## **Laboratory Report**

## **Experiment 9: Carboxylic acids & their derivative**

Name:	•••••

Tube no.	Test	Observation	Conclusion
1	Reaction of Carboxylic Acid and Bicarbonate.  Carboxylic acid +  1ml NaHCO3		
2	Esterification of Carboxylic Acid  1- 1ml of carboxylic acid (acetic acid)+1ml Ethanol +2drops Conc .H2SO4 +Δ  2-Add to test tube contain 10 ml water +Na2CO3		

- Q 1) Write the chemical equation of each one of these reactants with bicarbonate.
- a) Acetic acid
- b) Salicylic acid (2-hydroxybenzoic acid)
- $\mathbf{Q2})$  Write the chemical equation of the reaction for esterification reaction and hydrolysis of ester.