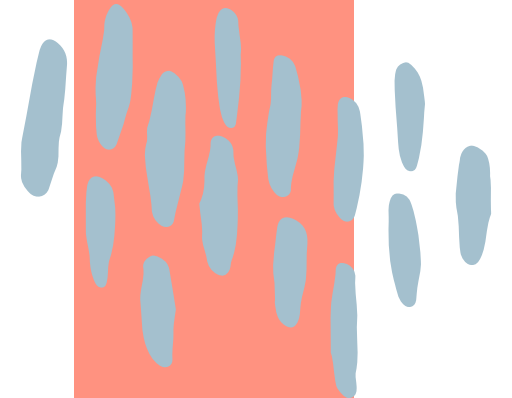


BCH 447 Practical Metabolism

Estimation of Glutathione in Plasma

Objectives

To estimate the amount of glutathione in serum sample using a standard curve



Glutathione

- It occurs naturally as a **tripeptide** (gamma-**glutamylcysteinylglycine**)
- It has **two form**: reduced form (**GSH**) and oxidized form (**GSSG**)
- It has reducing and nucleophilic properties, those properties due to the presence of **Sulfhydryl group or thiol group (-SH)**.

- **Reducing**, means it will give the other protein the H ion to maintain them in reducing form.
- **Nucleophilic**, means GSH will loss the H ion then become nucleophilic (**contain negative charge**) can react with the compound which carrying positive charge.

Functions of Glutathione

- It acts as anti-oxidant, blood booster, and cell detoxifier in the body.

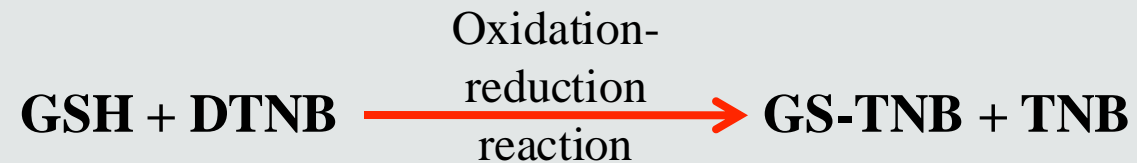
GSH is crucial to a variety of life processes, including

Amino acid transport across membrane	Maintenance of the SH level of proteins	Detoxification of xenobiotics such as drugs	Removal of hydroperoxide and free radicals.	Coenzyme with glutathione peroxidase, glutathione S-transferase and thiol transferase.
---	--	--	--	---

- Physiological values of intracellular GSH generally range **from 1 to 10 mM**.
- Glutathione deficiencies have been linked to many forms of cancer.

Assay Principle

- The principle of the assay is based on the **oxidation of the reduced form of glutathione (GSH)** by the aromatic disulphide compound [5,5-dithiobis-2-nitrobenzoic acid (DTNB)] to form GS-TNB and the aromatic thiol, 5-thio-2-nitrobenzoic acid (TNB).



- **Yellow color is formed (TNB)** It can be measured at **412nm** and is proportional to the amount of glutathione present in the sample.

Note: DTNB is irritating to the eyes, respiratory system and skin. Wear suitable protective clothing.

Note : This method is also useful to determine the GSSG.

First , GSSG will be converted to GSH by glutathione reductase (GR) and NADPH assay then it will react with DTNB reagent

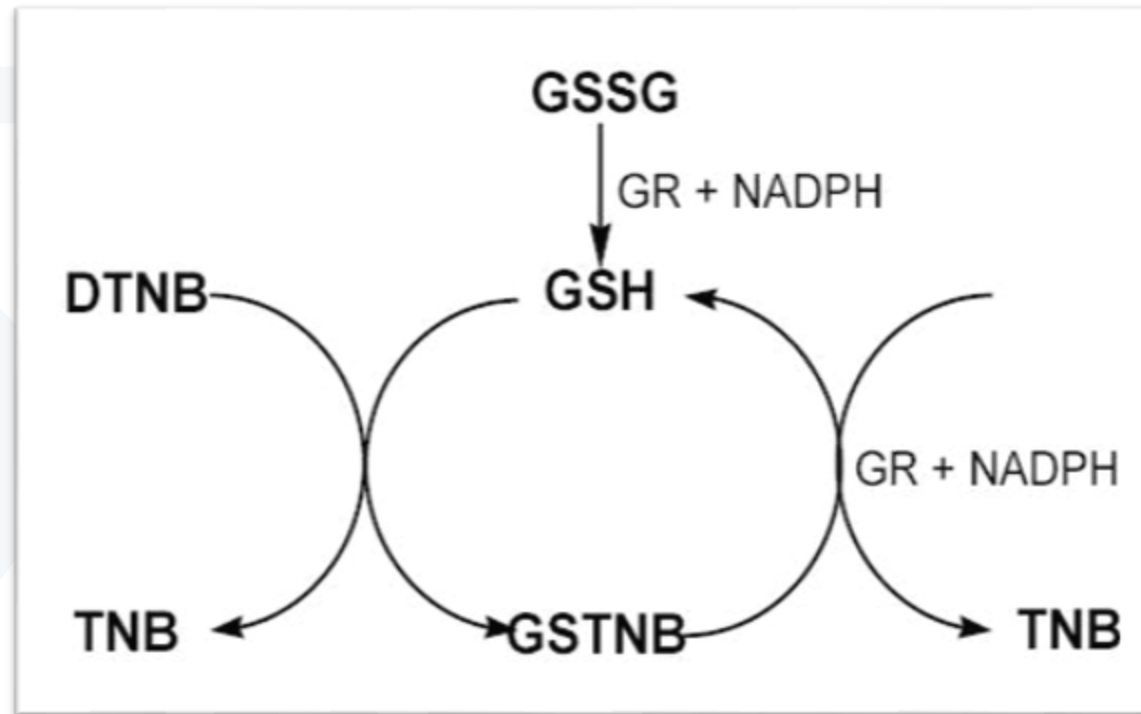


Figure 3. Glutathione (GSH) reaction with DTNB reagent

Method

1- Prepare serial GSH concentration, as the following table:

Tube No.	GSH stock Solution (ml)	Phosphate Solution (ml)	Total Volume (ml)	GSH Conc. (mg/dl)	GSH Conc. ($\mu\text{g/ml}$)
Blank	0	3	3	0	0
1	0.6	2.4	3	2	20
2	1.2	1.8	3	4	40
3	1.8	1.2	3	6	60
4	2.4	0.6	3	8	80
5	3	0	3	10	100

2- For standard curve (Tubes from blank to 5)

In a separate test tube, take 0.5 ml of solution (from step 1) + 2 ml of Phosphate Solution + 0.25 ml DTNB + 0.25 ml H₂O.

3- *For Sample*: Take 0.3 ml of sample +2 ml of Phosphate Solution + 0.25 ml DTNB+ 0.45 ml H₂O.

**Incubate all tubes for 10min at 37°C → Read absorbance at 412 nm **

Results

Tube No.	GSH Conc (mg/dl)	Absorbance At 412 nm
1	2	
2	4	
3	6	
4	8	
5	10	
Blank	0	
Sample	From Standard Curve	

- Plot the standard curve and determine glutathione concentration from the graph expressed as $\mu\text{g/ml}$

Homework

- What are the **7 mechanisms** of antioxidant?
- What are the **three classes** of antioxidant defense systems in human?