

BCH 445- Biochemistry of Nutrition [Practical]
Estimation of Proline in Honey



Honey

- Honey is a naturally sweet and viscous fluid produced by honeybees (*Apis mellifera*) from the nectar of flowers.
- It is a supersaturated complex natural liquid that contains about 31% glucose, 38% fructose (honey also contains other sugars with lower concentration).
- In addition, there is a great variety of minor components, including phenolic acids and flavonoids, the enzymes glucose oxidase and amylase, ascorbic acid, carotenoids, organic acids, free amino acids, proteins, and α -tocopherol.
- The actual composition of honey varies, depending on many factors such as the floral source, climate, environmental conditions, and the processing it undergoes.

Component	Average
Carbohydrates	82.4g
Fructose	38.5g
Glucose	31g
Sucrose	1g
Other Sugars	11.7g
Dietary fiber	0.2g
Fat	0g
Protein	0.3g
Water	17.1g
Riboflavin (Vit. B2)	0.038mg
Niacin (Vit. B3)	0.121mg
Pantothenic acid (Vit. B5)	0.068mg
Pyridoxine (Vit. B6)	0.024mg

Figure 1. Nutrient components of honey per 100 g

Proline in honey

- There are approximately 26 free amino acids in honey.
- The **major** amino acid is **proline** (up to 80%).
- Proline content varies in different honeys according to its **floral type**.
- **Also**, Proline comes **mainly** from honey bee during the conversion of nectar into honey, which leads to a high variability of the proline content within honeys from the same **botanical source**.

Proline in honey

- The **proline content** in honey is related to the degree of nectar processing by the bees.
- Also, proline content in some cases used as indicator for sugar adulteration.
- It was proposed that **natural honey** should have a proline content of **more than 180mg/kg**.
- A **lower proline** content could mean that the honey has been **adulterated with sugar**.
- **However**, this value can be higher for certain honeys as the proline content depends on honey types.

Practical Part

Objective:

- To determine proline concentration in Honey sample.

Principle

- Ninhydrin is used to assay amino acids.

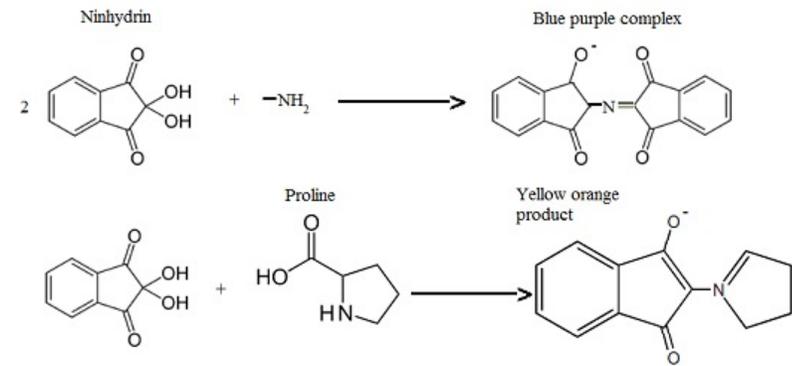
1. At neutral pH:

- It destroys each primary α -amino acid and also reacts with the released NH_3 to form a deep purple chromogen referred to as Ruhemann's Purple, which has a maximum absorption at about 570 nm.
- The reaction with proline and other imino acids yields a **yellow- orange product** at neutral pH, as the cyclized N-group is not released.

💡 Pause and Think **imino acids** ?

2. At low pH (a pH of approximately 1.0) (The principle of experiment):

- Ruhemann's purple is also yielded, but it quickly loses an amine residue, which results into colorless derivatives.
- With proline, a **red water-insoluble** reaction stable product is formed



When handling formic acid, **safety goggles**, **face mask** and **gloves** should be worn.

Method

	B	1	2	3	4	5	S1
Standard	0	0.2	0.4	0.6	0.8	1	--
Sample	--	--	--	--	--	--	1
H ₂ O	1	0.8	0.6	0.4	0.2	0	
Formic acid	0.5 ml						
Ninhydrin	2 ml						

- Mix thoroughly after each addition.
- Boiling water bath for 10 min and then allow to cool at room temperature for 10 min.
- (a deep red color should develop).
- Add 10 ml. of the 2-propanol-water solution (1:1) were added to each tube.
- Mix well using Vortex.
- Measure the absorbance at 520 nm.

Results

Tubes	Abs. At 520 nm	Proline concentration 200mg/dl
1		
2		
3		
4		
5		
Sample		

Calculations

- **The result you got from the curve=** A (mg/dl)

- Conc. of proline= A mg/dl x 1 dl= A (mg)

- A (mg) → 10 grams

- ? → 1000 grams (1Kg)

Note: *(preparation of our sample: 10g of honey in 100ml water)*

- **The proline content=** -----mg/Kg
- For proline the 180 mg/kg minimum value is internationally accepted

Homework

What are the different techniques for adulterants detection in honey? Name “4”