

- Blood, <u>fluid</u> that transports <u>oxygen</u> and <u>nutrients</u> to the <u>cells</u> and carries away <u>carbon</u>
 <u>dioxide</u> and other <u>waste</u> products.
- Plasma is the <u>liquid portion</u> of blood, it constitutes about 55 % of blood volume and 90% of plasma is water.
- **Serum** resembles plasma in composition <u>but lacks the coagulation factors</u>.

(Serum = Plasma – clotting factors)

Collection of Blood Specimens

Blood tests could be performed on serum, plasma, or whole blood depending on the type of the test.



Note: Serum is preferred for many tests (e.g. *determination of lactate dehydrogenase*) as <u>the anticoagulants in plasma</u> <u>can sometimes interfere with the results.</u>

Collection of Blood Specimens



- 1. Can you spot the <u>differences</u> between the two tubes?
- 2. Which tube will clot if it's allowed to set?

Regarding the yellow layer.

- a. Which tube contains clotting factors? Why?
- b. Which tube contains anticoagulant? Why?

Serums	Plasma
Fluid obtained when coagulated blood has been centrifuged	Anti- coagulated blood has been centrifuged
No anticoagulant	Anti-coagulants are needed for separation
Clotting factors are absent	Clotting factors are present in plasma

Types of Anticoagulants

If **whole blood** or **plasma** is desired, an <u>anticoagulant must be added</u> to the specimen <u>immediately</u> after it is drawn or placed into the tube into which the blood is collected.

Tube Cap Color		Anticoagulants	Function
	Green	Heparin	It inhibits the formation of thrombin from prothrombin and thus preventing the formation of fibrin from fibrinogen.
	Purple	EDTA	It is a chelating agent, it binds calcium , which is essential for the clotting mechanism.
	Light Blue	Sodium Citrate	It inhibits blood coagulation by converting calcium into a non- ionized form , and hence prevent clotting of blood.
	Dark Gray	Potassium Oxalate	It inhibits blood coagulation by forming insoluble complexes with calcium ions, which is necessary for coagulation.
	Light Gray	Sodium Fluoride	It has been used chiefly as a preservative since it inhibits red cell metabolism and bacterial action.

Note: Free ionized calcium is required for most events in the blood coagulation process.

Biochemical Changes in Blood Upon Storage

- 1. Loss of carbon dioxide.
- 2. Conversion of glucose to lactic acid (glycolysis).
- 3. Conversion of pyruvate into lactate acid.
- 4. Increase in plasma inorganic phosphate.
- 5. Formation of ammonia from nitrogenous substances.
- 6. Passage of intracellular materials of RBC into plasma.

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Blood Proteins

- Blood contains 8% proteins.
- Plasma Proteins (PPS), plasma contains >300 different proteins.
- > Many pathological conditions affect level of PPS.

The main plasma proteins are:

- Albumin (36-50 g/l), Mw 66 kDa
- Globulins (18-32 g/l), Mw cover a wide range
- Fibrinogen (2-4 g/l). Mw 341 kDa



Site of synthesis

Mwt

All plasma proteins are synthesized in the liver except gamma globulins (immuno-globulins) are synthesized by B cells.

Conc.

Note: All plasma proteins are <u>water soluble</u>, Why?

Major Plasma Proteins

➤ Albumin

It is the most abundant protein present in plasma, and synthesized solely in the liver.

Functions:

- 1. Maintenance of **colloidal osmotic pressure**, thus maintaining blood volume.
- 2. Serves in the transport of bilirubin, hormones, fatty acids, metals, vitamins, and drugs.
- 3. Antioxidation activity, via neutralization of free radicals.

> Globulins

Alpha (α) and beta (β) globulins are transport proteins, but gamma (γ) globulins are part of the immune system.

> Fibrinogen

- 1. Is a soluble glycoprotein with a central role in **blood clotting.**
- 2. It is converted by **thrombin** to fibrin during blood coagulation.

Pause and Think Which of these proteins is absent from serum? Why?



Major Serum Proteins

- Total serum protein consists of two main fractions, **albumin** and **globulin**.
- In normal people the A/G ratio is from 1.2 to 1.5.
- Generally, the <u>decrease in total protein</u> is due to <u>decrease in albumin</u> fraction and <u>increase</u> is due to <u>increase in globulin</u> components.
- **Dehydration** is one condition in which the increase in total protein is due to <u>increase in both albumin and</u> <u>globulin</u> fractions because of **haemoconcentration** \rightarrow In this case the <u>A / G ratio remains unaltered</u>.

Note:	↓ albumin ↓ A/G ratio
	↑ globulin ↑ A/G ratio
In dehydration: ↑	globulin \uparrow albumin = A/G ratio

Major Serum Proteins

A low serum albumin (<u>Hypoalbuminemia</u>) may be due to:

- 1. A heavy loss of albumin in urine.
- 2. Decreased synthesis by the liver due to defective liver.
- 3. Insufficient intake of protein in the diet.
- 4. Malabsorption of protein from the digestive tract.
- 5. Increase catabolism of protein.

A high serum globulin (<u>Hyperglobulinemia</u>) occurs commonly in:

- 1. Advanced liver disease.
- 2. Certain types of cancer, such as multiple myeloma. (**y-globulins**),
- 3. A number of chronic infections.



Objectives

- To **separate plasma** and **serum** from whole blood.
- To separate **blood proteins** using **salting-out** method.
- Identification of blood proteins using **Biuret** and **heat coagulation** method.

Experiment (1): Separation of Plasma and Serum from Whole Blood



Experiment (2): Separation and identification of Proteins in Plasma/Serum

A) Plasma/Serum proteins can be separated by:

- 1. Salting-Out
- 2. Ultracentrifuge
- 3. Electrophoresis
- 4. Chromatography

B) Plasma/Serum protein can be identify by:

- 1. Biuret test
- 2. Heat coagulation

A) Principle of Salting Out

When high concentrations of salt is added to the protein solution, <u>the solubility decreases</u>, and the protein precipitates.

This can be explained by the following:

- The <u>salt molecules compete with the protein</u> molecules in binding with water, leading to **dehydration**.
- The salt concentration needed for the protein to precipitate out of the solution <u>differs from protein to</u>

protein. How?







Biuret test

In the presence of peptides that contain <u>at least two peptide bonds</u> (i.e. it is not given by dipeptides and free amino acids), a copper(II) ion forms violet/blue-colored complexes in an <u>alkaline solution.</u>

Protein + Biuret reagent → violet/blue color

 The intensity of the color is <u>proportional</u> to the number of peptide bonds and thus is a measure of the concentrations of proteins.



Heat coagulation

Protein + weak acid Heating protein precipitate (cloudiness)



Homework

- 1. How albumin contribute to Kwashiorkor oedema?
- 2. How can fibrinogen be used as a drug?
- 3. How decreased temperature can contribute to frostbite?

