PREPARATION OF LABORATORY FOOD SAMPLES

- Aims:

1. Food pre-treatment for analysis.

2. Preventing Changes in Sample.

3. Sample Identification.

4. Sources of Experimental Error.

- Sample preparation:

- Sample preparation is one of the key steps for the development of any new analytical methodology.
- Advances in sample preparation aim to minimize laboratory <u>solvent use</u> and <u>hazardous</u> <u>waste production</u>, save employee time, and reduce the cost per sample, while <u>improving the efficiency</u> of the analytic isolation.
- The food material is usually <u>heterogeneous</u>.
- It is usually necessary to make samples **homogeneous** before they are analysed.

1-Food pretreatment :

- It is often necessary to wash, remove, or drain irrelevant extraneous matter.
- Examples:
 - I. Soil or sand that adheres to fresh fruit or vegetables can be removed by washing or wiping the surface of the produce .
 - II. Removing the skin of a fruit .
 - III. Meat is removed as completely as possible from bone.
 - IV. Eggs are broken to isolate the liquid interior.

1-Food pretreatment:

- A suitable method is then required to make the material less heterogeneous.
- Homogenization can be achieved using:
 - 1. <u>Mechanical devices (e.g., grinders, mixers, slicers, blenders)</u>.
 - 2. <u>Enzymatic methods</u> (e.g., proteases, cellulases, lipases).
 - 3. <u>Chemical methods (e.g., strong acids, strong bases, detergents)</u>.
- The type used depends on the properties of the food being analyzed (e.g., solid, semisolid, liquid).

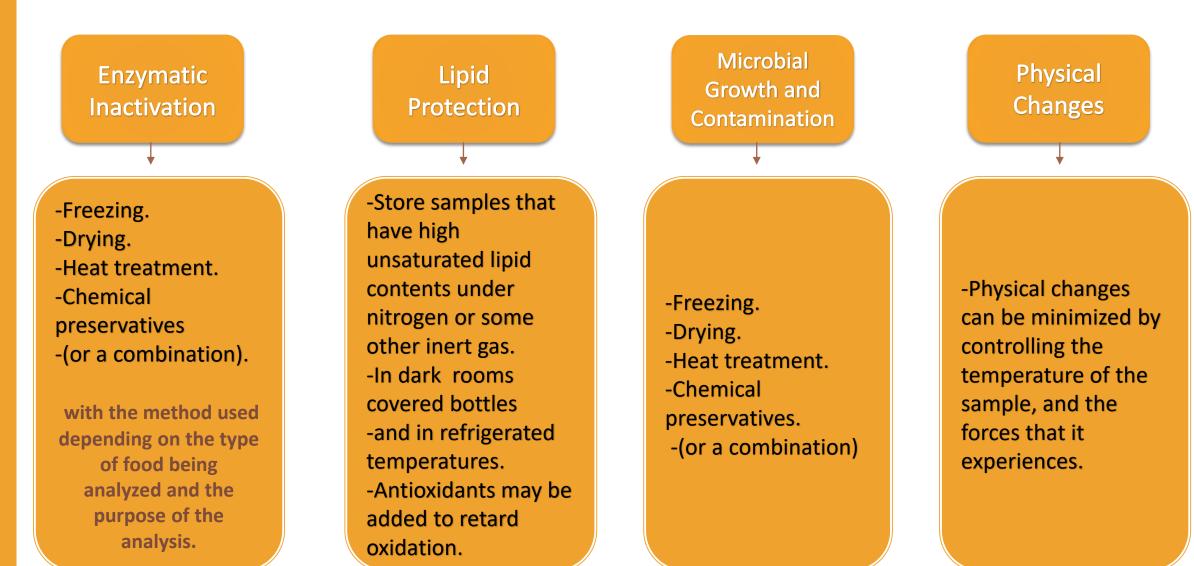
1-Food pretreatment cont':

- The analyses of liquid food samples have an advantage over those associated with solid samples in that they usually require **one less pretreatment step**, due to their liquid form.
- Techniques that <u>may used to prepare liquid samples prior to the analysis step include</u> <u>sample dilution and evaporation.</u>
- Once the sample has been made homogeneous, a small more manageable portion is selected for analysis. → This is usually referred to as a <u>laboratory sample.</u>

2-Preventing Changes in Sample:

- Once we have selected our sample we have to ensure that it does not undergo any significant changes in its properties from the moment of sampling to the time when the actual analysis is carried out.
- e.g., enzymatic, chemical, microbial or physical changes.
- There are a number of ways these changes can be prevented.

Ways that can prevent sample changes



3-Sample Identification:

- Laboratory samples should always be <u>labelled carefully</u> so that if any problem develops its origin can easily be identified.
- The information used to identify a sample includes:
 - a) Sample description.
 - b) Time when sample was taken.
 - c) Location where sample was taken from.
 - d) Person who took the sample.

- The analyst should always keep a detailed notebook clearly documenting the sample selection and preparation procedures performed and recording the results of any analytical procedures carried out on each sample.

4-Sources of Experimental Error:

- A. Personal Errors (Blunders).
- B. Random Errors.
- C. Systematic Errors.

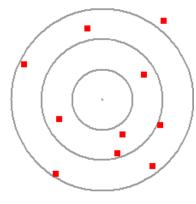
A-Personal Errors (Blunders):

- These occur when the analytical test is **not carried out correctly.**
- The wrong chemical reagent or equipment might have been used.
- Blunders are usually <u>easy to identify and can be eliminated</u> by carrying out the analytical method again more carefully.

B- Random Errors:

• These produce data that vary in a non-reproducible fashion from one

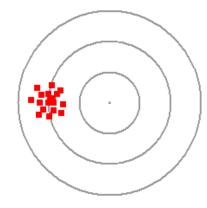
measurement to the next e.g., instrumental noise and environmental conditions.





C- Systematic Errors:

• A systematic error produces results that deviate from the



Systematic Error

true answer in some systematic way.

• This type of error would occur if the volume of a pipette was different from the stipulated value.



- Sampling and Sample Preparation for Field and Laboratory: Fundamentals and New Directions in Sample Preparation, Volume 37Janusz Pawliszyn, 2002.
- 445 BCH Lab note.