

Experiment 3: D- Xylose Absorption Test

Procedure:

- Set up 7 test tube the Please follow the table:

Chemical	Test		Test		standard	Blank
	1	2	3	4	5	
Sample A	0.1 ml	0.1 ml	-	-	-	-
Sample B	-	-	0.1 ml	0.1 ml	-	-
Working standard	-	-	-	-	0.1 ml	-
O-toluidine reagent	3 ml	3 ml	3 ml	3 ml	3 ml	3 ml

- Cover all tube by aluminum foil then place all tube in water bath at boiling degree for 5 min.
- Read the absorption of 6 tubes against blank at 475 nm. and record your result

Results:

Test tube		Absorption at 475 nm
Sample A	Test 1	
	Test 2	
Sample B	Test 3	
	Test 4	
Standard	Test 5	

Calculation :

- Concentration of Standard = 0.5g/ml

Concentration of D-Xylose = $\frac{\text{Means Ab Test}}{\text{Means Ab Std.}} \times \text{Concentration of Standard} \times 10(\text{D.F}) = \dots\dots\dots$

Concentration of D-Xylose **in sample 1**

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Concentration of D-Xylose **in sample 2**

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Example:

- Means Absorption of sample 1= 0.145
- Means Absorption of Standard= 0.357

- Conc. Of urine D-xylose = $0.145 \div 0.357 \times 0.5 \times 10 = 2.03 \text{ g/ml}$