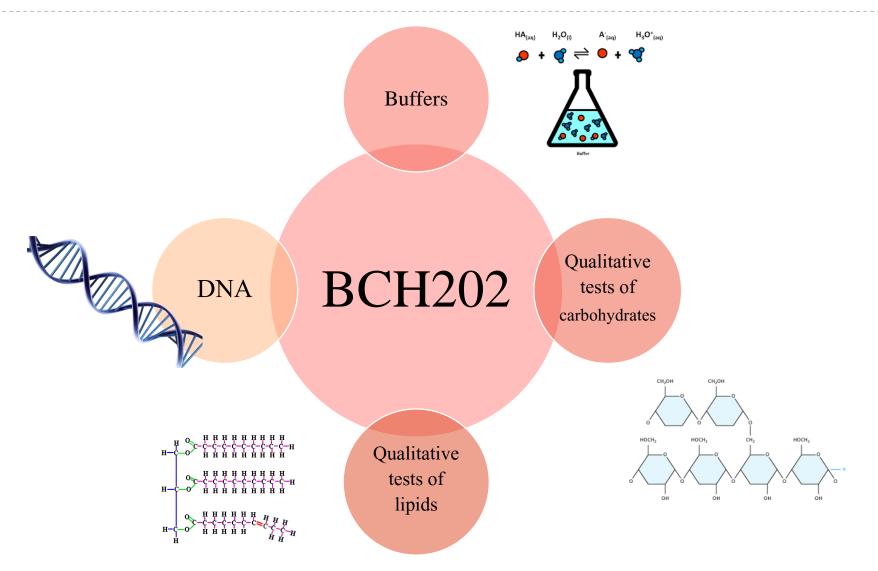
Introduction and lab safety

BCH202 [Practical]

Outline:

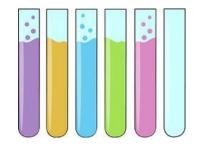




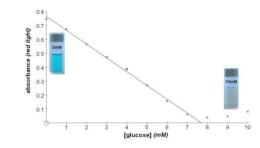
Assa

Qualitative assays

Determine if specific substance is there or not, by color or some other quality.



Quantitative assays Determine the concentration of a substance.



Writing a Report:

- 1. **Cover page:** Title, course code, student name.
- **2. Introduction:** In this part you a background that will help to understand your topic.
- **3. Objectives:** you will write it by your own words.
- 4. Materials and method: As in the lab sheet.
- **5. Results:** You should report all your results that you got from your experiments. Any tables, figures or calculation.
- 6. Discussion: You must write a <u>description and reasons for why you got</u> <u>your results.</u>
- 7. References.



References:

Example:

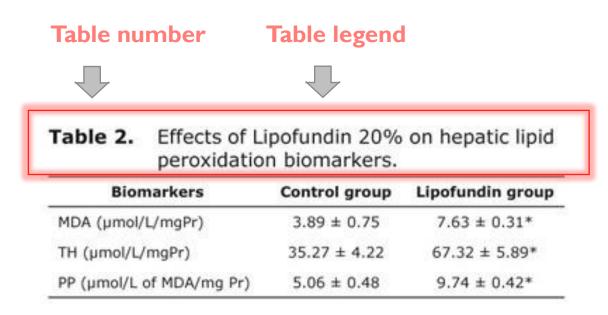
The disease begins as a benign adenomatous polyp, which develops into an advanced adenoma with high-grade dysplasia and then progresses to an invasive cancer ⁽¹⁾.

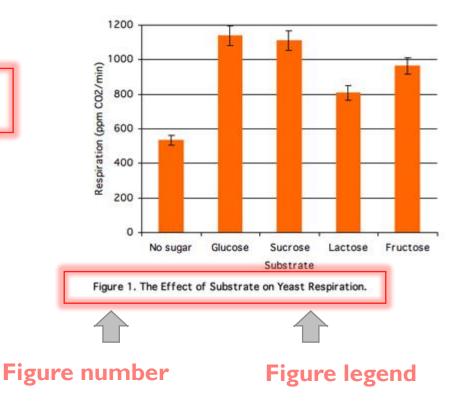
Like other cancers, colorectal cancers arise through a multistep process in which genetic and epigenetic alterations accumulate in a sequential order ⁽²⁾.

References:

- Markowitz SD, Bertagnolli MM. Molecular origins of cancer: Molecular basis of colorectal cancer. N Engl J Med 2009;361: 2449-60.
- Pancione M, Remo A, Colantuoni V. Genetic and epigenetic events generate multiple pathways in colorectal cancer progression. Patholog Res Int 2012; 509348:11.

Legend of tables and figures:



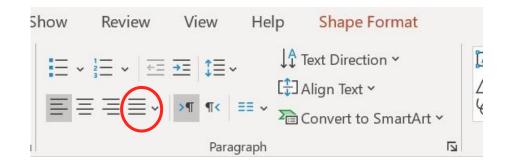


Justify the text:

Example:

It's consist of glass electrode which contain a very thin bulb, blown onto a hard glass tube which is sensitive to pH.The bulb contains a solution of hydrochloric acid and is connected to a platinum lead via silver -silver chloride electrode which is reversible with respect to hydrogen ions.

How?



It's consist of glass electrode which contain a very thin bulb, blown onto a hard glass tube which is sensitive to pH.The bulb contains a solution of hydrochloric acid and is connected to a platinum lead via silver -silver chloride electrode which is reversible with respect to hydrogen ions.

Font and size text:

Font: Times New Roman.

Size:

-Subtitle: 14 pt.

-Body: 12 pt.

General Laboratory Safety

Safe working protects:

You.
Other lab workers and visitors.
Your work.



General consideration:

- > Never eat, drink or chew gum in the lab.
- > Do<u>not</u>taste, smell or touch any chemical.
- > Tell your instructor about any accident .
- > Tie your hair before doing an experiment.
- Closed-toed shoes should be worn at all times.
- > Wash your **hands** with soap after an experiment.



- > You must know all exits in your lab, eye washer, and fire extinguisher, first aid kit.
- > Do <u>not</u> touch any electrical sources .



>Before start working, be sure to **label** the glassware.

≻Glassware should be clean before using.





- > After finishing the experiment turn off all the equipment, clean your work bench.
- > Glassware must be cleaned and kept back at the proper place.



Personal Protective Equipment:







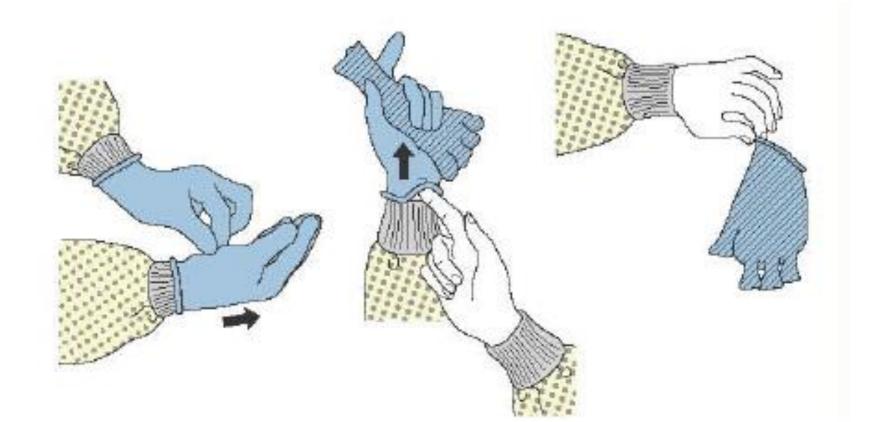


Place your bag in the correct area.

Lab coat should be worn all the time in the lab.

Protective gloves should be worn when handling hazardous materials. Protective glasses should be worn when using hazard chemicals.

How to remove gloves?



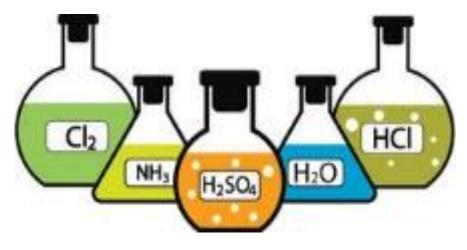
Dealing with chemicals:

- Consider <u>all chemicals to be hazardous.</u>
- > Know what chemicals you are using and notice the hazard symbols.
- Carefully **read the label** twice before taking anything from a bottle.
- > Never point a test tube that you are heating at yourself or your neighbour.
- > You must work at the **hood** when dealing with a chemical with fumes.
- ➢ If chemicals come into contact with your skin or eyes, flush immediately with water and consult with your instructor.



Dealing with chemicals cont?

- Always pour acids into water. If you pour water into acid, the heat of reaction will cause the water to explode into steam.
- > Do not forget to **label your tubes** before starting the lab.
- > Close all chemical bottles after finishing
- > Dispose chemicals properly.
- Don't use the same pipette for different chemicals without washing them.



Hazard symbols

العلامات الإرشادية للمواد الكيميائية





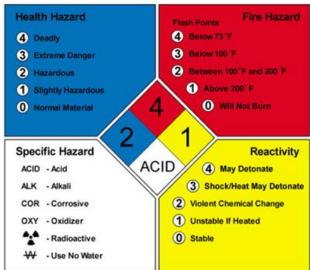
علامات خذيرية للمواد الكيميائية Chemical Warning Signs

Information about chemicals:

Material Safety Data Sheet (MSDS) is a document that contains

information on the potential hazards (health, fire, reactivity and environmental) and **how to work** safely with the chemical product. It also contains information on **the use, storage, handling** and **emergency** procedures all related to the hazards of the material.







General glassware and instrument





Pasteur pipette



Test tubes



Conical flask



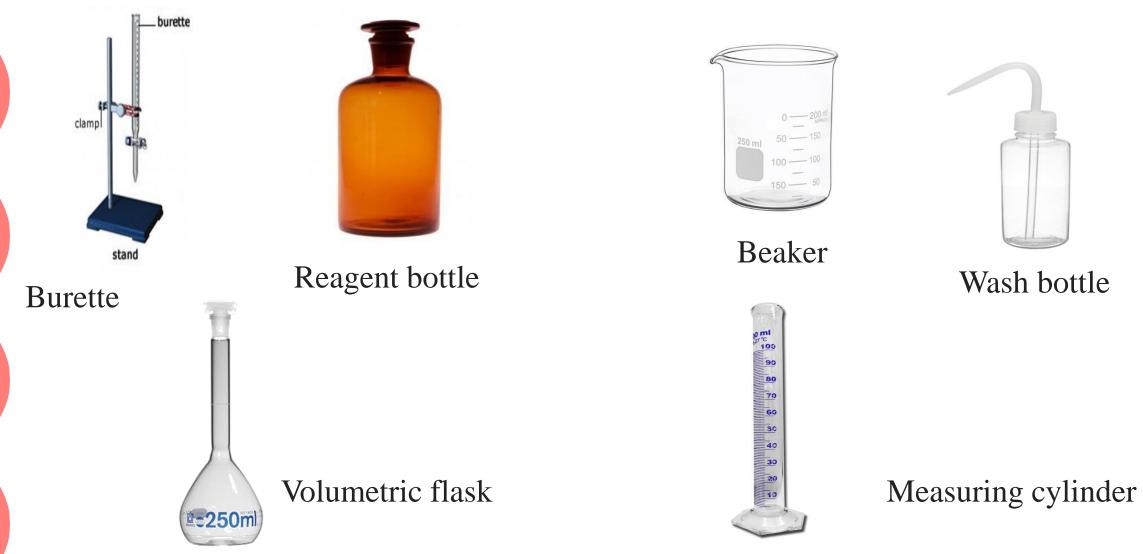
Pipette



Pipette pump

Cuvette





instruments:



Water bath



Spectrophotometer



Electronic balance



pH meter



Search for MSDS of H₂SO₄:

- > What you will do in case of eye contact?
- What you will do in case of skin contact?
- > Is it flammable?
- > Health Management Information Systems (HMIS).