

**First Mid-term Exam**

**Part One (close book). Time allowed: 40 minutes**

Student Name: .....

Student Number: .....

**Question One (20%)**

Answer with true (T) or false (F):

- ( ) (1) Water turbidity is a direct quantitative measurement of suspended solids.
- ( ) (2) The conductivity of distilled water is more than that of tap water.
- ( ) (3) For water with pH of about 8.3-8.5, the predominant alkalinity species present is  $\text{HCO}_3^-$ .
- ( ) (4) Unlike ammonium ion ( $\text{NH}_4^+$ ), ammonia ( $\text{NH}_3$ ) is very toxic to aquatic life.
- ( ) (5) The excessive growth of algae and green plants in water bodies is due to the presence of excessive levels of nutrients, mainly nitrogen and phosphorus.
- ( ) (6) Consumption of drinking water containing high levels of nitrite causes blue-baby disease for infants less than 6 month old.
- ( ) (7) Diseases transmitted by water are almost of intestinal (enteric) origin.
- ( ) (8) For sanitary wastewater, COD is usually less than BOD.
- ( ) (9) Groundwaters are usually free of pathogens and turbidity.
- ( ) (10) The design capacity of water treatment plants is usually based on the maximum hourly demand.
- ( ) (11) Water coagulation is a chemical process involving slow mixing of a chemical coagulant with water.
- ( ) (12) Water of suspended particles of different sizes is easier to coagulate than water of uniformly sized particles.
- ( ) (13) Addition of alum (aluminum sulfate) to water increases water alkalinity.
- ( ) (14) The presence of iron and manganese in water contribute to water hardness.
- ( ) (15) In the excess-lime softening process, excess lime is added for the removal of calcium.
- ( ) (16) True color of water is caused by suspended solids.
- ( ) (17) Highly alkaline water often has a high pH and generally contains high levels of dissolved solids.

( ) (18) Selection of water treatment processes depends only on the type of water source.

( ) (19) About 80% of the domestic water use is for toilet flushing and bathing.

( ) (20) A sand medium with an effective size of 4.0 mm has 10% of its particles (by weight) less than this size.

**Question Two (25%)**

**(A)** Briefly define the following:

1- Biochemical Oxygen Demand (BOD<sub>5</sub>) (5%):

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2- Water Alkalinity (5%)

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3- Water Hardness (5%)

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**(B)** Explain the importance of the jar test in coagulation/flocculation operations (5%)

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**(C)** What are the main advantages of dual-media anthracite-sand filters as compared to single-medium sand filters? (5%)

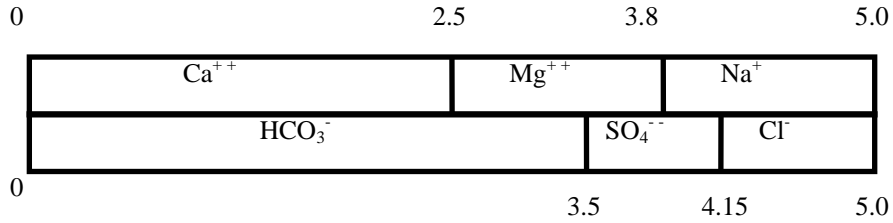
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**Question Three (15%)**

The meq/L bar graph of well water is shown below. Determine:

- (a) Total hardness and alkalinity in mg/L as CaCO<sub>3</sub> .
- (b) The softening chemicals (mg/L) required to remove the carbonate hardness.
- (c) The final hardness of the treated water.



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**Question Four (10%)**

Determine the diameters of two circular settling tanks, not over 3.0 m deep, to settle 6500 m<sup>3</sup>/day of water provided that the overflow rate should not exceed 30 m<sup>3</sup>/m<sup>2</sup>.day and the detention time should be at least 2.0 hours.

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**Question Five (8%)**

Multiple-tube fermentation analyses of well water gave the following results. Determine the total coliform and fecal coliform densities (MPN) and their ranges at the 95% confidence level.

Sample portion (mL)	Number of positive tubes out of five	
	total coliform test	fecal coliform test
10	5	2
1	5	1
0.1	2	0
0.01	1	0
0.001	0	0

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