




SODIUM DODECYL SULPHATE -POLYACRYLAMIDE
GEL ELECTROPHORESIS
[SDS-PAGE]

Lab#8
BCH 333



SDS-PAGE:

- ***SDS-PAGE***: Sodium Dodecyl Sulfate-PolyacrylAmide gel Electrophoresis.
- The separation of macromolecules in an electric field is called ***electrophoresis***.
- For separating proteins by electrophoresis uses a discontinuous polyacrylamide gel.
- SDS-PAGE aim to separate and identify proteins according to their **molecular weight**.

- ***Applications:***
 - Quantifying.
 - Comparing (relative abundance).
 - Characterizing proteins (e.g. determining MW of proteins).
 - Checking purity of protein samples.

Principle of SDS-PAGE:

- In general, fractionation by gel electrophoresis is based on differences in size, shape and net charge of macromolecules (in molecules native condition using native gel electrophoresis).
 - Systems where you separate proteins under native conditions cannot distinguish between these effects and therefore proteins of different sizes may have the same mobility in native gels.
- ➔ In SDS-PAGE, this problem is overcome by:
1. Using an anionic detergent Sodium Dodecyl Sulfate (SDS), which binds strongly to most proteins.
 2. Using disulfide reducing agent such as β -mercaptoethanol or DTT (dithiothreitol).

Principle of SDS-PAGE:

- Sodium Dodecyl Sulfate (SDS) is a detergent which **denature proteins** by binding to the hydrophobic regions, all non-covalent bonds will disrupt, and the proteins acquire a **negative net charge**.
 - A Concurrent treatment with a disulfide reducing agent such as β -mercaptoethanol or DTT (dithiothreitol) further breaks down the strong bound disulfide.
- ➔ *So, the proteins samples are having uniformed structure and charge (-ve).*

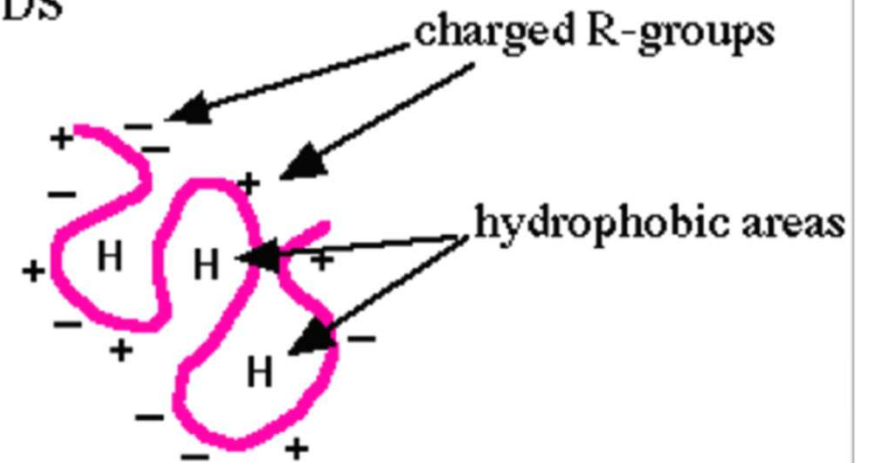
Effect of SDS

What is SDS?

The carbon tail has a neutral charge
and the sulphate group is negatively charged



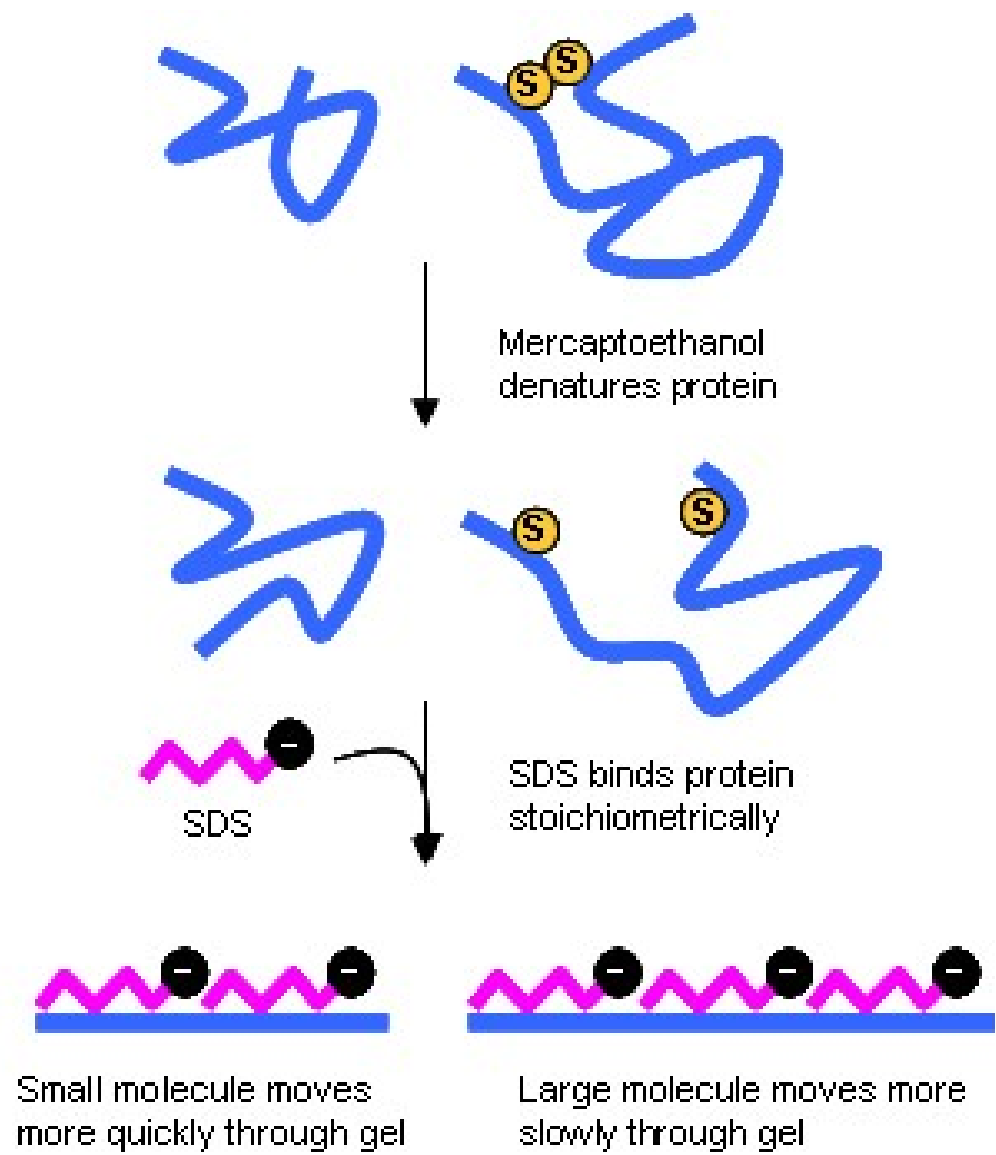
BEFORE SDS



AFTER SDS



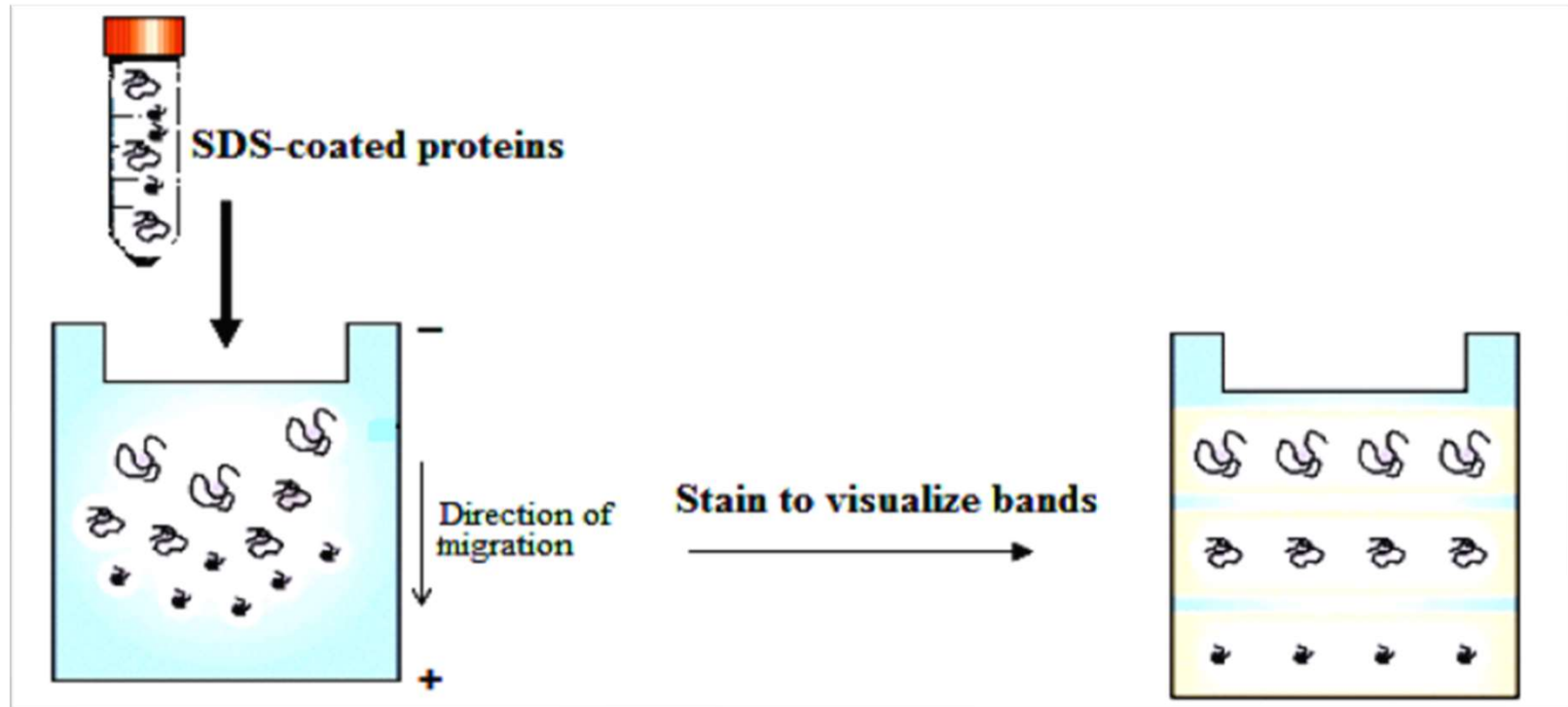
Effect of β -mercaptoethanol



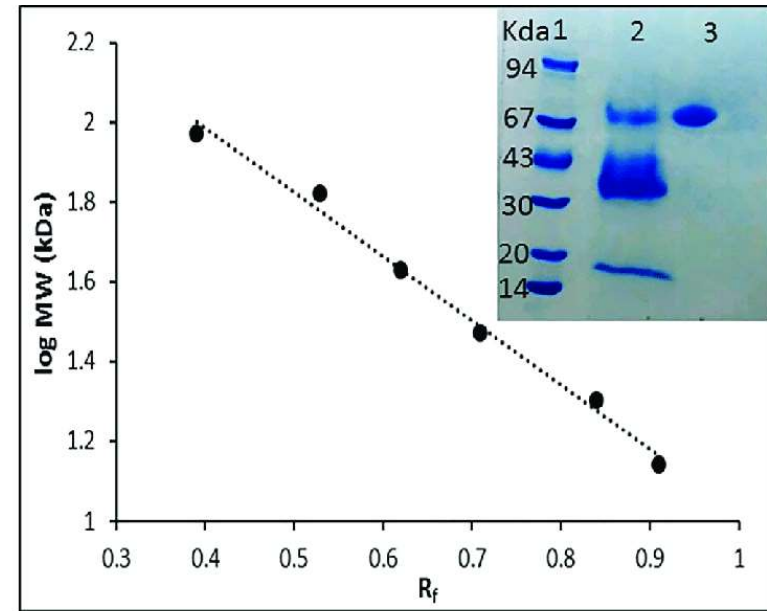
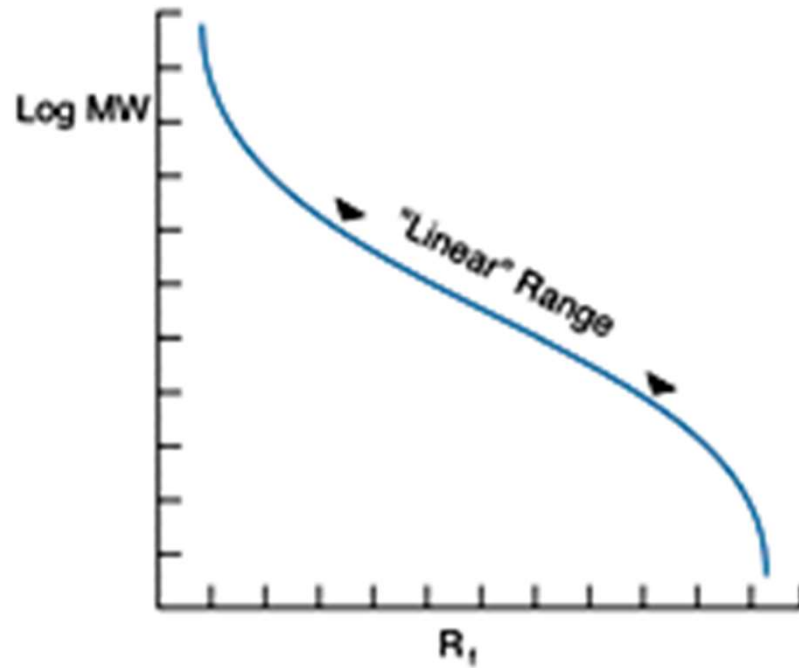
Principle of SDS-PAGE:

- Because the *charge-to-mass ratio is nearly the same* among SDS-denatured polypeptides, the final separation of proteins is dependent almost entirely on the differences in relative molecular mass of polypeptides (separation will depend on their molecular weight only).
 - i.e The electrophoretic mobility of the molecules is now considered to be a function of their size (the protein samples are having uniformed structure and charge → the separation will depend on their molecular weight only).
- Polyacrylamide gels *restrain larger* molecules from migrating *as fast as smaller molecules*.
- Thus, small proteins migrate faster through the gel under the influence of the applied electric field, whereas large proteins are successively retarded, due to the sieving effect of the gels.
 - i.e the migration of the (SDS-treated proteins), towards the anode (+) is inversely proportional to the logarithms of their molecular weights, or more simply expressed:

Small proteins migrate faster through the gel



SDS-treated proteins have very similar charge-to-mass ratios, and similar shapes. During SDS-PAGE, the rate of migration of SDS-treated proteins is effectively determined by **molecular weight**.



Graph of log MW vs. R_f is sigmoidal, it is nearly linear for a range of molecular weights excluding very small and very large MW.



Practical part

OBJECTIVES:

- To separate and calculate the molecular size of proteins by comparing the separated bands with known standard molecular weight marker.



SDS-Polyacrylamide Gel Electrophoresis preparation

1. Sample Preparation:

- 40 μ l of protein sample + 10 μ l of disruption buffer → boil the mixture 3 min at 99 °C.
- ***Disruption buffer [loading buffer] contain:***
 - 10% (w/v) SDS [?]
 - 1M Tris/HCl, pH 6.8
 - Glycerol [?]
 - β -Mercaptoethanol [?]
 - Bromophenol blue [?]

2. Polyacrylamide Gel (Acrylamide stock):

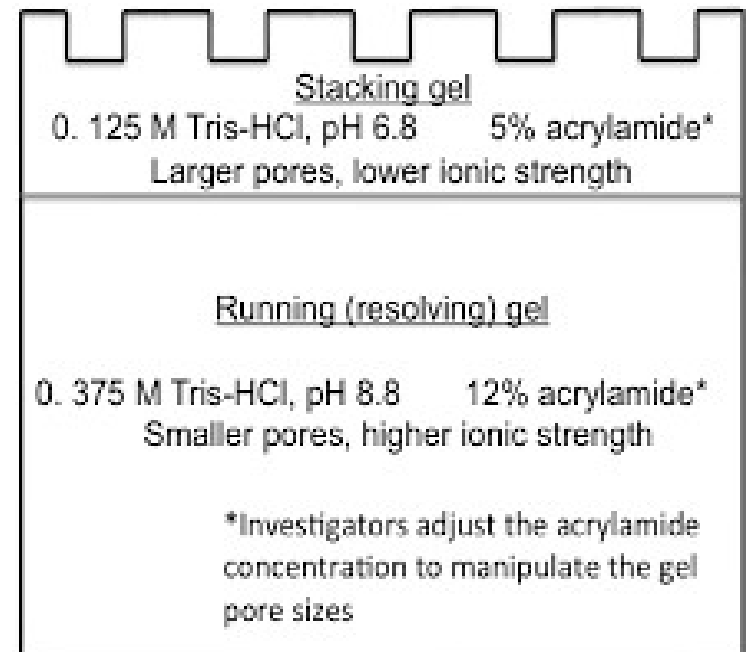
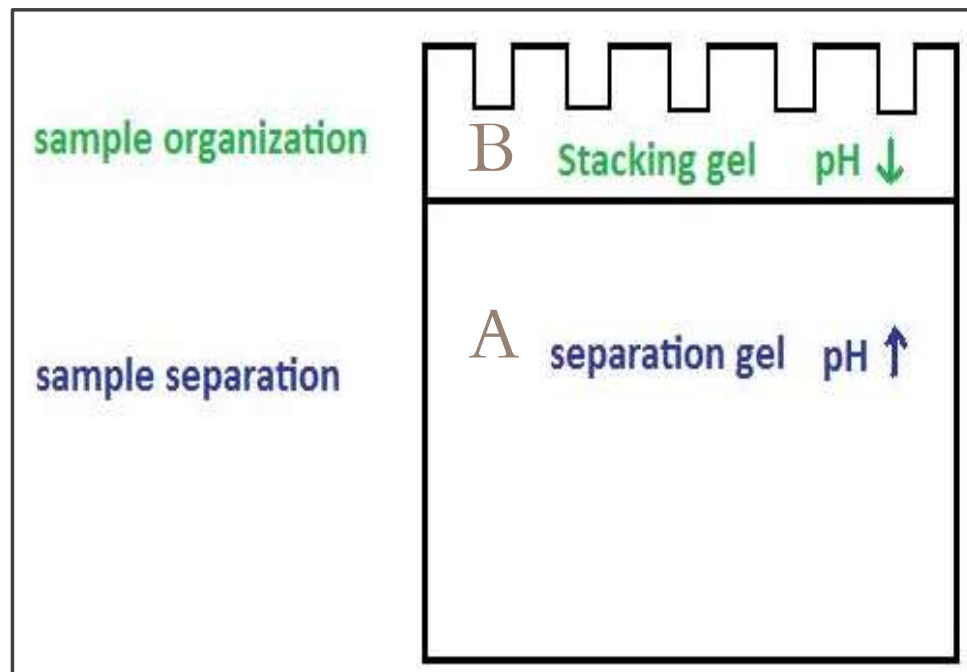
- The polyacrylamide gel is formed by co-polymerization of acrylamide and a *cross-linking* by N,N'-methylene-bis-acrylamide "bis-acrylamide".
- Polyacrylamide = acrylamide + bis-acrylamide.

- *To polymerize the gel a system:*
- 1- Ammonium PerSulfate (APS) [initiator].
- 2- TetraMethylene Ethylene Diamin (TEMED) [catalyst].

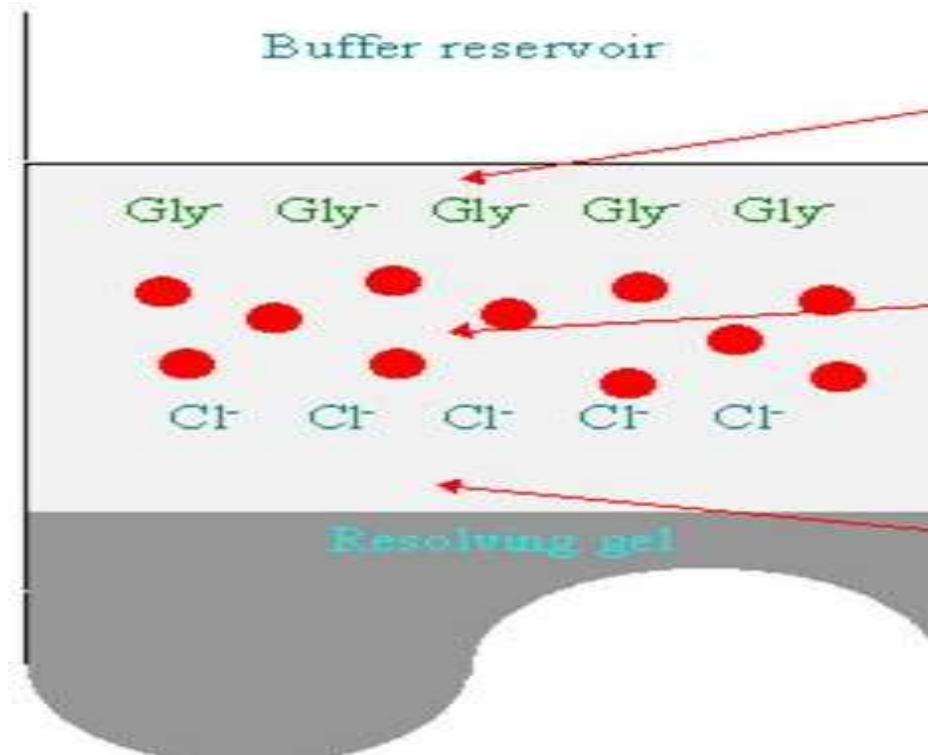
2. Polyacrylamide Gel:

- It is composed of *two* gels layers:
 - A. *Separation gel:*
 - HIGH concentration of the acrylamide stock and pH value.
 - B. *Stacking gel:*
 - LOW concentration of the acrylamide stock and pH value.

The gel's two layers:



Proteins are sandwiched between the Cl^- ions and the glycine molecules.



A. Separation Gel Contents:

- For 10 ml of separation gel:

<i>Acrylamide%</i>	6%	8%	10%	12%	15%
H₂O	5.2 ml	4.6 ml	3.8 ml	3.2 ml	2.2 ml
Acrylamide/Bis-acrylamide (30% / 0.8% w/v)	2 ml	2.6 ml	3.4 ml	4 ml	5 ml
1.5 M Tris (pH=8.8)	2.6 ml	2.6 ml	2.6 ml	2.6 ml	2.6 ml
10% w/v SDS	0.1 ml	0.1 ml	0.1 ml	0.1 ml	0.1 ml
10% w/v Ammonium persulfate (APS)	100 µl	100 µl	100 µl	100 µl	100 µl
TEMED	10 µl	10 µl	10 µl	10 µl	10 µl

*Note: APS and TEMED must be added right before each use.

- The acrylamide percentage in SDS PAGE gel depends on the size of the target protein in the sample.

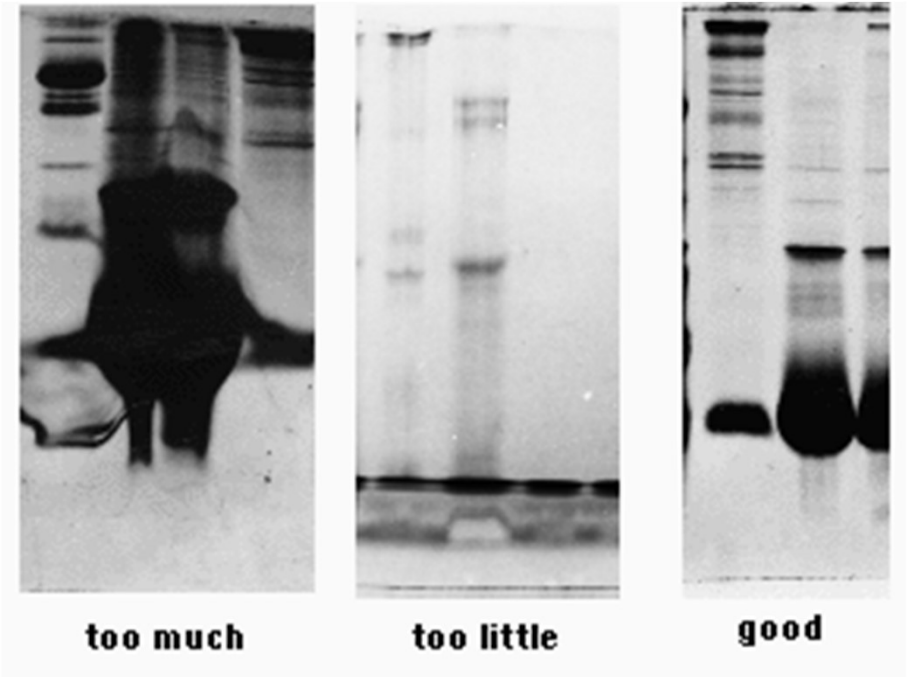
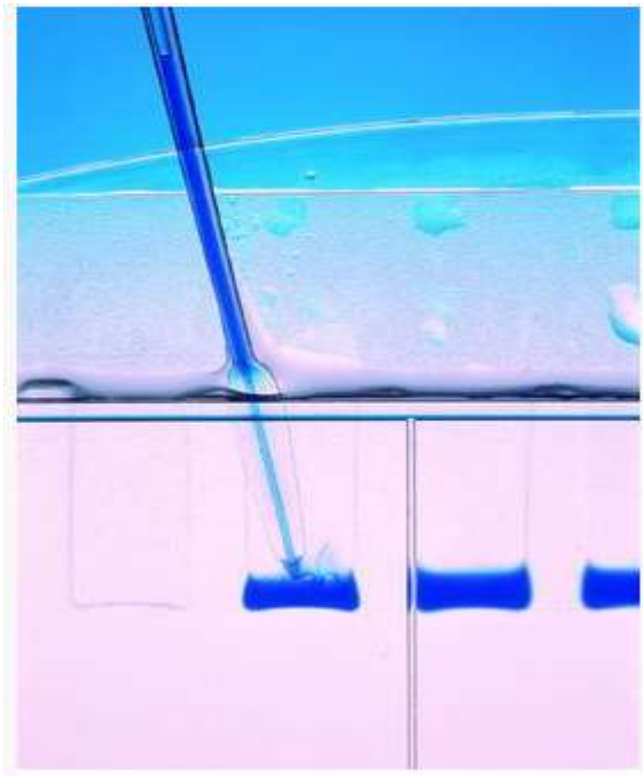
Acrylamide %	M.W. Range
7%	50 kDa - 500 kDa
10%	20 kDa - 300 kDa
12%	10 kDa - 200 kDa
15%	3 kDa - 100 kDa

B. Stacking Gel Contents:

- For 5 ml of stacking gel:

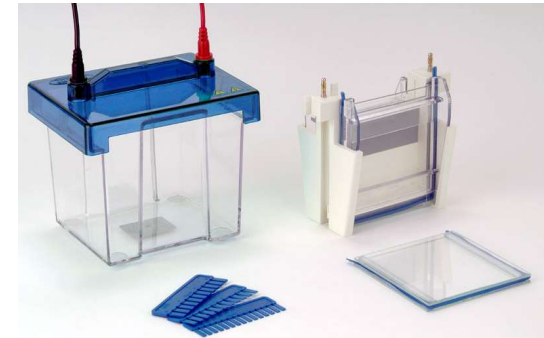
	<i>Volume</i>
<i>H₂O</i>	2.975 ml
<i>Acrylamide/Bis-acrylamide (30% / 0.8% w/v)</i>	0.67 ml
<i>0.5 M Tris-HCl (pH=6.8)</i>	1.25 ml
<i>10% w/v SDS</i>	0.05 ml
<i>10% w/v Ammonium persulfate (APS)</i>	0.05 ml
<i>TEMED</i>	0.005 ml

3. Sample Loading:



4. *Running the gel using, Running buffer (pH 8.3):* →

- It helps is deliver the electric current through the gel.
- It contains:
 - Tris-HCl.
 - Glycine.
 - SDS.



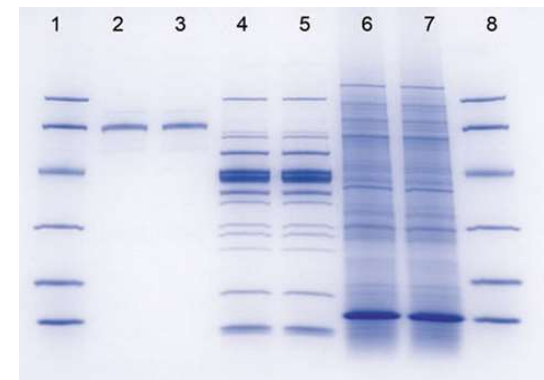
5. *Stain the gel using staining buffer :* →

- It contains:
 - Glacial acetic acid
 - Methanol
 - **Coomassie brilliant blue 250-R.**



6. *De-stain the gel using De-staining buffer:* →

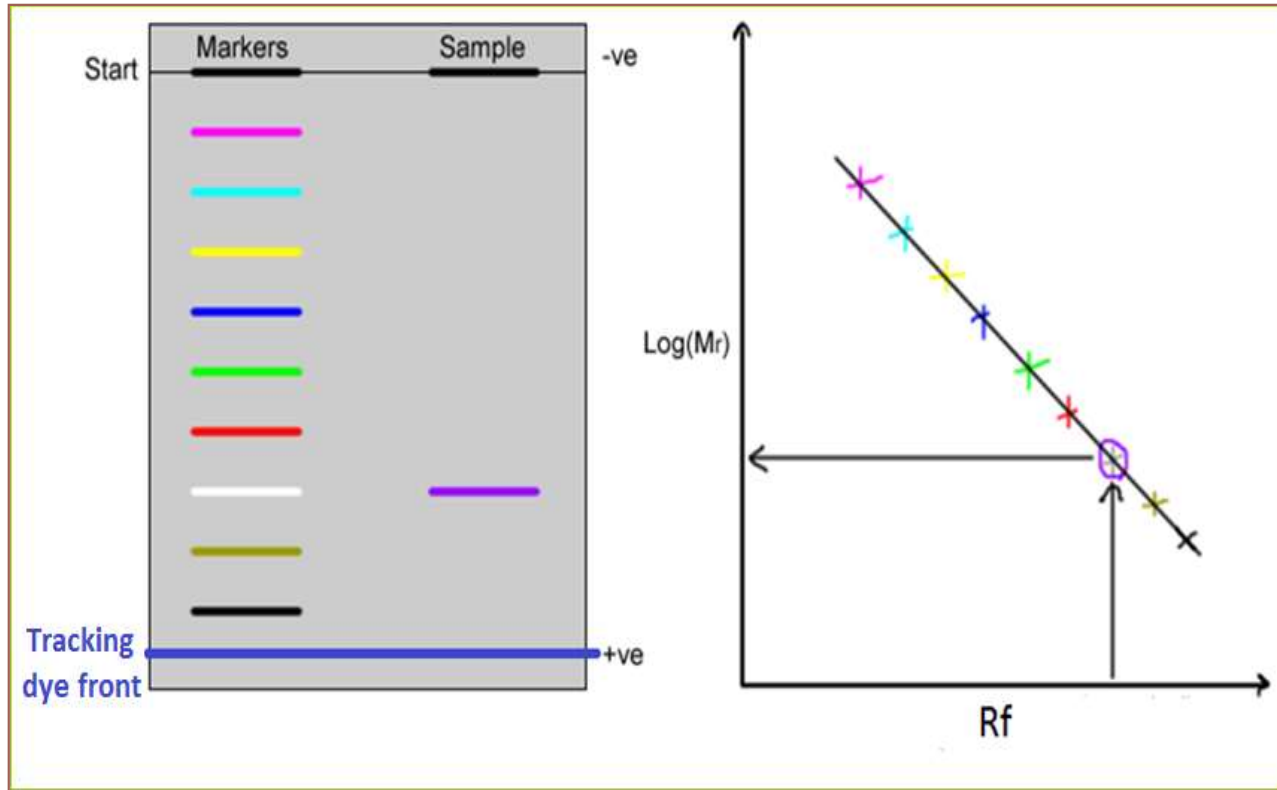
- It contains:
 - Glacial acetic acid
 - Methanol





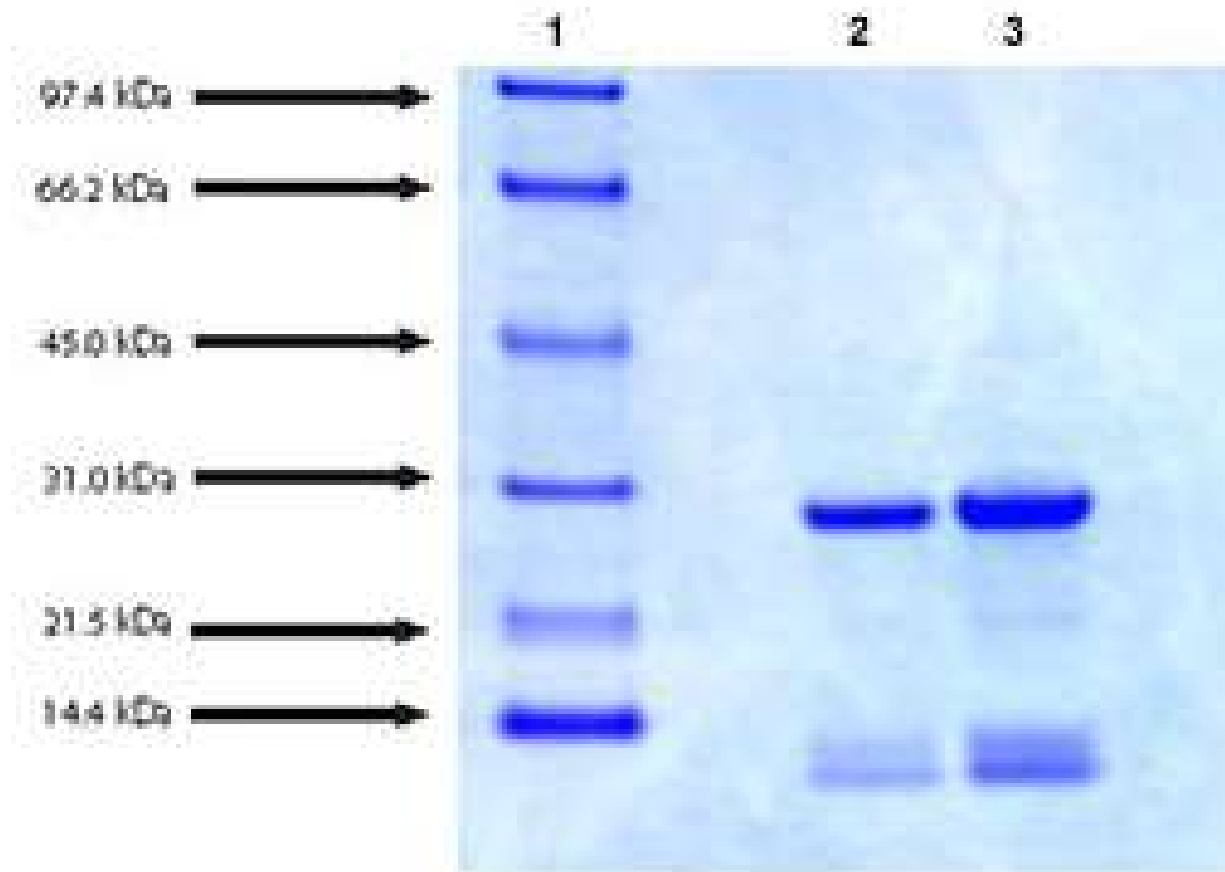
Data Analysis

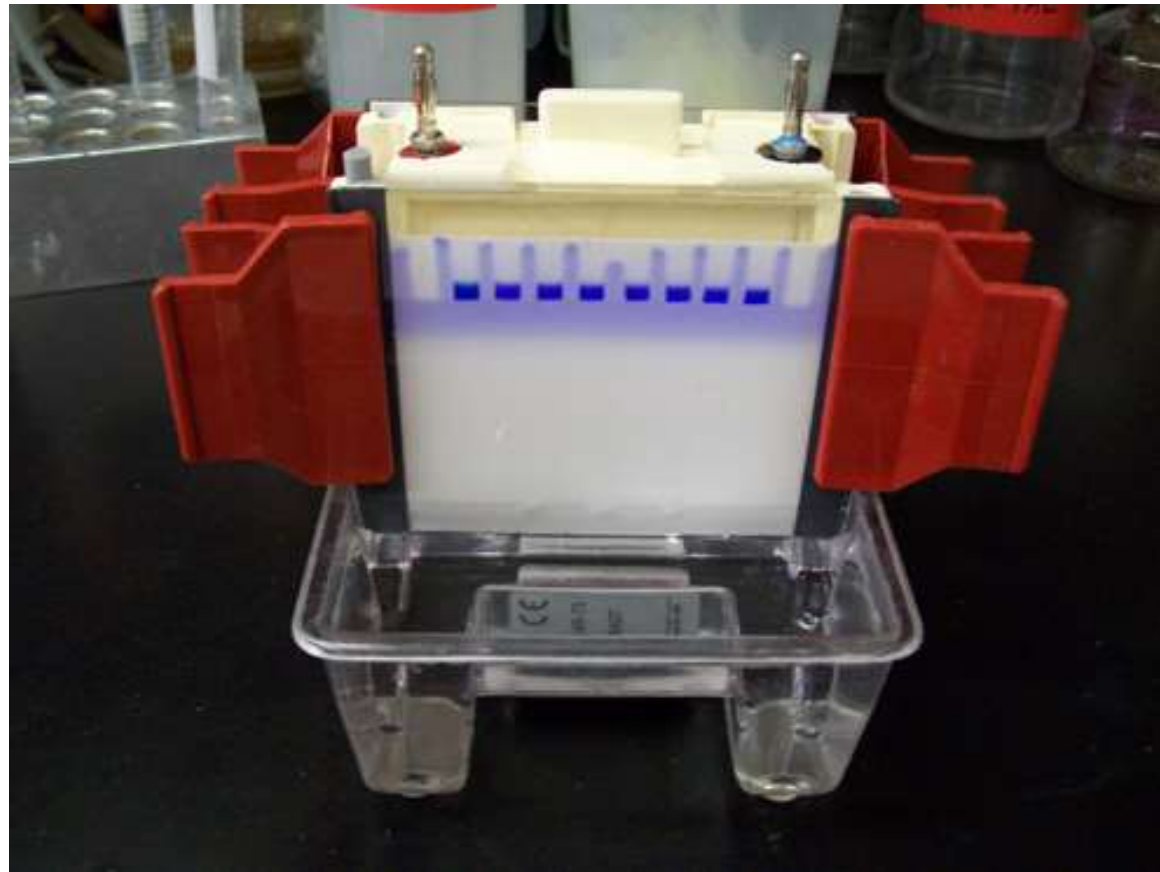
For Molecular Weight Determination:



$$- R_f = \frac{\text{Distance of migration of sample}}{\text{Distance moved by tracking dye}}$$

For Molecular Weight Determination cont':

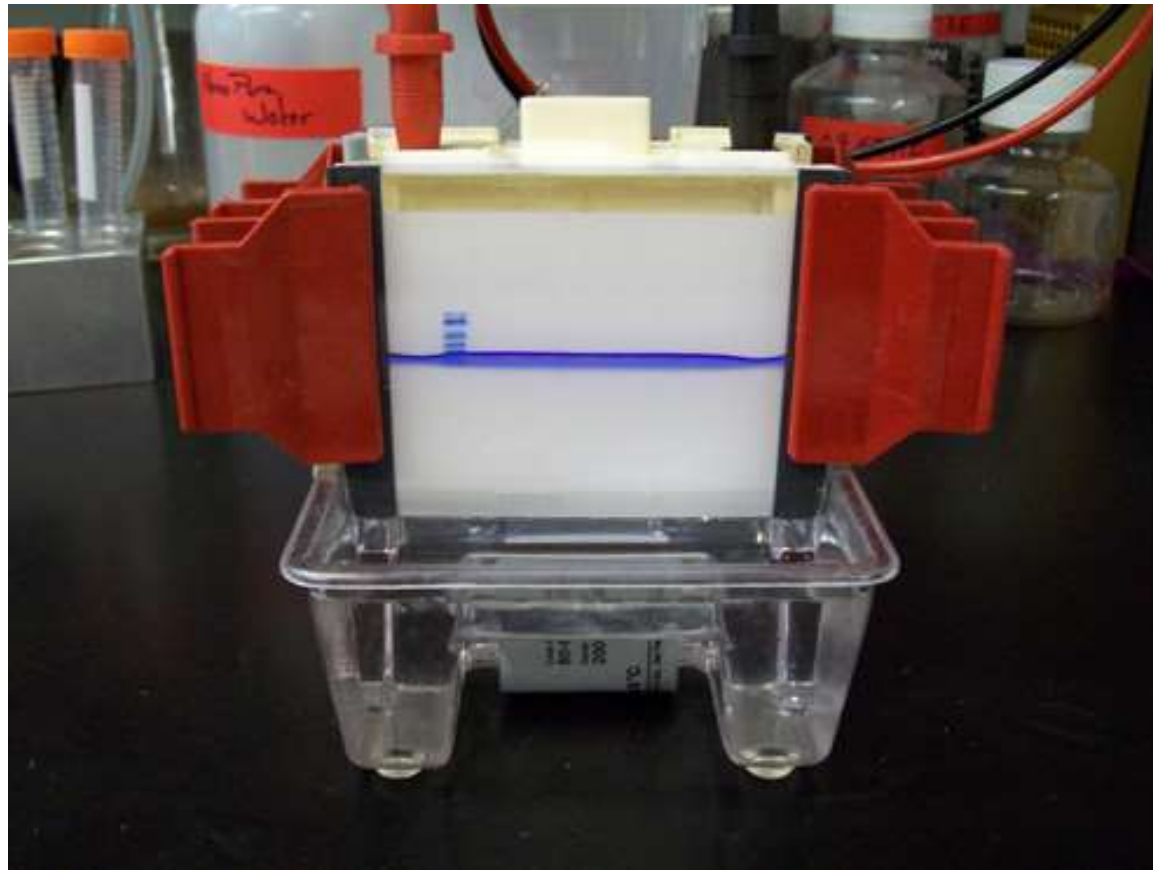




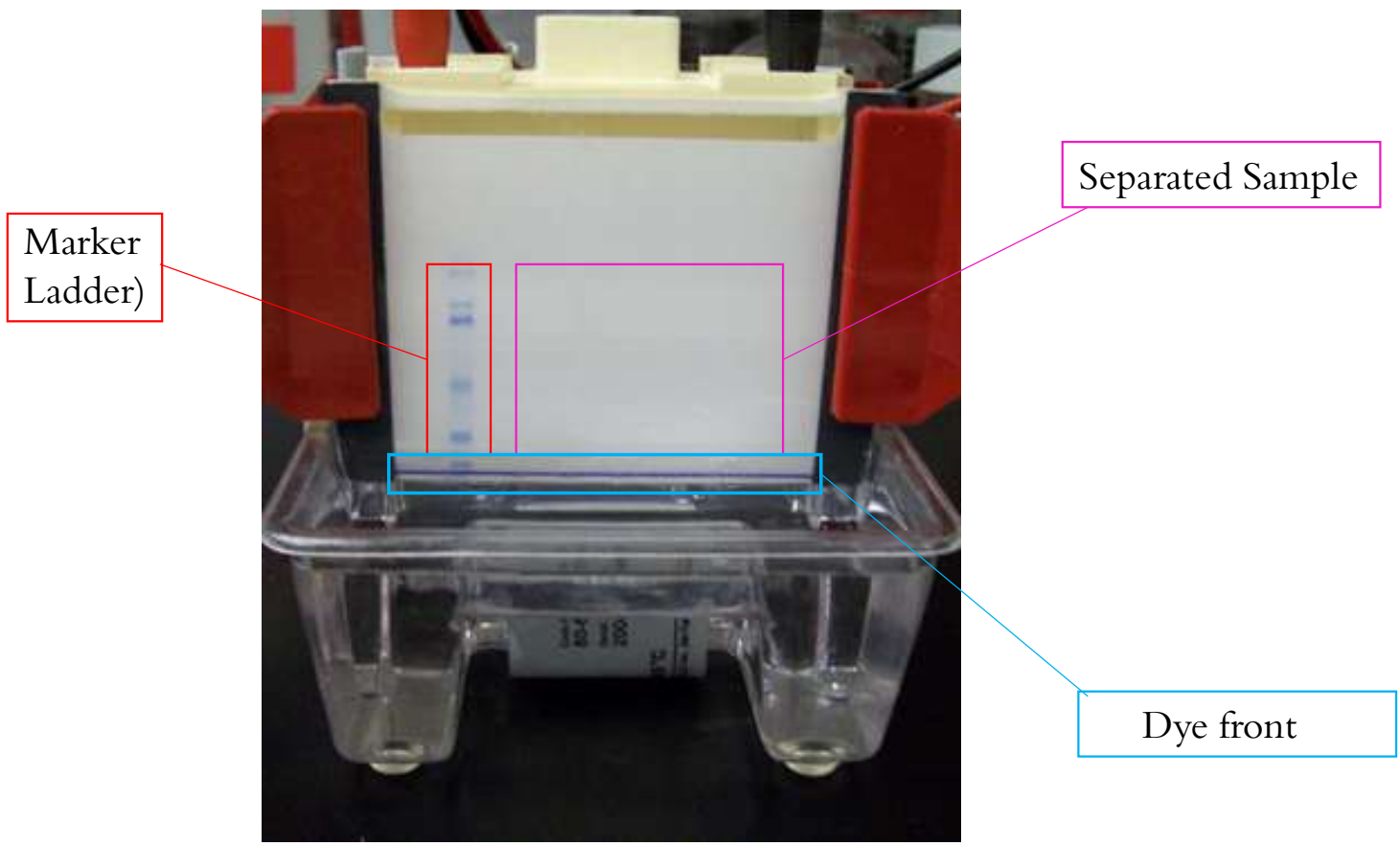
At beginning



After a while



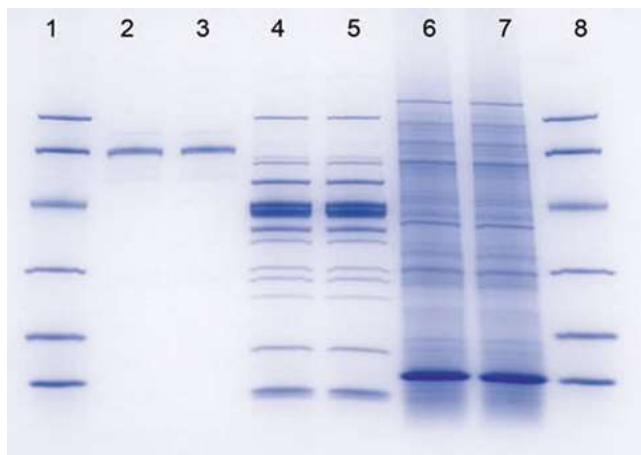
After more mins.



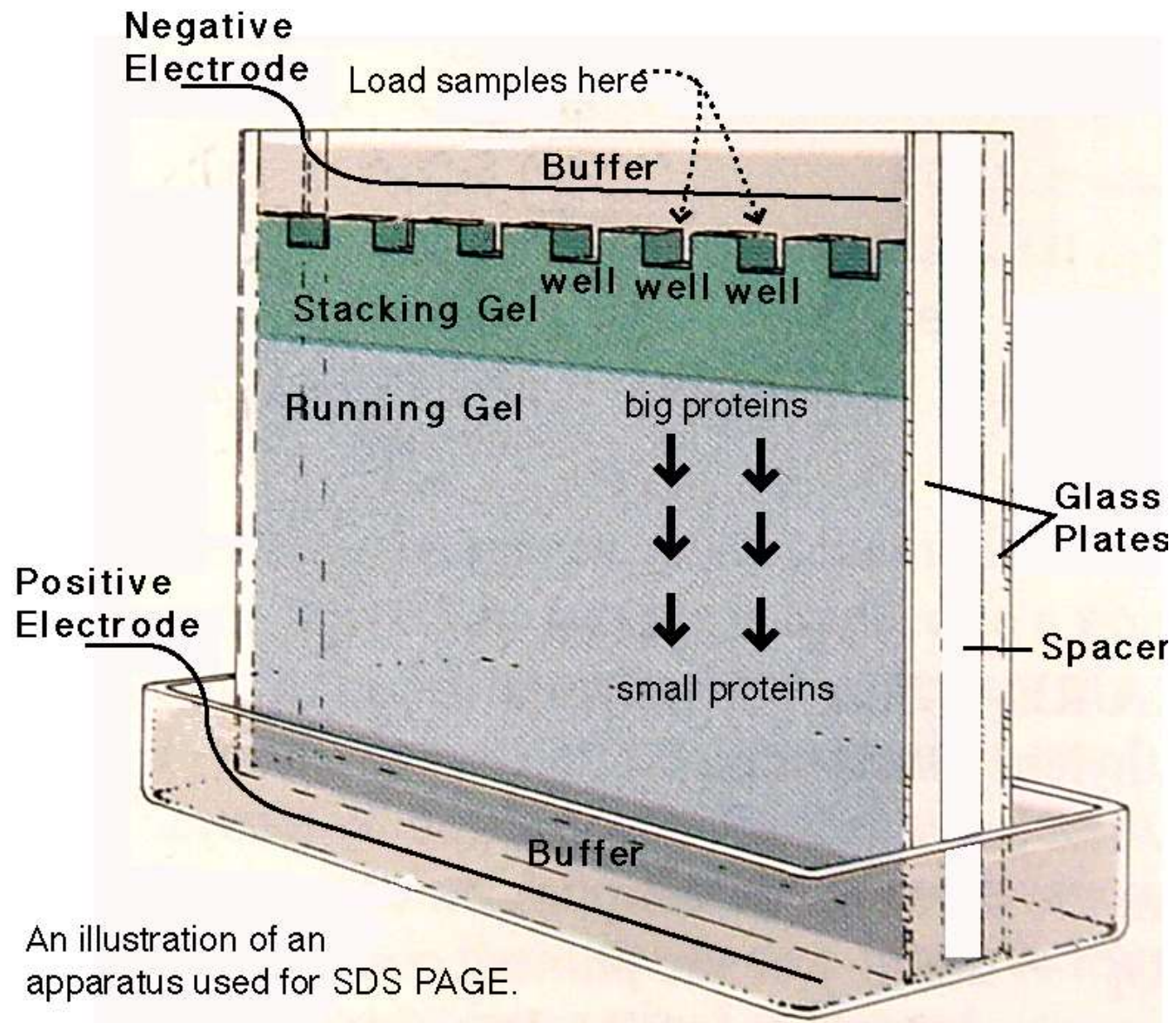
At the end



Staining



De Staining



Supporting Materials:

- *SDS PAGE principles - simple animated tutorial:*

<https://www.youtube.com/watch?v=3CrzY7jb9fQ>

https://www.youtube.com/watch?v=i_6y6Z5UvwE