Report No. (7)

DETERMINATION THE MOLAR MASS OF VOLATILE LIQUID

tudent Names:			Section No:	
The objective of the	e experiment:			
esults and calculat				••••••
m ₁ g	m ₂ g	V ml	тк	P atm

1. Calculate the mass of the unknown gas (it equals the mass of the condensed vapor)

2. Calculate the molar mass of the unknown substance

 $R = 82.06 \text{ cm}^3.\text{atm/K.mol OR } R = 0.0821 \text{ L.atm/K.mol}$