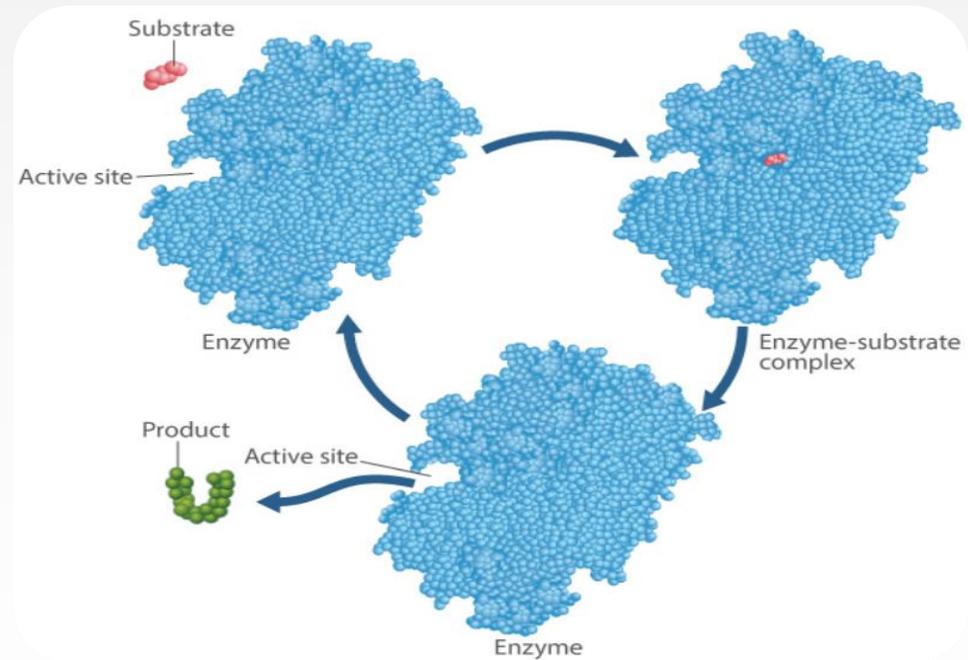
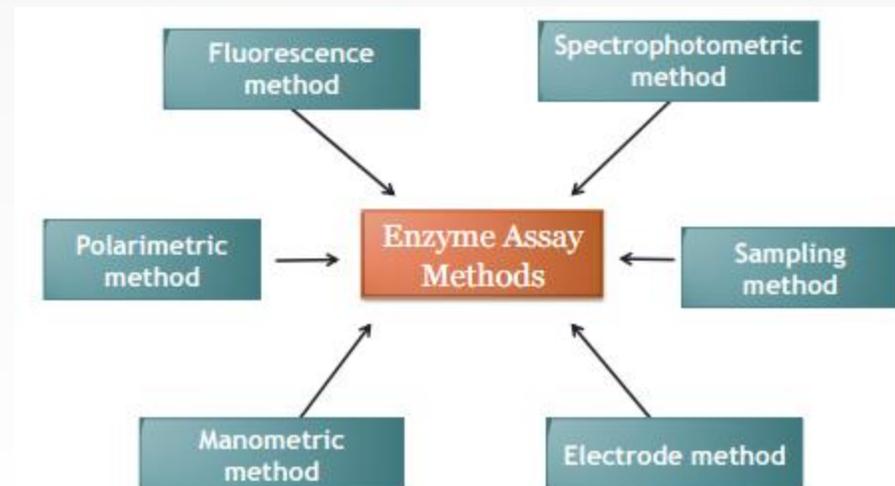


# Method of Enzyme Assay



## Objective:

- To study the different methods for determining enzyme activity.
- Use these method to diagnose certain diseases.



# How to follow a reaction?

*Enzyme assays:* Are laboratory methods for measuring **enzymatic activity**.

- Enzyme assays measure either the **consumption of substrate** or **production of product** over time.

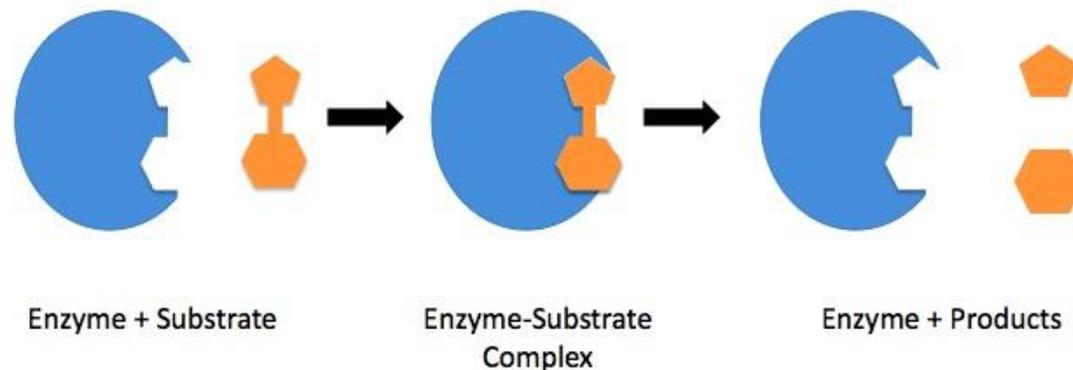


- **Different enzymes require different estimation methods** depending on the type of reaction catalyzed, the nature of S and P or coenzyme.

# Methods of quantitatively following enzyme reactions

## How to follow an enzymatic reaction?

- First you must have a complete knowledge about the reaction itself
- Does the substrate or product has the ability to absorb light, fluorescence, any production of gases, production of  $H^+$ ?
- After that you can use this properties to detect the reaction



# Methods of quantitatively following enzyme reactions

## Fluorescence Methods

Using fluorometer

e.g.  $\text{NAD}^+$  and  $\text{NADP}^+$  do not fluoresce in their oxidized forms, but the **reduced form have a blue fluorescence.**



## Manometric Methods

Using manometer.

It is suitable for reactions in which one of the components is a gas. e.g.

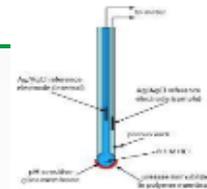
**Oxidases** ( $\text{O}_2$  uptake),  
**Decarboxylase** ( $\text{CO}_2$  output)



## Electrode Methods

Using a pH meter

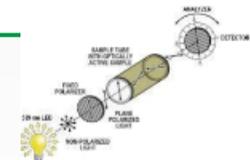
Reactions which involve the **production of acids** where  $\text{H}^+$  conc. is measured



## Polarimetric Method

Using polarimeter

For isomerases that convert one isomer to another.  
e.g.  $\text{D-glucose} \rightarrow \text{L-glucose}$

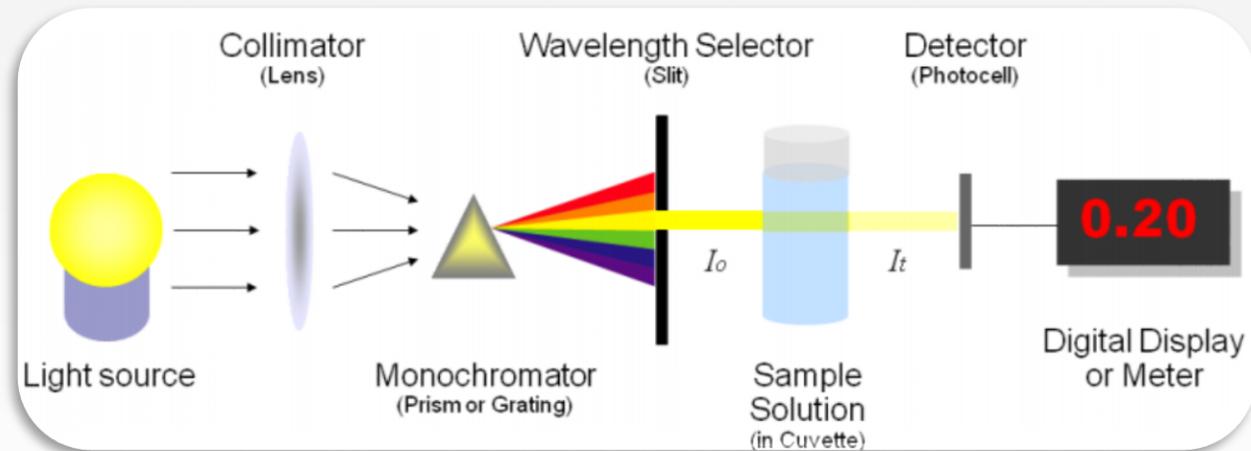


## Spectrophotometric Methods



# Spectrophotometric method

In spectrophotometric assays, you follow the course of the reaction **by measuring a change in how much light the assay solution absorbs**.



**What is blank solution?**

It is a solution that contains everything except the compound to be measured.

# Spectrophotometric methods

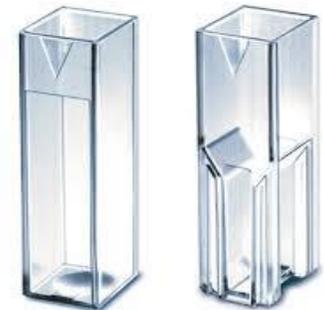
Sample absorb Invisible light

(ultraviolet "UV") 100 to 360 nm

**Quartz cuvette**

Sample absorb Visible range

**Glass or plastic cuvette**



If the light is in the visible region you can actually see a change in the color of the assay, these are called **[colorimetric assays]**

# Examples of using Spectrophotometric methods

1- cases in which product absorb but not the substrate.

e.g.

**Fumarate**

**Fumarate hydratase**

**malate**

Light

2- the Co-enzyme undergoes change in absorption upon reduction or oxidation

Oxidized form

**NAD**

**NADP**

Reduced form

**NADH**

**NADPH**

UV. Light  
340 nm

If **reduced form** was **product**: **increase** the absorbance / min

If reduced form was **substrate** : **decrease** the absorbance / min

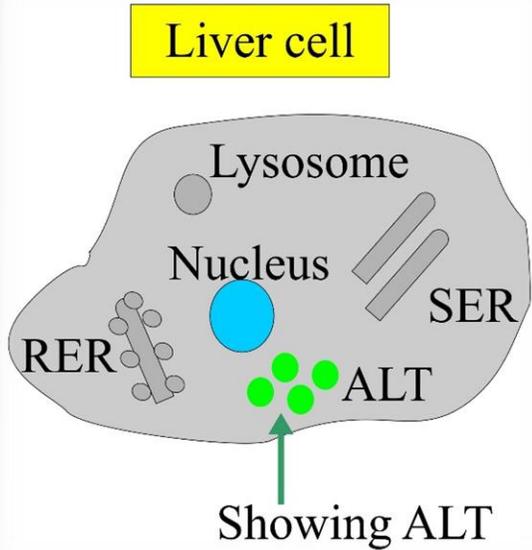
## Two types of Enzyme assays:

- **Continuous assays**, where the assay gives a **continuous reading** of activity.
- **Discontinuous (Endpoint) assays**, Where the reaction is **stopped** and then the concentration of substrates/products determined.

# Alanine transaminase (ALT) in serum

Continues assay

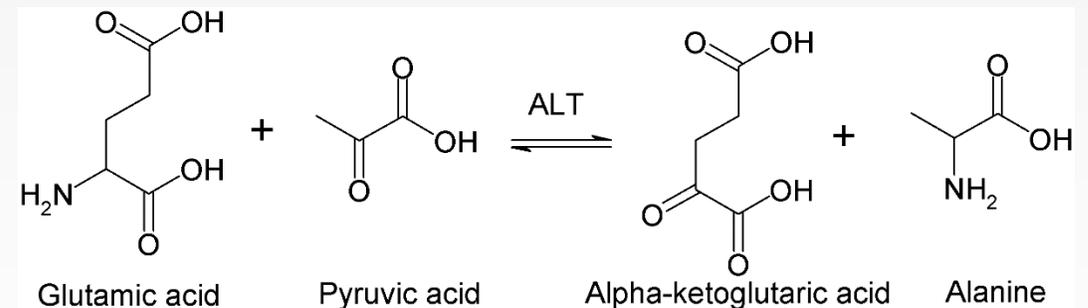
Discontinues assay



# Alanine transaminase (ALT)

- ALT is an enzyme that catalyzes a type of reaction (transamination) between an amino acid and  $\alpha$ -keto acid.
- It is important in the production of various amino acids.

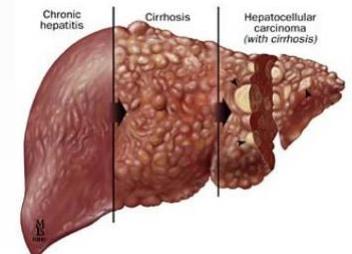
• Transfer of amino group of an amino acid to  $\alpha$ -keto acid resulting in formation of new amino acid and new keto acid.



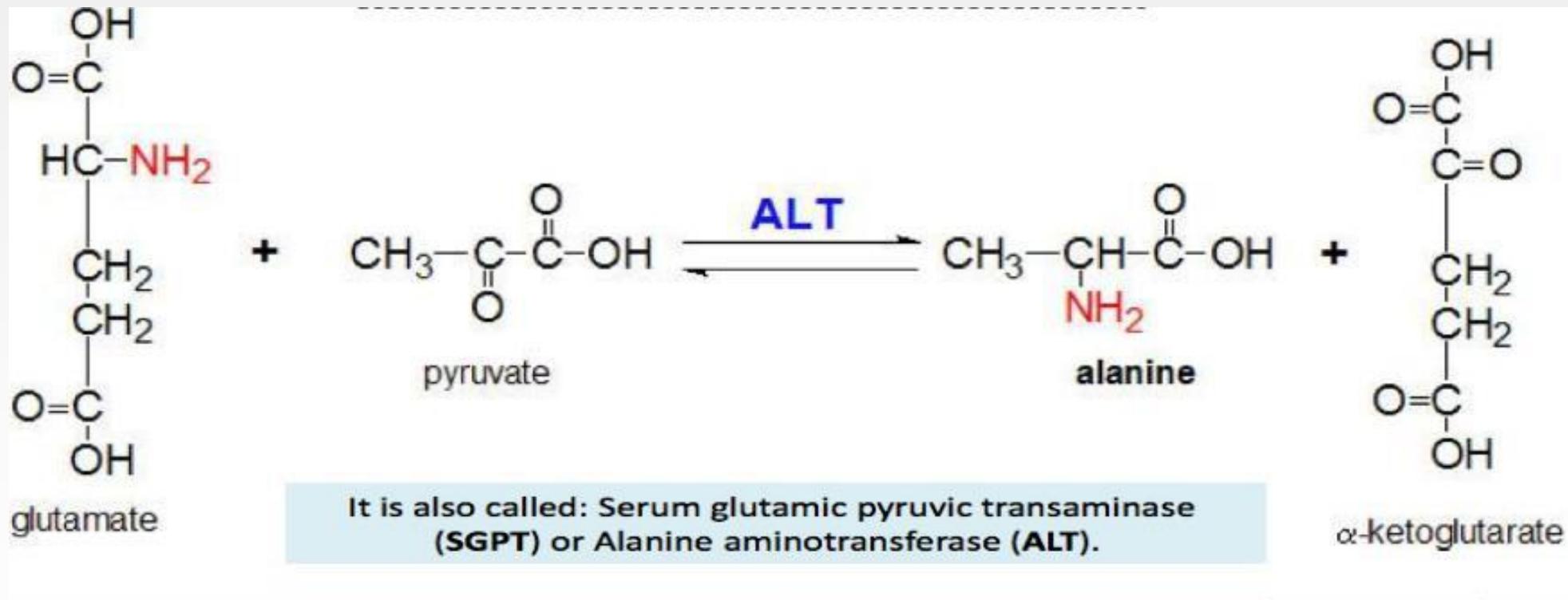
# ALT diagnostic importance



- ALT is found in serum (**at low level**) but is most commonly found in liver.
  - Thus , an **elevated** level of ALT is a sensitive index of ***acute hepatocellular injury***.
  - Elevated serum ALT levels are found in hepatitis, cirrhosis , and obstructive jaundice.
- **NORMAL RANGE OF ALT:** ( up to 42 ) U/L → males  
( up to 32 ) U/L → females



## If Substrate and Product can not absorb light? What is the solution?



# 1-Continuous Assay

- The reading will be continuous (1 min for 3 min).
- Because neither the S nor P can absorb light, the following can be done:

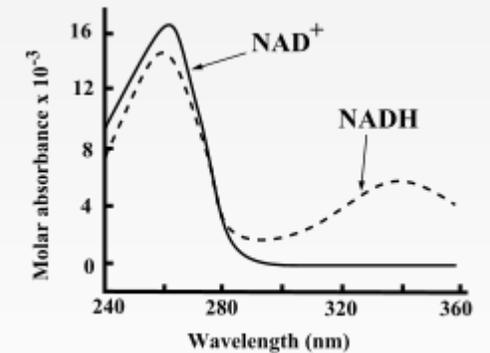
First, we will add the enzyme to the Substrate

1. Alanine +  $\alpha$ -ketoglutarate  $\rightarrow$  Pyruvate + glutamate

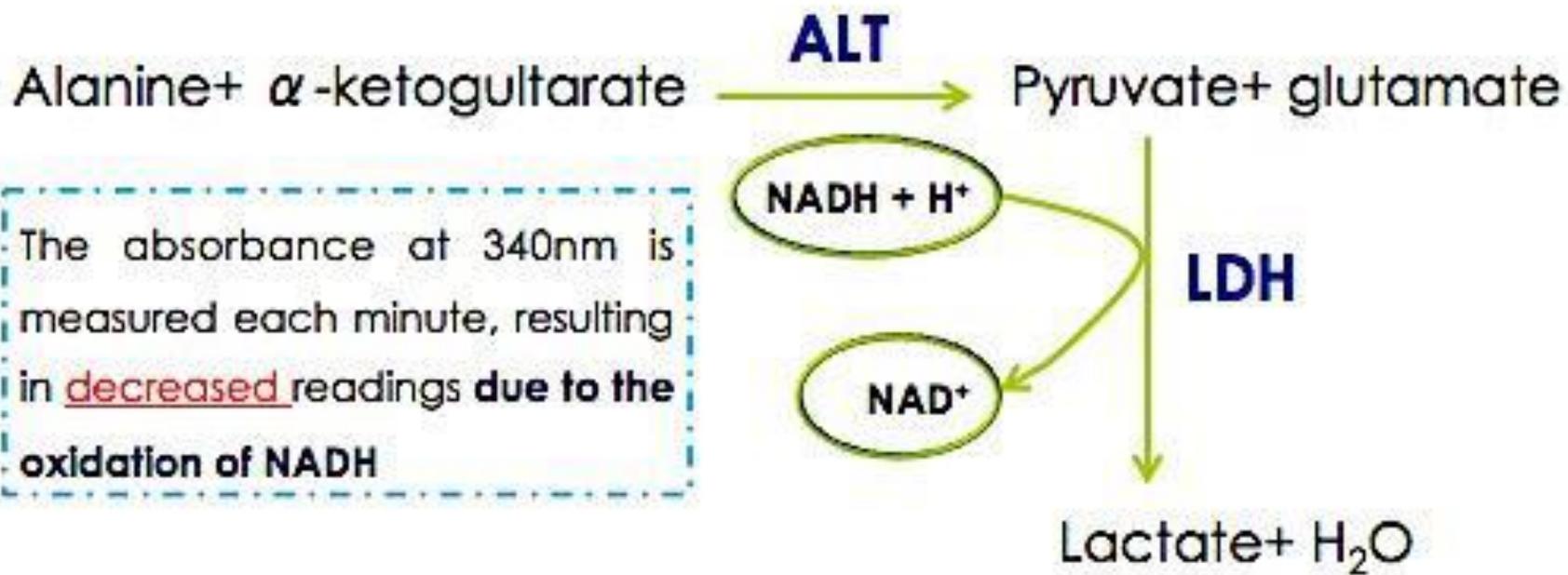
Then, Another enzyme (LDH) and NADH+H<sup>+</sup> will be added:

2. Pyruvate + NADH+H<sup>+</sup>  $\rightarrow$  L-Lactate+ NAD<sup>+</sup> + H<sub>2</sub>O

The absorbance at **340nm** is measured each minute without stopping the reaction, resulting in **decreased** readings **due to the oxidation of NADH.**



# Principle



# Method

Pipette into clean and dry test tubes:

<b>ALT Reagent</b>	3ml
Pre-warm at 37°C for 3 minutes and add	
<b>Serum Sample</b>	0.2 ml = (    μl) ??
Mix and incubated at 37 °C for 1 minute, then read absorbance ( at 340 nm against distilled water ) every minute for 3 minutes) and determine $\Delta A/\text{min}$	

Choose the following on the spectrophotometer:

2) Applications → 2) Simple Kinetics → wave length (340 nm) → Seconds → Duration (180 sec = 3 min) → Intervals (60 sec= 1 min) → Print Data Table (off) → Press start (2 times)

## Results

Time	Absorbance 340nm	$\Delta A/\text{min} = ((A1-A2)+(A2-A3))/2$
1 min	A1	
2 min	A2	
3 min	A3	

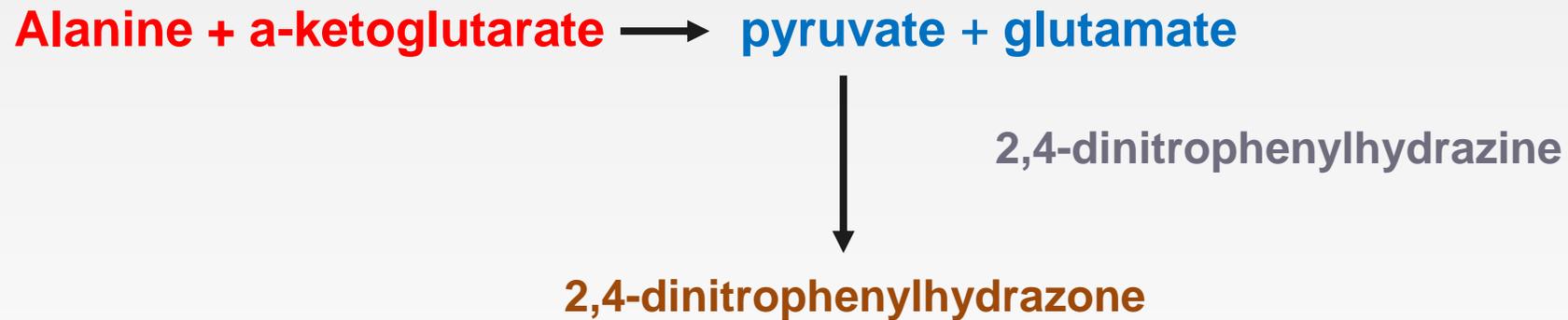
## Calculations:

$$\text{ALT Activity ( U/L)} = \Delta A/\text{min} \times 1768$$

## 2-Discontinuous Assay

- In this method **ALT** catalyzes the following reaction  
**Alanine + a-ketoglutarate**  $\longrightarrow$  **pyruvate + glutamate**
- ALT is assayed by following formation of **pyruvate**.
- Because neither the S nor P can absorb light, the following can be done:
  - The addition of acidic 2,4-dinitrophenylhydrazine (DNPH) leads to the formation of **2,4-dinitrophenylhydrazone (colored product that can absorb light)**,
  - Then, NaOH will be added (to stop the reaction), so that the **colored product** may be measured at **546nm**.

# Reaction-discontinuous



- This assay as an example of **colorimetric\ endpoint assay**

## Method:

	BLANK	SAMPLE
ALT Reagent	0.5 ml	0.5 ml
Pre-warm at 37 °C for <u>5 minutes</u> and add:		
Distilled Water	0.1 ml	-
Serum Sample	-	0.1 ml
Mix, and incubate at 37 °C for exactly <u>30 minutes</u> , and add:		
Color Reagent (DNPH)	0.5 ml	0.5 ml
Mix, and return at 37 °C for exactly <u>10 minutes</u> , then add:		
Color Developer (NaOH)	5.0 ml	5.0 ml
Mix, and return to 37 °C for exactly <u>5 minutes</u> . Read absorbance of all tubes at 546nm against blank.		

## Precautions



- COLOR REAGENT contains 1 N Hydrochloric acid which **causes burns**.
- COLOR DEVELOPER contains 0.5 N Sodium hydroxide which is **corrosive**.

In case of contact, flush affected area with large amounts of water. Seek medical attention.

## Results:

Absorbance at 546 nm	ALT Activity (U/L)
0.025	2.5
0.050	5.5
0.075	9
0.100	12
0.125	17
0.150	21
0.175	25
0.2	30
0.225	35
0.250	41

- The data shown in the table is used to convert absorbance at 546 nm into enzymatic activity in U/L of serum.
- Draw graph using the data in table with absorbance on the Y- axis and enzymatic activity in U/L on the X-axis.

**Note:** Don't forget title of the graph "Standard Curve" and the x- axis and y- axis with their units

**-Absorbance at 546 nm = .....**

**-ALT (SGPT) activity (from graph)= .....**

## Discussion:

- Mention the diagnostic importance of ALT (Introductory paragraph)
- Explain the difference in the principle of each ALT assay.
- Compare your result with ALT normal range [in males], and diagnose the patient's state (what disease could the patient have or not).
- Compare between continuous and discontinuous assays values.