section:**

- In this section you will answer all the Questions after the experiment.

# *Course References:*



- Laboratory Manual in General Biochemistry :  
University of Santo Tomas.
- Johnson, C.R., Miller, M.J., Pasto, D.J. (1998).  
Experiments and Techniques in Biochemistry.  
United States of America: Prentice Hall Inc.
- Landgrebe, J.A. (1993). Theory and Practice in the  
Biochemistry Laboratory: with Microscale and  
Standard Scale Experiments. California:  
Wadsworth Inc



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Download of ALL the lectures and Lab sheet  
from website :

<http://fac.ksu.edu.sa/alaamal/home>



*Good Luck*