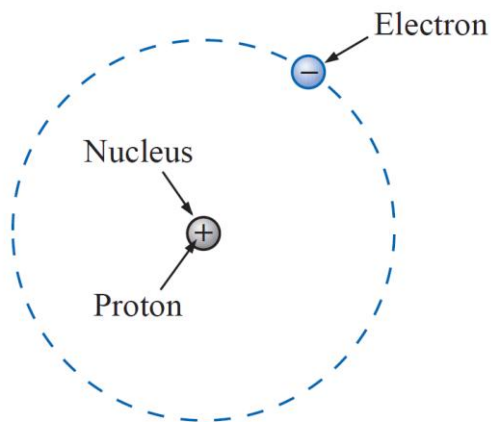


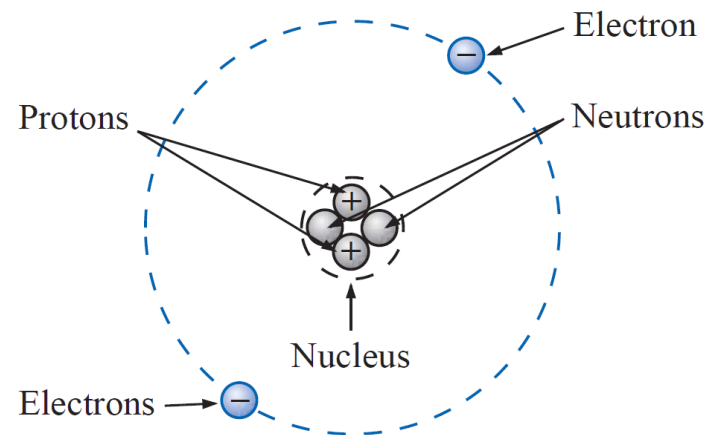
# Current and Voltage

The fundamental concepts of current and voltage requires knowledge of the atom and its structure.

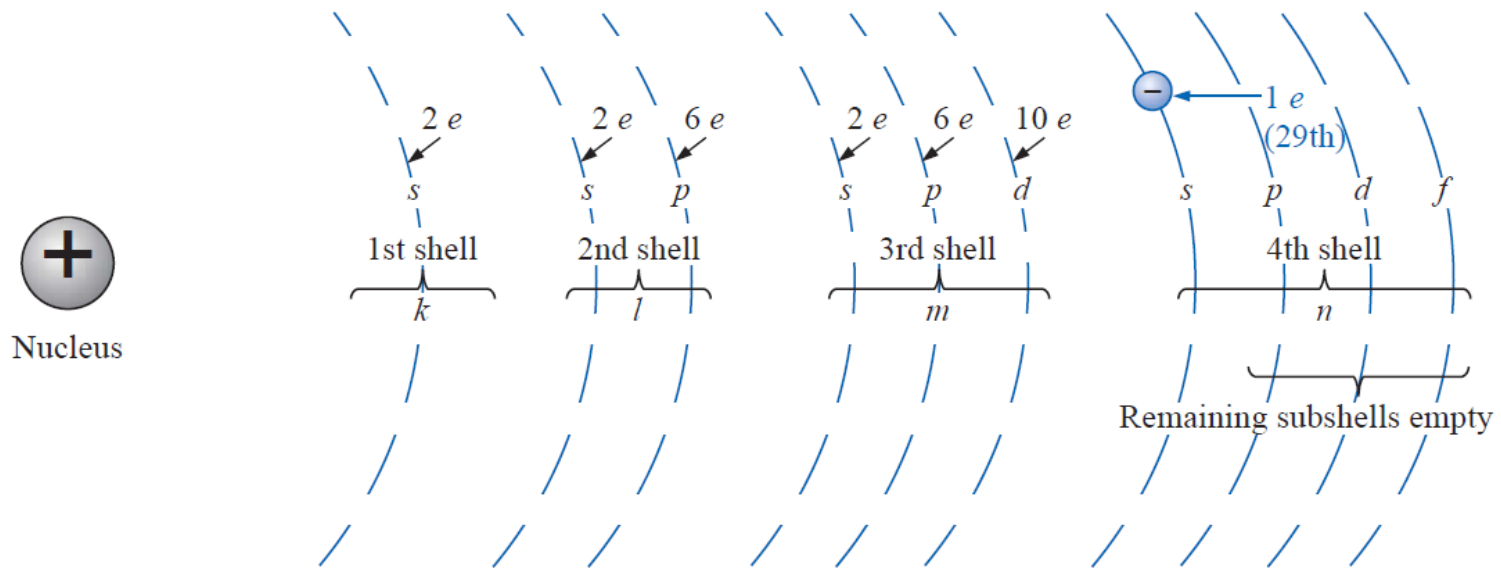
- Nucleus, heavy it contains:
  - ▶ **protons** positively charged
  - ▶ **neutrons** with no charge
- Orbiting negatively charged **electrons**



(a) Hydrogen atom



(b) Helium atom



## *The copper atom.*

One electron in outer most shell it is almost free: with very small energy it becomes **free electron**.

In one  $\text{cm}^3$  of copper there is  $8.54 \times 10^{22}$  such free electrons,

## 2.2 CURRENT

No external forces applied:

At room temperature there exists a random motion of the free electrons created by the thermal energy.

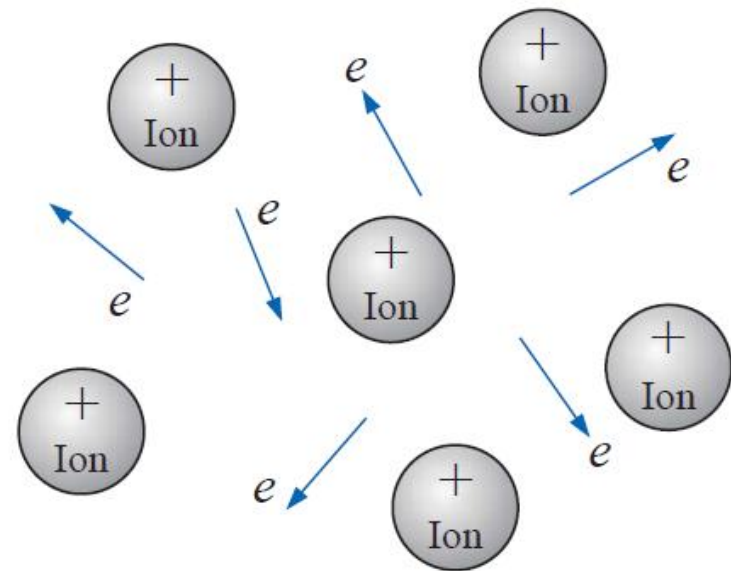
Atoms losing their electron become:  
**positive ions**

The free electrons are able to move within these positive ions.

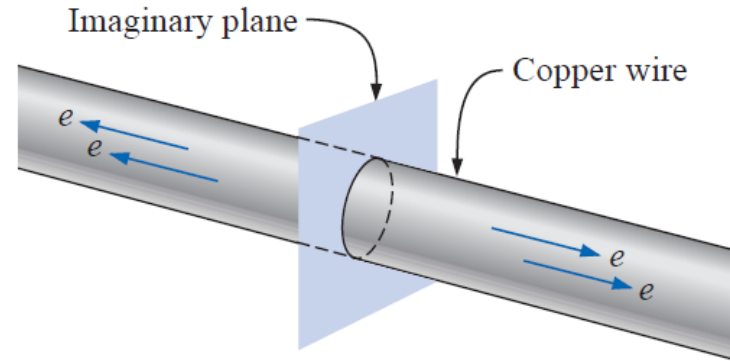
The positive ions can only oscillate around their fixed position.

*The free electron is the **charge carrier** in a copper wire or any other solid conductor of electricity.*

**No External force**



*With no external forces applied, the net flow of charge in a conductor in any one direction is zero.*



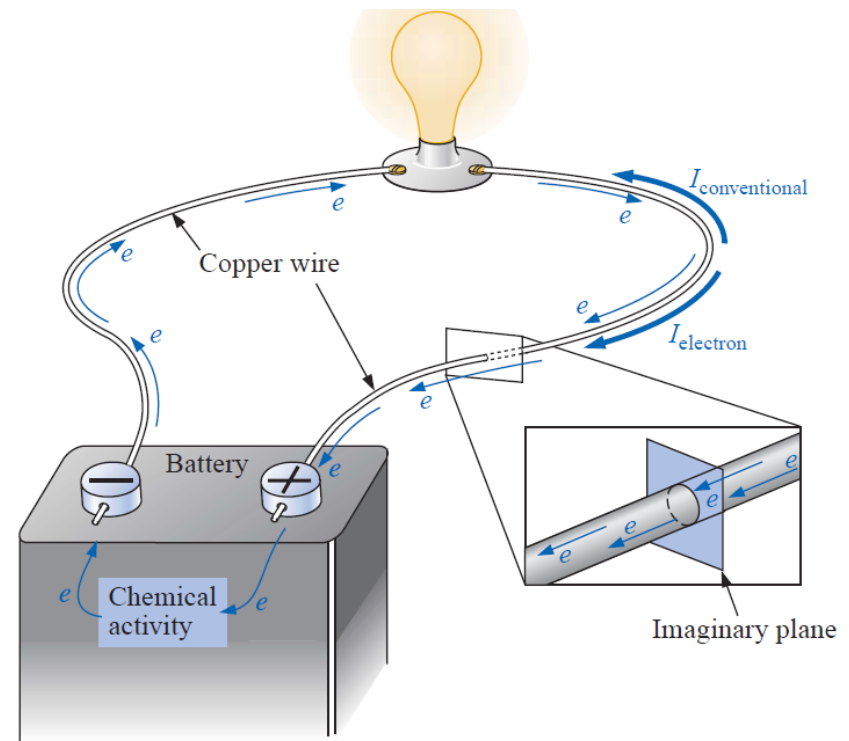
**Battery:** Use chemical energy to put:

- + charge at one terminal
- - charge at the other

**Light bulb:** emit light when electrons pass through its filament.

When the circuit is closed:

- Free  $e^-$  drift toward (+) terminal.
- The (-) terminal is a supply of  $e^-$
- Chemical activity of the battery:
  - Absorb  $e^-$  at (+) terminal
  - Supply  $e^-$  at (-) terminal



*Basic electric circuit.*

When  $6.242 \times 10^{18} \text{ e}^-$  drifts through the imaginary cross section in **1 s** the flow of charge (**current**) is **1 Ampere (A)**.

1 Coulomb (C) = the charge of  $6.242 \times 10^{18} \text{ e}^-$

One  $\text{e}^-$  has a charge of:

$$Q_e = \frac{1 \text{ C}}{6.242 \times 10^{18}} = 1.6 \times 10^{-19} \text{ C}$$

The current is:

$$I = \frac{Q}{t}$$

$I$  = Amperes (A)

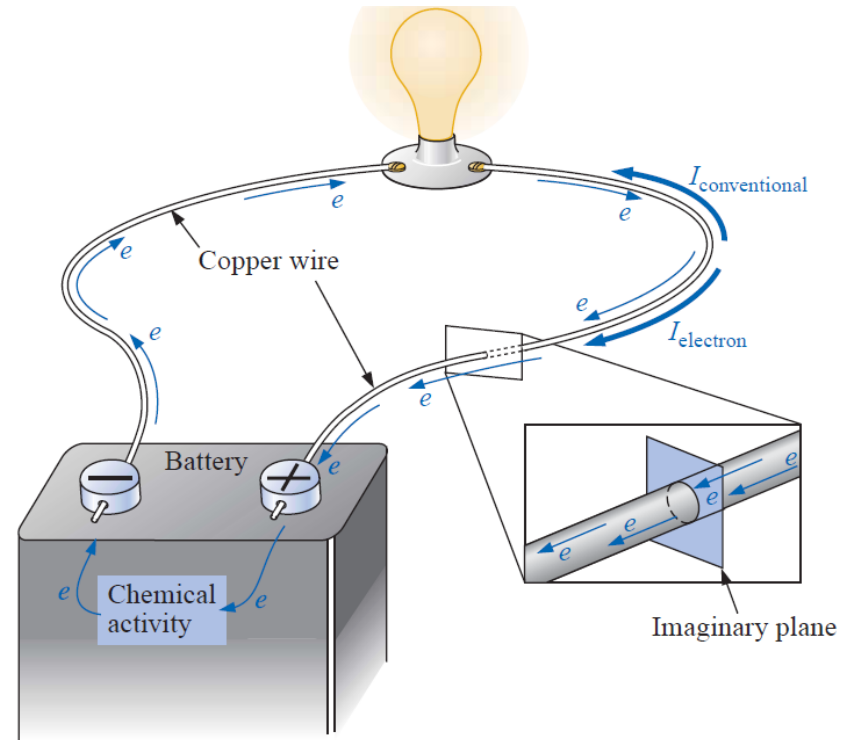
$Q$  = Coulomb (C)

$t$  = second (s)

$$Q = It \quad t = \frac{Q}{I}$$

**Conventional flow:** The direction of  $I$  (direction of positive charges)

**Electrons flow:** Inverse of conventional flow of current



*Basic electric circuit.*

**EXAMPLE 2.1** The charge flowing through the imaginary surface of Fig. 2.7 is 0.16 C every 64 ms. Determine the current in amperes.

**Solution:** Eq. (2.2):

$$I = \frac{Q}{t} = \frac{0.16 \text{ C}}{64 \times 10^{-3} \text{ s}} = \frac{160 \times 10^{-3} \text{ C}}{64 \times 10^{-3} \text{ s}} = \mathbf{2.50 \text{ A}}$$

**EXAMPLE 2.2** Determine the time required for  $4 \times 10^{16}$  electrons to pass through the imaginary surface of Fig. 2.7 if the current is 5 mA.

**Solution:** Determine  $Q$ :

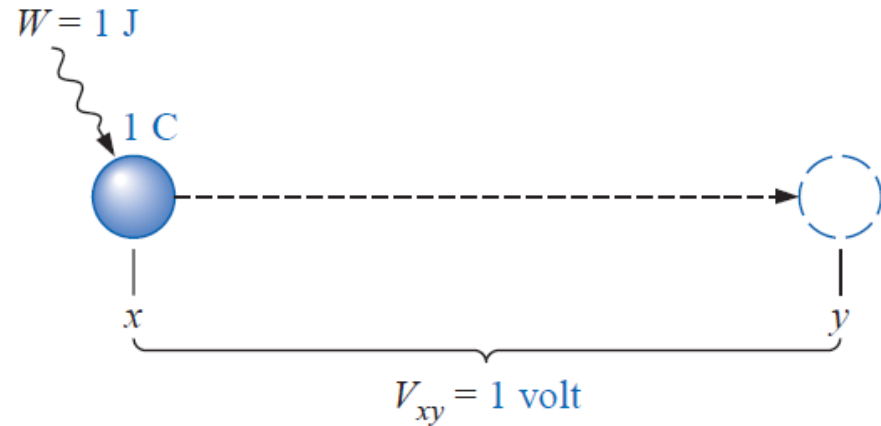
$$\begin{aligned} 4 \times 10^{16} \cancel{\text{electrons}} \left( \frac{1 \text{ C}}{6.242 \times 10^{18} \cancel{\text{electrons}}} \right) &= 0.641 \times 10^{-2} \text{ C} \\ &= 0.00641 \text{ C} = 6.41 \text{ mC} \end{aligned}$$

Calculate  $t$  [Eq. (2.4)]:

$$t = \frac{Q}{I} = \frac{6.41 \times 10^{-3} \text{ C}}{5 \times 10^{-3} \text{ A}} = \mathbf{1.282 \text{ s}}$$

## 2.3 VOLTAGE

*A potential difference of 1 volt (V) exists between two points if 1 joule (J) of energy is exchanged in moving 1 coulomb (C) of charge between the two points.*



The unit of measurement is the **Volt**

Voltage is an indication of how much energy is involved in moving a charge between two points in an electrical system.

*A potential difference or voltage is always measured between **two points** in the system. Changing either point may change the potential difference between the two points under investigation.*

$$V = \frac{W}{Q}$$

(Volts, V)

$$W = QV$$

(Joules, J)

$$Q = \frac{W}{V}$$

(Coulombs, C)

---

**EXAMPLE 2.3** Find the potential difference between two points in an electrical system if 60 J of energy are expended by a charge of 20 C between these two points.

**Solution:** Eq. (2.6):

$$V = \frac{W}{Q} = \frac{60 \text{ J}}{20 \text{ C}} = 3 \text{ V}$$

---

**EXAMPLE 2.4** Determine the energy expended moving a charge of 50  $\mu\text{C}$  through a potential difference of 6 V.

**Solution:** Eq. (2.7):

$$W = QV = (50 \times 10^{-6} \text{ C})(6 \text{ V}) = 300 \times 10^{-6} \text{ J} = 300 \mu\text{J}$$

---

Distinguish between sources of voltage (batteries and the like) and losses in potential across dissipative elements:

***E*** for voltage sources (volts)

***V*** for voltage drops (volts)



**Potential:** *The voltage at a point with respect to another point in the electrical system. Typically the reference point is ground, which is at zero potential.*

**Potential difference:** *The algebraic difference in potential (or voltage) between two points of a network.*

**Voltage:** *When isolated, like potential, the voltage at a point with respect to some reference such as ground (0 V).*

**Voltage difference:** *The algebraic difference in voltage (or potential) between two points of the system. A voltage drop or rise is as the terminology would suggest.*

**Electromotive force (emf):** *The force that establishes the flow of charge (or current) in a system due to the application of a difference in potential. This term is not applied that often in today's literature but is associated primarily with sources of energy.*

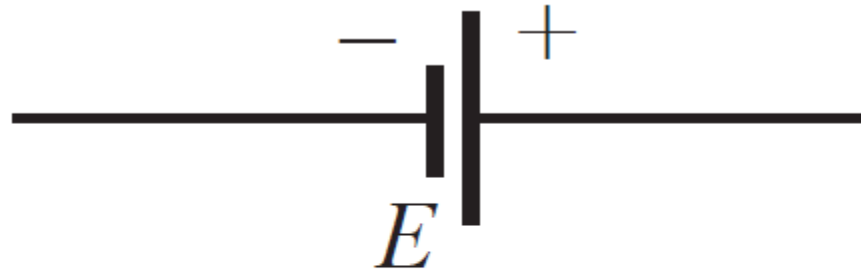
## 2.4 FIXED (dc) SUPPLIES

dc  $\equiv$  direct current  $\Rightarrow$  the flow of charge only in one direction  
 $\Rightarrow$  current only in one direction

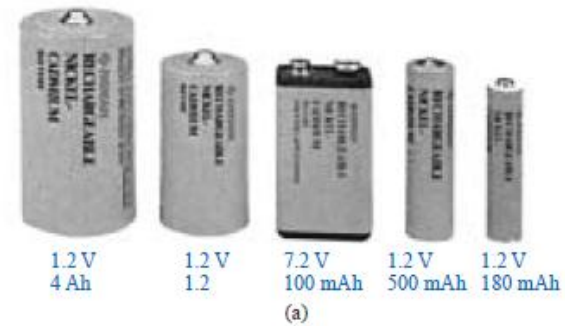
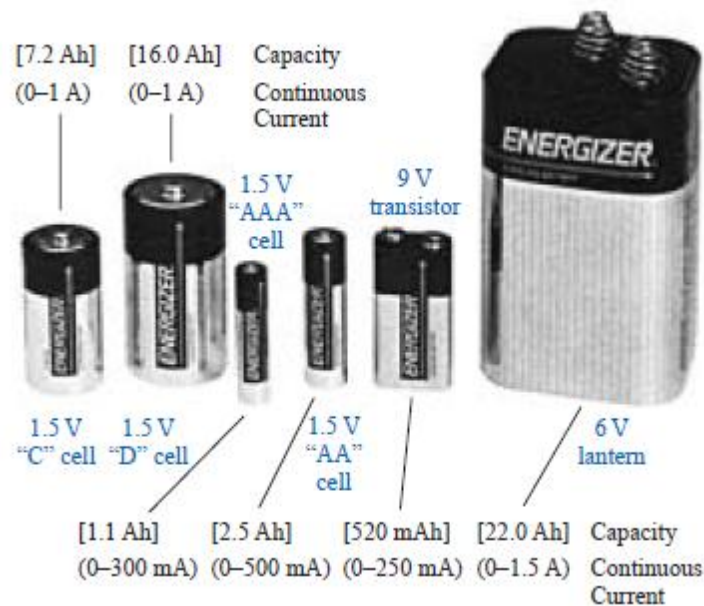
### dc Voltage Sources

Dc voltage sources can be divided into three broad categories:

- (1) batteries (chemical action),
- (2) generators (electromechanical), and
- (3) power supplies (rectification).



*Symbol for a dc voltage source.*



Eveready® BH 500 cell  
1.2 V, 500 mAh  
App: Where vertical height is severe  
limitation

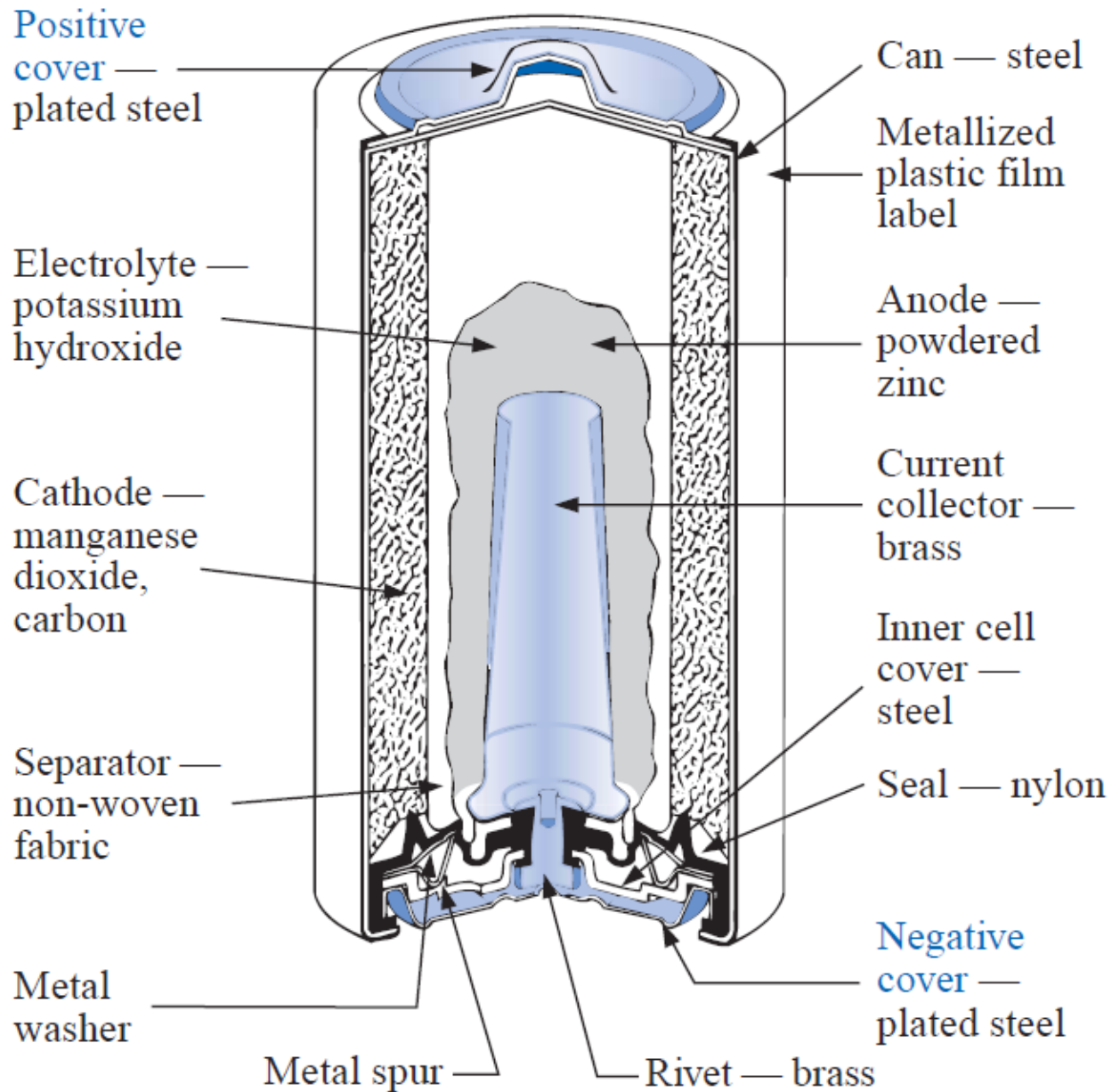
(b)

FIG. 2.15

Rechargeable nickel-cadmium batteries. (Courtesy of Eveready Batteries.)

- (1) Battery is a source of electrical energy developed through the conversion of chemical or solar energy
  - a. Primary type
  - b. Secondary type (rechargeable)

Each battery has: “+” and “-” electrode and an electrolyte:



## Ampere-Hour Rating

Batteries have a **capacity rating** given in ampere-hours (**Ah**) or (**mAh**).

A battery with **100 Ah** rating will theoretically provide steady current of:

1 A for 100 h, or

2 A for 50 h, or

10 A for 10 h,

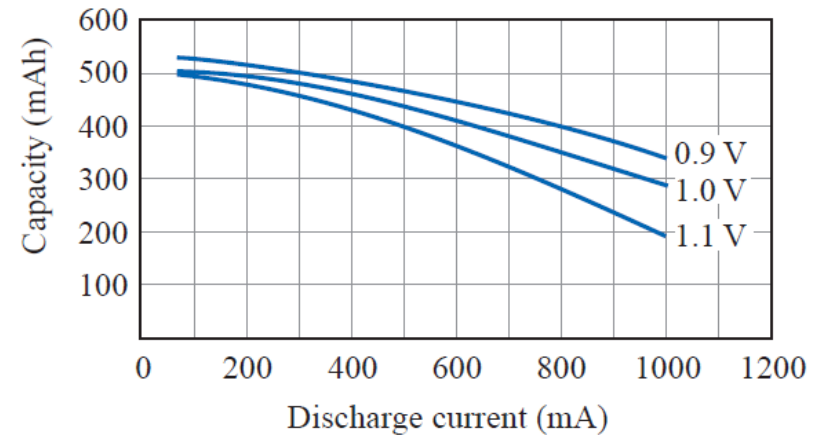
and so on, .... as determined by the following equation:

$$\text{Life (hours)} = \frac{\text{ampere-hour rating (Ah)}}{\text{amperes drawn (A)}}$$

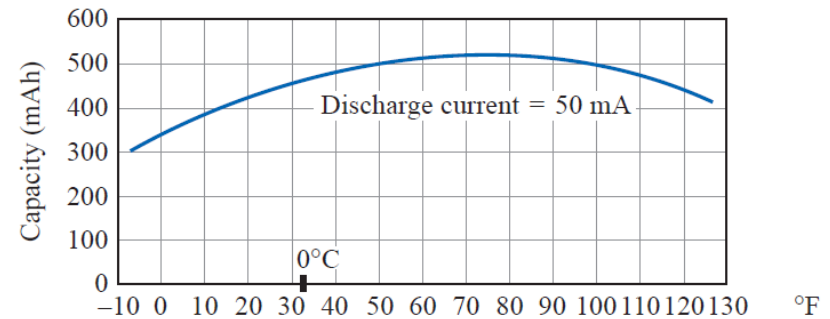
The rating is affected by two factors:

- 1) Temperature
- 2) Rate of discharge (current)

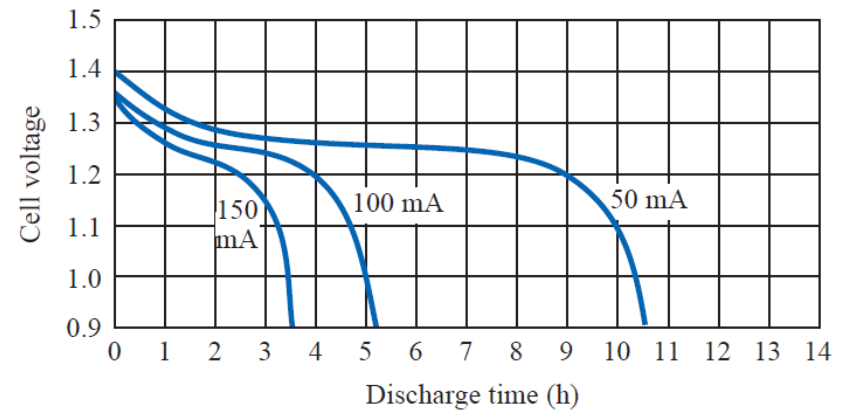
*The capacity of a dc battery decreases with an increase in the current demand*



*The capacity of a dc battery decreases at relatively (compared to room temperature) low and high temperatures.*

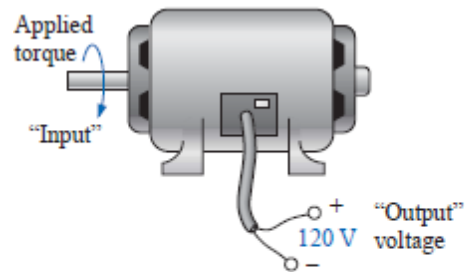


*The terminal voltage of a dc battery decreases with the length of the discharge time at a particular drain current.*



## Power supplies

**Generators**  $\Rightarrow$  120 V or 240 V



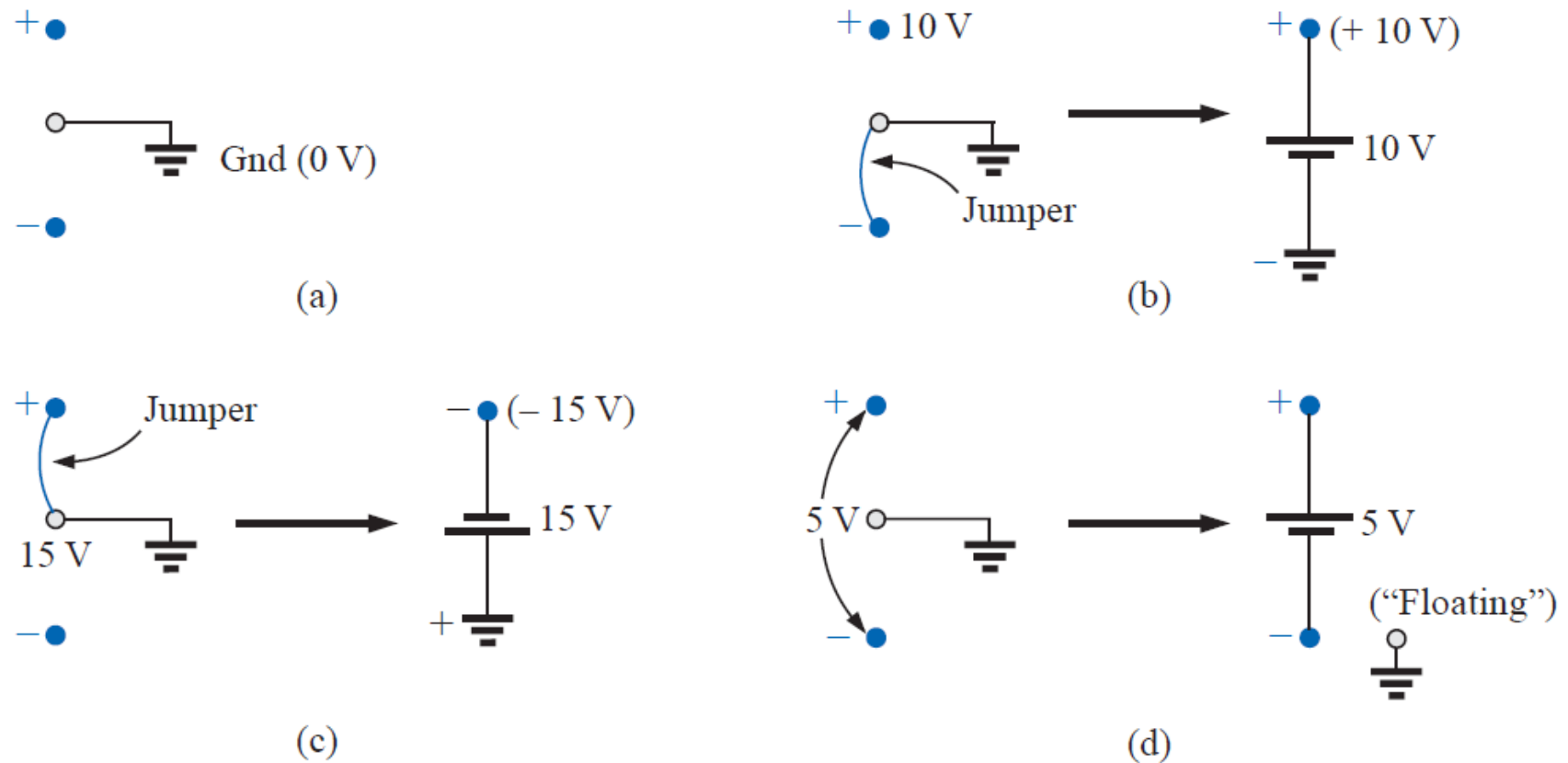
**FIG. 2.20**  
*dc generator.*

From (AC outlet)  $\Rightarrow$  rectification and  
filtering  $\Rightarrow$  dc



**FIG. 2.21**  
*dc laboratory supply. (Courtesy of Leader Instruments Corporation.)*

They must provide three connections: “+” and “-” and “ground”



**FIG. 2.22**

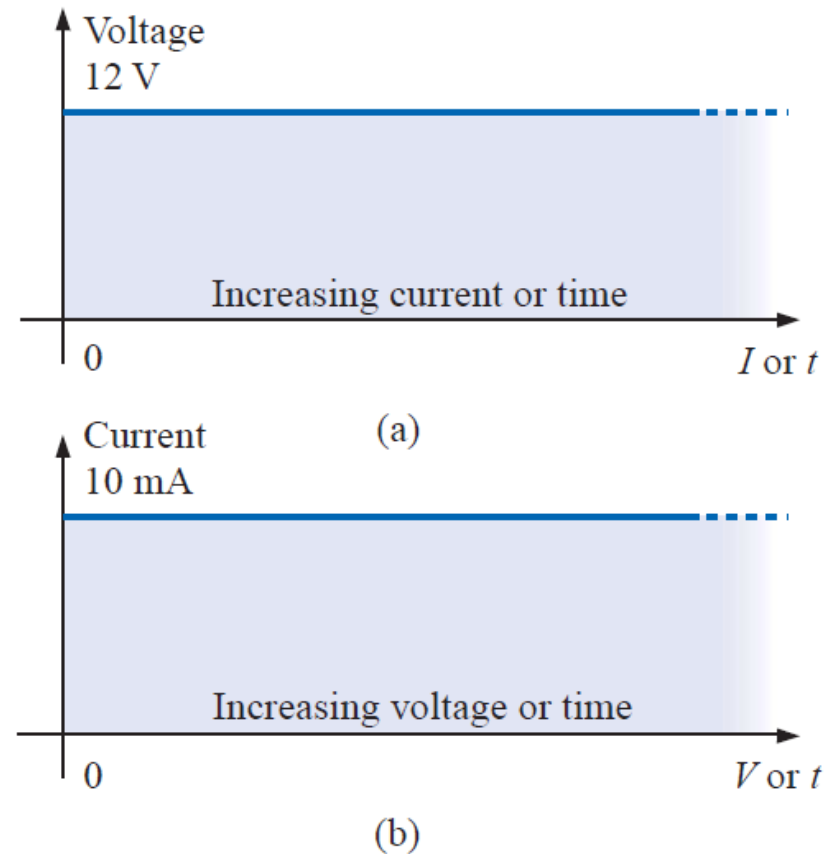
*dc laboratory supply: (a) available terminals; (b) positive voltage with respect to (w.r.t.) ground; (c) negative voltage w.r.t. ground; (d) floating supply.*



## dc Current Sources

*dc voltage*  $\Rightarrow$  *fixed voltage for any current drawn*

*the current source will supply, ideally, a fixed current to an electrical system, even though there may be variations in the terminal voltage as determined by the system,*



**FIG. 2.23**

*Terminal characteristics: (a) ideal voltage source; (b) ideal current source.*

# CONDUCTORS AND INSULATORS

**Conductors** *are those materials that permit a large flow of electrons with very little external force (voltage) applied.*

*Good conductors typically have only one electron in the valence (most distant from the nucleus) ring.*

**Insulators** *are those materials that have very few free electrons and require a large applied potential (voltage) to establish a measurable current level.*

Insulator will break down (permit charge to flow through it) if a sufficiently large potential is applied across it.